科目名 Course Title	巡△ル半時回	研究[Laboratomy Evonoice in Chamic-1 C-i	need and Engineering 1]		
科日名 Course Inte 講義題目 Subtitle	1676子符別	研究[Laboratory Exercise in Chemical Scien			
	総合化学院代議員(大学院総合化学院)				
責任教員 Instructor 担当教員 Other Instructors	総合化子阮化 Provided by su				
赵当教員 Other Instructors 科目種別 Course Type	Provided by st	Ipervisor			
	0004	は間刻楽日へ			
開講年度 Year	2024	時間割番号 Course Number	10		
期間 Semester	Full Year	単位数 Number of Credits	10		
授業形態 Type of Class	Experiment	対象年次 Year of Eligible Student	1~2		
対象学科 クラス Eligible Depa					
ナンバリングコード Numbering	Code	CHEM_REQUI 6302			
補足事項 Other Information					
授業実施方式 Class Method		1 対面授業科目《対面のみ》			
キーワード Key Words					
Chemical Sciences and Engine		hesis			
授業の目標 Course Objective		mahlama in chomisters bes there and the	a andust personal In addition		
		s problems in chemistry, solve them, and to r the guidance of instructors in order to			
achievements with excellent ac		-	acquire the ability to complete the		
	adennic research	papers.			
了定日禄 Oodise Goals Complete Master's thesis.					
授業計画 Course Schedule					
	f supervisor(s). I	Please contact to your supervisor for specifi	c research plan.		
準備学習(予習・復習)等の内部					
		o analyze the data, to prepare for presentat	ion, and to write a paper.		
成績評価の基準と方法 Gradir					
Submission of a master thesis i	s required. Evalı	uation is based on the thesis and daily activi	ity in laboratory.		
他学部履修の条件 Other Fac	ulty Requiremen	ts			
テキスト・教科書 Textbooks					
講義指定図書 Reading List					
参照ホームページ Websites					
研究室のホームページ Websit	es of Laborator	У			
備考 Additional Information					
Register this course at the sem	nester of graduat	ion.			

러 다 수 가 가 가 가 하 하 하 하 하 하 하 하 하 하 하 하 하 하 하	1	# ¥[4,1, , , , , , , , , , , , , , , , , , ,	. 7		
科目名 Course Title	物理化学先端詞	講義[Advanced Lecture of Physical Chemis	stry		
講義題目 Subtitle					
責任教員 Instructor	佐田 和己 [SADA Kazuki] (大学院理学研究院)				
担当教員 Other Instructors	ISHIMORI Koichiro (理学研究院), TAKEUCHI Hiroshi (理学研究院)				
科目種別 Course Type					
開講年度 Year	2024	時間割番号 Course Number	094051		
期間 Semester	Summer	単位数 Number of Credits	1		
授業形態 Type of Class	Lecture	対象年次 Year of Eligible Student	~		
対象学科・クラス Eligible Depar					
	バリングコード Numbering Code CHEM_REQEL 5002				
補足事項 Other Information					
授業実施方式 Class Method		2 対面授業科目《一部遠隔》			
キーワード Key Words					
Condensed matter, Macromolec		structure, Magnetic resonance			
授業の目標 Course Objectives					
		atal physical chemistry (physical propertie			
	Secondly, this o	course provides the skill of understanding	ng advanced application of physical		
chemistry in material science.					
到達目標 Course Goals					
		tant matters of physical chemistry and to	apply them to design, synthesis, and		
study of functional properties of	f new materials.				
授業計画 Course Schedule					
Session 1 (1 \sim 3) Instructor: Le					
Basic concepts of nuclear magn					
(reference: ATKINS Physical C	hemistry 10th e	dition; chapter 14, Magnetic resonance)			
Session 2 (4 ~6) Instructor: Pro					
Basic theory and physical prope					
(reference: ATKINS Physical C	hemistry 10th e	dition; chapter 17, Macromolecules and se	lf-assembly)		
Session 3 (7, 8) Instructor: Prof					
		lication of dipole-dipole interactions			
(reference: ATKIN5 Physical C	_nemistry 10th e	dition; chapter 16, Molecular interactions)	!		
	しいまい	•			
準備学習(予習・復習)等の内容					
		y 10th edition; chapter 14 (Magnetic			
	omolecules and s	self-assembly) or equivalent chapters of pr	revious editions. Review according to		
instructors. 武德预压负其维上士社 Quarter	- O				
成績評価の基準と方法 Grading		ttondongo attitudo (25%)			
Final paper for each instructor	-				
Participation more than 70% is r 他学部履修の条件 Other Facu					
他子 部 復修 の 宋件 Other Facu	ity Requirement	S			
テキスト・教科書 Textbooks 会考書、アレキンフ「物理ル学」					
参考書:アトキンス「物理化学」					
講義指定図書 Reading List					
参照ホームページ Websites					
					
研究室のホームページ Website	es of Laboratory				
进去 A J III					
備考 Additional Information					
備考 Additional Information Prerequisite: Students are requested to have					

科目名 Course Title	無機化学生得到	義[Advanced Inorganic Chemistry]			
講義題目 Subtitle	1100110千万场时	₽₽₩LAUVAIICEU IIIOI BAIIIC CHEIIIISUTY]			
責任教員 Instructor	松井 雅樹 [MATSUI Masaki] (大学院理学研究院)				
担当教員 Other Instructors					
科目種別 Course Type					
開講年度 Year	2024	時間割番号 Course Number	094052		
期間 Semester	Summer	単位数 Number of Credits	1		
授業形態 Type of Class	Lecture	対象年次 Year of Eligible Student	~		
対象学科・クラス Eligible Depa					
ナンバリングコード Numbering	Code	CHEM_REQEL 5012			
補足事項 Other Information					
授業実施方式 Class Method		2 対面授業科目《一部遠隔》			
キーワード Key Words					
Powder X-ray diffraction, Bray	ais lattice, space	group, crystal structural factor, Rietveld i	refinement		
授業の目標 Course Objective					
		tructural analyses method in inorganic	materials chemistry. In this class we		
		(RD technique. Advanced measurement			
introduced in the class.			· ····· ······························		
到達目標 Course Goals					
Understand the rietveld refiner	nent technique.				
授業計画 Course Schedule	inenie econinquer				
1. Fundamental of powder X-ra	av diffraction				
2. Measurement and analysis o	-				
3. Measurement and analysis o					
4. Description of crystallograph					
5. Point group and space group					
6. Calculation of peak intensity					
7. Rietveld refinement 1					
8. Rietveld refinement 2					
準備学習(予習・復習)等の内	容と分量 Homewo	rk			
Check the class text in advance					
Summarize your questions just					
成績評価の基準と方法 Gradir					
Attendance 30%, Homework: s		±50%			
他学部履修の条件 Other Fac					
	andy norquinerna				
テキスト・教科書 Textbooks					
Materials will be provided via H	ELMS in advance				
講義指定図書 Reading List	Livio in auvance.				
粉末X先回席の実際 第3版	/中井泉 泉宣-	·夫(編)·朝倉出版 2021			
物質の対称性と群論/今野豊					
X線構造解析/早稲田嘉夫、					
る照ホームページ Websites		- цемрен, 1990			
研究室のホームページ Websit	es of Laboratory				
備考 Additional Information					

	甘水牛州十	版化学性教[Internation Discussion of the second	4]		
科目名 Course Title	▲ 磁 生 物 月 税	&化学特論[Introductory Bio-organic Chemis	stry_		
講義題目 Subtitle					
責任教員 Instructor	永木 愛一郎 [NAGAKI Aiichiro] (大学院理学研究院)				
担当教員 Other Instructors					
科目種別 Course Type					
開講年度 Year	2024	時間割番号 Course Number	094053		
期間 Semester	Summer	単位数 Number of Credits	1		
授業形態 Type of Class	Lecture	対象年次 Year of Eligible Student	~		
対象学科・クラス Eligible Depa					
ナンバリングコード Numbering	; Code	CHEM_REQEL 5022			
補足事項 Other Information					
授業実施方式 Class Method		1 対面授業科目《対面のみ》			
キーワード Key Words					
		l Synthetic Chemistry, Organic Synthetic Ch	emistry		
授業の目標 Course Objective					
Integrated synthetic chemistr	ry, or reaction-	integrated synthetic chemistry, is a synthetic	etic chemistry in which a series of		
reactions are planned and ca	arried out in a	$\operatorname{coordinated}$ manner, rather than in which	each of the reactions required for		
synthesis is planned and carr	ied out separat	ely and independently. In this lecture, the	characteristics of organic synthetic		
reactions using microflow syst	ems and the int	egration of reactions using these characteris	tics will be discussed, and the latest		
examples will be introduced.					
到達目標 Course Goals					
Understand the features relate	ed to microflow s	synthesis and acquire the ability to construct	integrated synthesis based on these		
features.					
授業計画 Course Schedule					
1. Organic synthesis based on	n fast mixing				
2. Organic synthesis based on	n reaction time o	control			
3. Organic synthesis based on	use of short-liv	ved active species			
4. Reaction integration					
準備学習 (予習・復習)等の内	容と分量 Home	work			
It is effective to review the ha	ndouts distribut	ed during the lecture.			
成績評価の基準と方法 Gradi	ng System				
The attendance rate must be o	over 70% to be q	ualified to take the final exam. Evaluations w	ill be made based on report scores.		
他学部履修の条件 Other Fac	ulty Requireme	nts			
テキスト・教科書 Textbooks					
講義指定図書 Reading List					
Lecture 時に指定する。					
Introduced as appropriate in c	lass.				
参照ホームページ Websites					
研究室のホームページ Websi	tes of Laborato	ry			
		•			
研究室のホームページ Websi https://www.chem.sci.hokudai. 備考 Additional Information		•			

科目名 Course Title	生物化学先端講義[Intermediate Biological Chemistry]			
講義題目 Subtitle				
責任教員 Instructor	坂口 和靖[SAK	AGUCHI Kazuyasu] (大学院理学研究院)	
担当教員 Other Instructors	KAMADA Rui (理	皇学研究院), NAKAGAWA Natsumi (理学	研究院)	
科目種別 Course Type				
開講年度 Year	2024	時間割番号 Course Number	094054	
期間 Semester	Spring/Summer	単位数 Number of Credits	2	
授業形態 Type of Class	Lecture	対象年次 Year of Eligible Student	~	
対象学科・クラス Eligible Depa	rtment/Class			
ナンバリングコード Numbering	ナンバリングコード Numbering Code		CHEM_REQEL 5032	
補足事項 Other Information				
受業実施方式 Class Method 1 対面授業科目《対面のみ》				
キーワード Key Words				

Biomolecule, Protein, Protein Structure, Regulation of Protein Function, Folding, Molecular Recognition, Enzyme, Bioinformatics

授業の目標 Course Objectives

The protein function is attributed to its 3D structure and is regulated via control of protein level, activity, and localization by interactions with other biomolecules and posttranslational modification. The class focuses on fundamental aspects of the mechanisms for regulation of protein function based on protein structures. This course also introduces frontier topics of protein function and structures and the course will help the student to expand an understanding of fundamentals of protein structure and function.

In the latter part of the lecture, students participate in virtual research proposals on raising problems and their solutions related to protein structure, function, and control, in Active learning method by the group.

到達目標 Course Goals

After successful completion of this course, you will be able to:

- 1. Understand the regulation mechanism of protein function based on protein structures.
- 2. Obtain basic abilities to search the problems in scientific fields and solve them.

授業計画 Course Schedule

- In the half of the course, the following items are outlined.
- 1. Basic structure and stability of protein
- 2. Molecule recognition of proteins and enzymes
- 3. Control of protein function
- 4. Complex formation and ligand binding
- 5. Protein structure / function prediction

We will also conduct a virtual research proposal by the group on raising problems and their solutions related to protein structure, function and control.

準備学習 (予習・復習)等の内容と分量 Homework

Students are expected to review the material provided by the instructors.

成績評価の基準と方法 Grading System

Problem-based learning on a specific topics of this course (35%). Term examination (40%)

In addition, we also consider it as the important factor for assessment how actively students participate in each class (25%).

他学部履修の条件 Other Faculty Requirements

テキスト・教科書 Textbooks

Materials will be provided in each lecture

講義指定図書 Reading List

タンパク質の構造と機能/グレゴリー A. ペツコ, ダグマール リンゲ著; 宮島郁子訳:メディカル・サイエンス・インターナショナル, 2005

"Protein Structure and Function"/Gregory A. Petsko and Dagmar Ringe:New Science Press, 2004

参照ホームページ Websites

研究室のホームページ Websites of Laboratory

https://www.chem.sci.hokudai.ac.jp/~biochem/en/

科目名 Course Title	実践的計算化	公学[Practical Computational Chemistry]			
講義題目 Subtitle					
責任教員 Instructor		「AKETSUGU Tetsuya] (大学院理学研究院)			
担当教員 Other Instructors	ITOH Hajime (工学研究院), SHIMADA Toshihiro (工学研究院), HASEGAWA Junya (触媒科学研究所)				
科目種別 Course Type					
開講年度 Year	2024	時間割番号 Course Number	094055		
期間 Semester	Fall	単位数 Number of Credits	2		
受業形態 Type of Class	Lecture	対象年次 Year of Eligible Student	~		
対象学科・クラス Eligible Depa	rtment/Class				
ナンバリングコード Numbering	Code	CHEM_REQEL 5200			
補足事項 Other Information					
受業実施方式 Class Method		4 遠隔授業科目《遠隔のみ》			
キーワード Key Words					
-	oretical Chemis	try, Molecular Orbital Theory, Density Func	tional Theory		
授業の目標 Course Objective					
Computational chemistry has k	een a very imp	ortant research technique in chemistry field.	This course is for the students whe		
nave no experience of calculat	ion. Objectives	of this course is to make the students ma	ster how to use calculation on thei		
research issues in accompany v	vith understand	ings on general aspects of computational che	mistry.		
到達目標 Course Goals					
1. Understand the basics of	computational	chemistry, theoretical chemistry, molecular	r orbital theory, density functiona		
heory, excited state calculatio	n.				
2. Use Gaussian and GaussVie	w.				
受業計画 Course Schedule					
1. General Introduction of Con	putational Che	mistry – Prof. T. Taketsugu			
2. Computational Analysis of C	rganic Reaction	ns – Prof. H. Ito			
3. Physical Properties Calculat	ions of Inorgani	c Materials and Organic Semiconductors –	Prof. T. Shimada		
4. Excited State Calculations	- Prof. J. Hase	egawa			
準備学習(予習・復習)等の内容					
Students should have a note PO		7 or later.			
Calculation homework and repo					
The attitude at the lecture (20	%) and report so				
The attitude at the lecture (20	%) and report so				
The attitude at the lecture (20 他学部履修の条件 Other Fac テキスト・教科書 Textbooks	%) and report so ulty Requirement	nts) 亚尼 八卒 (恥攸)・準款灶斗ノー、		
The attitude at the lecture (20 他学部履修の条件 Other Fac テキスト・教科書 Textbooks 新版 すぐできる 量子化学計算	%) and report so ulty Requirement), 平尾 公彦 (監修):講談社サイエン		
The attitude at the lecture (20 他学部履修の条件 Other Fac テキスト・教科書 Textbooks 新版 すぐできる 量子化学計算 ティフィク, 2015	%) and report so ulty Requirement	nts), 平尾 公彦 (監修):講談社サイエン		
The attitude at the lecture (20 他学部履修の条件 Other Fac テキスト・教科書 Textbooks 新版 すぐできる 量子化学計算 ティフィク, 2015 講義指定図書 Reading List	%) and report so ulty Requiremen iビギナーズマニ	nts ニュアル (KS 化学専門書)/武次 徹也 (編集)), 平尾 公彦 (監修):講談社サイエン		
The attitude at the lecture (20 他学部履修の条件 Other Fac テキスト・教科書 Textbooks 新版 すぐできる 量子化学計算 ティフィク, 2015 講義指定図書 Reading List Gaussian プログラムで学ぶ情報	%) and report so ulty Requiremen iビギナーズマニ	nts), 平尾 公彦 (監修):講談社サイエン		
The attitude at the lecture (20 他学部履修の条件 Other Fact テキスト・教科書 Textbooks 新版 すぐできる 量子化学計算 ティフィク, 2015 講義指定図書 Reading List Gaussian プログラムで学ぶ情報 電子構造論による化学の探究	%) and report so ulty Requiremen iビギナーズマニ	nts ニュアル (KS 化学専門書)/武次 徹也 (編集)), 平尾 公彦 (監修):講談社サイエン		
ティフィク, 2015 講義指定図書 Reading List	%) and report sc ulty Requiremen (ビギナーズマニ 最化学・計算化学	nts ニュアル (KS 化学専門書)/武次 徹也 (編集) 学実験/堀 憲次, 山本 豪紀:丸善, 2006), 平尾 公彦 (監修):講談社サイエン		
The attitude at the lecture (20) 他学部履修の条件 Other Face テキスト・教科書 Textbooks 新版 すぐできる 量子化学計算 ティフィク, 2015 講義指定図書 Reading List Gaussian プログラムで学ぶ情報 電子構造論による化学の探究 参照ホームページ Websites	%) and report sc ulty Requiremen (ビギナーズマニ 最化学・計算化学	nts ニュアル (KS 化学専門書)/武次 徹也 (編集) 学実験/堀 憲次, 山本 豪紀:丸善, 2006), 平尾 公彦 (監修):講談社サイエン		

If many applicant, the student will be determined by lottery.

Campus licensed software will be used (no extra cost). No advance preparation is required. Students aiming to real skill acquisition are favorable.

科目名 Course Title	構造古趣ル学	د [Structural Organic Chemistry]		
	博垣 1 機 化 子	-[Structural Organic Chemistry]		
講義題目 Subtitle	<u> </u>	· · · · · · · · · · · · · · · · · · ·		
責任教員 Instructor	鈴木 孝紀 [SUZUKI Takanori] (大学院理学研究院)			
担当教員 Other Instructors				
科目種別 Course Type				
開講年度 Year	2024	時間割番号 Course Number	094056	
期間 Semester	Fall	単位数 Number of Credits	1	
授業形態 Type of Class	Lecture	対象年次 Year of Eligible Student	~	
対象学科・クラス Eligible De				
ナンバリングコード Numberin		CHEM_REQEL 5050		
補足事項 Other Information				
授業実施方式 Class Method	ł	4遠隔授業科目《遠隔のみ》		
キーワード Key Words				
Structural Organic Chemistr	у			
Host-guest complexation				
Supramolecules				
授業の目標 Course Objecti				
·····		by proper designing organic pi-electron syst	ome Students are provided with the	
		to comprehend this area of organic chemist		
complexation.	li are necessary	to comprehend this area of organic chemist	ry, by using the topic of nost guest	
到達目標 Course Goals				
	round and basic i	dea to understand the various intriguing phe	nomona in the functionalized organic	
pi-electron systems/organic	-	dea to understand the various intriguing prie	siomena in the functionalized of game	
授業計画 Course Schedule	301103.			
	-guest complexat	ion and supramolecule formation"		
The major topic is the Trost	, guest complexat			
The close instruction will be	dono in Iononaco			
The class instruction will be 準備学習(予習・復習)等の				
华順子自(ア自・後自)寺の The following text book is us				
The following text book is us 成績評価の基準と方法 Grad				
Presentations and reports	ung System			
他学部履修の条件 Other Fa	outy Poquiromo			
テキスト・教科書 Textbooks				
構造有機化学 基礎から物	性へのアプローチ	すまで/中筋 一弘:東京化学同人,2020		
講義指定図書 Reading List				
構造有機化学 基礎から物	性へのアプローチ	すまで/中筋 一弘:東京化学同人,2020		
参照ホームページ Websites				
研究室のホームページ Web	sites of Laborato	N		
		,		
備考 Additional Information				

취미수 ~ ~~~				
科目名 Course Title	超分子化学[Supramolecular Chemistry]			
講義題目 Subtitle				
責任教員 Instructor	猪熊 泰英 [INOKUMA Yasuhide] (大学院工学研究院)			
担当教員 Other Instructors	ITOH Hajime (工学研究院)			
科目種別 Course Type				
開講年度 Year	2024	時間割番号 Course Number	094058	
期間 Semester	Fall	単位数 Number of Credits	1	
授業形態 Type of Class	Lecture	対象年次 Year of Eligible Student	~	
対象学科・クラス Eligible Depa				
ナンバリングコード Numbering	Code	CHEM_REQEL 5102		
補足事項 Other Information				
授業実施方式 Class Method		1 対面授業科目《対面のみ》		
キーワード Key Words				
	olecular intera	ctions, hydrogen bond, macrocyclic mol	ecules, ion recognition, structure,	
stereochemistry, chirality				
授業の目標 Course Objective				
		asis of supramolecular chemistry including		
	r design and syi	nthesis, higher-order structures, and function	ons as materials.	
到達目標 Course Goals				
Students will be able to explain				
•		interactions (hydrogen bond, CH- π interactions)	eractions, dipole-dipole interactions,	
Coulomb interactions) from the				
	-	cular structures and their principles cyclic compounds, rotaxanes, and catenanes	and their drawheak and advantage	
		ctions from chemical structures of building u	-	
4. expected 5 dimensional str 授業計画 Course Schedule	uctures and fun			
1. what is 'supramolecules', int	ermolecular inte	practions		
2. molecular recognition, ion re				
3. self-assembly, giant suprame				
4. reactions and supramolecula				
5. from current research topics				
6. summary				
準備学習 (予習・復習)等の内容	容と分量 Home	vork		
Students are expected to prep	are the lecture	by reading textbook or handouts which wi	ill be delivered in class, and to read	
reference scientific papers which	ch will be introd	uced in the lecture.		
成績評価の基準と方法 Gradin	g System			
Evaluation will be based on rep				
他学部履修の条件 Other Face	ulty Requiremer	nts		
テキスト・教科書 Textbooks				
大学院 Lecture 有機化学 I	. 分子構造と反	応・有機金属化学/野依良治ほか:東京化	学同人, 1999	
超分子化学/Jean-Marie Lehr				
講義指定図書 Reading List				
参照ホームページ Websites				
研究室のホームページ Websit	es of Laborator	у		
https://www.eng.hokudai.ac.jp	/labo/lor/HP/i	ndex_e.html		
備考 Additional Information				
Students are strongly recomme	nded to check I	ELMS frequently.		

科目名 Course Title	化学工学熱力学!	化学工学熱力学特論[Chemical Engineering Thermodynamics]		
講義題目 Subtitle				
責任教員 Instructor	菊地 隆司[KIKI	UCHI Ryuji] (大学院工学研究院)		
担当教員 Other Instructors				
科目種別 Course Type				
開講年度 Year	2024	時間割番号 Course Number	094059	
期間 Semester	Intensive	単位数 Number of Credits	1	
授業形態 Type of Class	Lecture	対象年次 Year of Eligible Student	~	
対象学科・クラス Eligible Depa	rtment/Class			
ナンバリングコード Numbering	Code	CHEM_REQEL 5111		
補足事項 Other Information	補足事項 Other Information			
授業実施方式 Class Method		1 対面授業科目《対面のみ》		
キーワード Key Words				
Chemical Engineering Thermodynamics, Phase Equilibrium, Chemical Equilibrium, Material-Energy Conversion, Exergy				
授業の目標 Course Objective	S			
Thermodynamics is lectured to	o utilize it in chem	nical engineering. Basic laws of heat phene	omena are reviewed for advanced	
applications. This lecture he	lps you understan	nd that thermodynamics deals with conv	ersion of materials and energy.	
Preservation and loss of energy	gy is lectured by i	introducing a concept of "exergy". You c	an learn the quality of energy is	
expressed in terms of exergy, a	and energy/materia	l conversion systems are to be analyzed to	minimize exergy loss for designing	
clean energy systems. Fuel cel	l systems and hydro	ogen production processes are used as exam	ples for exergy analysis.	
到達日標 Course Goals				

You can extend basic knowledge on thermodynamics in small closed systems to large open systems such as reactors, power plants, and chemical plants. You can understand the concept of exergy, that is, exergy quantifies the available amount of energy based on environmental conditions, and learn the method to calculate exergy for respective energy forms. You can also learn to express exergy losses accompanied with energy conversion by using energy conversion diagram.

授業計画 Course Schedule

First half of this course you will review and expand the concept of chemical thermodynamics to chemical engineering thermodynamics. Second half you will learn the concept of exergy, calculation procedure of exergy, and drawing of energy conversion diagram.

- 1. World trends regarding hydrogen and energy, introduction to hydrogen production
- 2. Basic concept of chemical engineering thermodynamics, chemical thermodynamics, energy balance in closed and flow
- systems, energy balance of chemical processes
- 3. Ideal gas and real gas, compression and expansion
- 4. Chemical equilibrium, equilibrium of heterogeneous reactions
- 5. Introduction to exergy concept, exergy change in energy conversion, energy diagram for energy conversion
- 6. Calculation procedure for exergy of various energy forms
- 7. Exergy for mixing and separation processes, synthesis of process systems
- 8. Exergy analysis of conversion processes in chemical engineering

準備学習 (予習・復習)等の内容と分量 Homework

It is required to study physical chemistry for preparation for the class. Materials are distributed for each class. Homework is assigned every class to well understand the course content. Unit of class is 1, which corresponds to 45 hours study. By considering total time of class, additional study of 3.6 hours is necessary before and after each class.

成績評価の基準と方法 Grading System

Grade will be evaluated based on the grades of small questions and report assignments assigned during the lecture. The evaluation is based on 40% of the small questions and 60% of the report assignments.

他学部履修の条件 Other Faculty Requirements

テキスト・教科書 Textbooks

必要な教材は毎回配布する。参考書は、講義指定図書のとおり。

Handout made by the instructor will be delivered.

講義指定図書 Reading List

熱力学(基本の理解と応用)/石田愈:培風館,1995 演習化学工学熱力学(第2版)/大竹伝雄・平田光穂:丸善,1991 エクセルギー工学/吉田邦夫編:共立出版,1999

参照ホームページ Websites

This course will be provided as part of the Hokkaido Summer Institute., For more information (invited lecturers, coursedetails,etc.),pleasevisitthewebsitebelow:,https://hokkaidosummerinstitute.oia.hokudai.ac.jp/en/courses/CourseDetail=G061

研究室のホームページ Websites of Laboratory

https://apchem.eng.hokudai.ac.jp/en/lab/chemical-system-engineering/

科目名 Course Title	有機反応•構造論[Organic Chemistry of Reaction Mechanism and Molecular Structure]			
講義題目 Subtitle				
責任教員 Instructor	大熊 毅 [OHKU	大熊 毅 [OHKUMA Takeshi] (大学院工学研究院)		
担当教員 Other Instructors	ARAI Noriyoshi (ARAI Noriyoshi (工学研究院)		
科目種別 Course Type				
開講年度 Year	2024	時間割番号 Course Number	094060	
期間 Semester	Spring/Summer	単位数 Number of Credits	2	
授業形態 Type of Class	Lecture	対象年次 Year of Eligible Student	~	
対象学科・クラス Eligible Dep	artment/Class			
ナンバリングコード Numbering Code		CHEM_REQEL 5122		
補足事項 Other Information				
授業実施方式 Class Method	2業実施方式 Class Method 1 対面授業科目《対面のみ》			
t_D_K Kay Wanda				

Molecular Orbital, Chemical Bonding, Reactive Intermediates, Stereochemistry, Molecular Recognition, Pericyclic reactions, The Woodward-Hoffmann rules, Cycloaddition reactions, Electrocyclic reactions, Sigmatropic rearrangements, Group transfer reactions

授業の目標 Course Objectives

1. Pericyclic reactions are the third type of organic mechanism along with ionic and radical reactions. This course explains features of these reactions using a basic molecular orbital theory without the mathematics. The Woodward-Hoffmann rules are introduced to analyze the stereochemical outcome of a series of pericyclic reactions, including cycloaddition reactions, electrocyclic reactions, signatropic rearrangements, and group transfer reactions.

2. In the first half of this course, students learn the behavior of electrons in an atom and/or a molecule from a quantum theoretical point of view, and understand the chemical bonding and the electronic properties of molecules. Based on this achievement, they learn the structure and properties of chemical species, such as carbocations, carbanions, radicals, and carbenes. In order to understand the chemical behavior of molecules, they also learn the stereochemistry that includes the concept of chirality, diastereomeric isomerism, and conformational analysis. Finally, the molecular recognition through intermolecular interaction, mainly hydrogen bonding, is briefly explained. The lecture materials will be uploaded in on-demand form by using "lecture group" at the ELMS. Students will be assigned homework to check understanding.

到達目標 Course Goals

1. Pericyclic reactions include some of the most useful synthetic reactions, such as the Diels-Alder reactions, 1,3-dipolar cycloadditions, and Claisen rearrangements. By learning to recognize the various types of pericyclic reactions and details of their mechanisms through the cyclic transition structures, students will learn to predict whether these reactions are allowed in individual cases.

2. Our goal is understanding of

- the chemical bondings and the electronic properties of molecules based on the behavior of electrons.
- the structure and properties of chemical species , such as carbocations, carbanions, radicals, and carbenes.
- the concept of chirality, diastereomeric isomerism, and conformational analysis.
- the molecular recognition.

授業計画 Course Schedule

1. The nature of pericyclic reactions (1): The basis and four classes of pericyclic reactions are introduced.

2. Cycloaddition reactions (2): A wide range of cycloadditions and their regio- and stereochemical properties are presented.

3. The Woodward-Hoffmann rules and molecular orbitals (2): The Woodward-Hoffmann rules based on the fundamental molecular orbital theory are discussed.

4. Electrocyclic reactions (1): The reaction pathway and the stereoselective outcome are interpreted by using the Woodward-Hoffmann rules.

5. Signatropic rearrangements and group transfer reactions (1): [1,n] and [m,n] rearrangements of suprafacial or antarafacial type are examined. The features of group transfer reactions are explained using two typical examples, diimide reductions and the ene reactions.

6. Electronic structure of atoms (1): The behavior of electrons in an atom is introduced based on the quantum theory.

7. Chemical bonding, molecular orbital, orbital interaction (2): Expression of molecular orbitals by the linear combination of atomic orbital and their interaction are discussed, followed by an explanation of some electronic properties of molecules.

8. Structure and properties of reactive intermediates (2): Chemical structure and properties of typical reactive intermediates, such as carbocations, carbanions, radicals, and carbenes are discussed.

9. Stereoisomerism, chirality, and conformational analysis (1): The way of expression of molecular chirality and stereoisomerism are instructed, followed by introducing the relationship between structure and properties. The method for the conformational analysis is also discussed.

10. Molecular recognition (1): Molecular interaction through hydrogen bonding is briefly discussed.

準備学習 (予習・復習)等の内容と分量 Homework

1. The first half of this course: Students are expected to review the lessons presented in the textbook as well as their own lecture notes. Students are sometimes required to submit assignments.

2. The second half of this course: Students are expected to read relevant contents in the textbook previous to each class (apprx. 15 pages). Students may have short exam or homework, if necessary.

成績評価の基準と方法 Grading System

Grades are awarded based on attitudes through the course and regular assignments (20%) as well as examinations (80%). Students should attend more than 70% of classes.

他学部履修の条件 Other Faculty Requirements

テキスト・教科書 Textbooks

Pericyclic Reactions, Second edition/Ian Fleming:Oxford University Press, 2015 大学院 Lecture 有機化学 I 第2版/野依良治 他:東京化学同人, 2019 March's advanced organic chemistry: reactions, mechanisms, and structure, 7th Ed./Smith, M. B.:John Wiley & Sons, 2013 **講義指定図書 Reading List**

参照ホームページ Websites

研究室のホームページ Websites of Laboratory

https://orgsynth.eng.hokudai.ac.jp/en/ 備考 Additional Information

원모성 가지					
科目名 Course Title	反応上字符論[UI	nemical Reaction Engineering]			
講義題目 Subtitle					
責任教員 Instructor	中坂 佑太 [NAKASAKA Yuta] (大学院工学研究院)				
担当教員 Other Instructors					
科目種別 Course Type					
開講年度 Year	2024	時間割番号 Course Number	094061		
期間 Semester	Spring/Summer	単位数 Number of Credits	2		
授業形態 Type of Class	Lecture	対象年次 Year of Eligible Student	\sim		
対象学科・クラス Eligible Depa	rtment/Class				
ナンバリングコード Numbering	Code	CHEM_REQEL 5132			
補足事項 Other Information					
授業実施方式 Class Method		1 対面授業科目《対面のみ》			
キーワード Key Words					
-		lailan ilailan Diffusion atta Tara			
		leal/non-ideal flow, Diffusion rate, Tra	isport phenomena		
授業の目標 Course Objective			d		
	-	ant to understand ideal and non-ideal	-		
		epts, and methods for the chemical rea			
		ough the interfaces between solid-gas			
		tion of differential equations describing			
		usion and reaction rates on rate-limit	ing step are discussed, based on the		
Thiele modulus and Effectivene	ss factor.				
到達目標 Course Goals					
By the end of this course, you	will				
1. estimate pressure drop and a	residence time in th	e reactor.			
2. analyze non-ideal flow react	or.				
3. estimate diffusion coefficient	in gas and liquid pl	nase.			
4. analyze simultaneous reaction	on and diffusion phe	nomena around the interface between	different phases and within the porous		
materials.					
5. design porous catalysts utiliz	zing effectiveness fa	ctor.			
授業計画 Course Schedule					
1. Reaction kinetics and homo	geneous reactions				
 Reaction kinetics and nome Flow patterns in reactors 	Selleous reactions				
 Continuous reactions in the 	non-ideal flow rea	rtor			
4. Base of mass transport pher					
		around the interfaces between differen	t phasos		
 Simultaneous reaction and o Simultaneous reaction and o 			t phases.		
	-				
7. Thiele modulus and effective 準備学習(予習・復習)等の内					
	ture content about	2 hours per 1 lecture. You are recom	imended to derive equations shown in		
lecture by yourself.					
成績評価の基準と方法 Gradin					
Grading will be based on quizze		s (70%).			
他学部履修の条件 Other Face	ulty Requirements				
テキスト・教科書 Textbooks					
講義指定図書 Reading List					
Chemical Reaction Engineering	∕O. Levenspiel:Jo	ohn Wiley & Sons, 1999			
Elements of Chemical Reaction	Engineering/H. F	`ogler:Pearson, 2020			
反応工学/橋本健治:培風館,	1993				
参照ホームページ Websites					
研究室のホームページ Websit	es of Laboratory				
	-				
備考 Additional Information					

Basic understanding of reaction kinetics and chemical reaction engineering is required. Students should have calculators for each class.

科日夕 Courses Title	右继合武化学[1]	vanaad Organia Synthesia]			
科目名 Course Title	有機合成化子[Ad	vanced Organic Synthesis]			
講義題目 Subtitle					
責任教員 Instructor	石山 竜生 [ISHIYAMA Tatsuo] (大学院工学研究院)				
担当教員 Other Instructors	SENBOKU Hisanc	ml(上字研究院)			
科目種別 Course Type	00004		00.1000		
開講年度 Year	2024	時間割番号 Course Number	094062		
期間 Semester	Fall	単位数 Number of Credits	2		
授業形態 Type of Class	Lecture	対象年次 Year of Eligible Student	~		
対象学科・クラス Eligible Depar					
ナンバリングコード Numbering	Code	CHEM_REQEL 5142			
補足事項 Other Information					
授業実施方式 Class Method		1 対面授業科目《対面のみ》			
キーワード Key Words					
		ction Mechanism, Selectivity, Control of S	tereochemistry		
授業の目標 Course Objectives					
		anic synthesis. In this course, students le	-		
		or realizing these high selectivities. Mo			
		ne papers published in academic journals			
	how to explain the i	reasons why these high selectivities can be	e realized from the basis of learned		
reaction mechanism.					
到達目標 Course Goals	J	f			
		sms for realizing high selectivities in organ			
	concrete selective i	ransformations used in synthesis of natu	rai products and highly functional		
• Boing able to discuss and evol	ain reasons of solo	tivities in several organic transformations.			
授業計画 Course Schedule		tritles in several organic transformations.			
1. Oxidation of Organic Compo	unds				
2. Reduction of Organic Compo					
3. Generation of Enolate and A					
4. Olefination Reaction includin		nd Reaction of Ylides			
5. Stereoelectronic Effects and					
6. Cram Rule and Felkin-Anh M	Aodel				
7. Radical Reaction and Cycliza	ation				
8. Protection of Functional Gro					
9. Attend a seminor or a lectur	-				
10. Drill problems on organic sy	ynthesis				
準備学習 (予習・復習)等の内容	容と分量 Homework				
Before a lecture, students has	ve to learn basic o	organic reactions, such as oxidation, red	luction, aldol reaction and Wittig		
reaction, and their mechanisms	sufficiently.				
After a lecture, students have	to learn again orgar	nic transformations, their selectivities, and	I the reason why their selectivities		
can be realized, which are given					
成績評価の基準と方法 Gradin	g System				
Examination (100%) (Senboku)					
Attendance attitude (20%) and a		na)			
他学部履修の条件 Other Facu	Ity Requirements				
テキスト·教科書 Textbooks					
教科書は使用しない。必要な資料は適宜配布する。 講義 生 - 図ま					
_	講義指定図書 Reading List 大学院 Lecture 有機化学 I 分子構造と反応・有機金属化学/野依良治他:東京化学同人, 1999				
		金属化学/野依良宿他:東京化学问入, 有機化学/野依良治他:東京化学同人,1			
人学院 Lecture 有機化学 II 有 参照ホームページ Websites	的短口现们上于"土物"	日1%11十/ 均仅仅11世·米尔11千円八,」			
研究室のホームページ Websit	es of Laboratory				
備考 Additional Information	11 1 1				
For attending this course, gene	ral knowledge on or	ganic chemistry should be needed.			

科目名 Course Title	無機材料化学特	論[Inorganic Materials Chemistry]	
講義題目 Subtitle			
責任教員 Instructor	忠永 清治[TAD	ANAGA Kiyoharu] (大学院工学研究院)	
担当教員 Other Instructors			
科目種別 Course Type			
開講年度 Year	2024	時間割番号 Course Number	094063
期間 Semester	Spring/Summer	単位数 Number of Credits	2
授業形態 Type of Class	Lecture	対象年次 Year of Eligible Student	~
対象学科・クラス Eligible Depa	rtment/Class		
ナンバリングコード Numbering	Code	CHEM_REQEL 5152	
補足事項 Other Information			
授業実施方式 Class Method		1 対面授業科目《対面のみ》	

Solution process, glass formation, powder preparation, sintering, microstructure and properties, Structural materials, Electric and electronic materials, Optical materials

授業の目標 Course Objectives

This course provides major processes for obtaining various ceramics such as thin films, powders, polycrystals, glasses and so on, which can efficiently yield the excellent property of each ceramic material and lead to practical usage. Additionally, important and close relationship between their physical and chemical properties and microstructure can be also understood. Students also learn the basic properties, production and future issues of ceramic materials, such as structural materials, electric and electronic materials, and optical materials which are particularly important among those produced industrially.

到達目標 Course Goals

1. Understanding of a basic relationship between a variety of functions of ceramics, material forms which can realize those excellent functions, and various processes for fabricating each ceramic with specified material form

2. Understanding of the features of the physical and chemical processes to produce functional ceramics and factors to be controlled in each process

3. Understanding various properties of ceramics such as brittleness, electrical conduction, optics and luminescence.

4. Understanding applications of ceramics to high strength and high toughness materials, semiconductors, polarizers, phosphors, scintillators and solid state laser materials.

授業計画 Course Schedule

1. Introduction of Preparation of Ceramics by sol-gel process

2. Preparation of Ceramics by various solution processes

3. Preparation of thin films by solution processes

4. Preparation of thin films by CVD and PVD

5. Glass formation: process, composition and structure

6. Structural analysis of glasses, crystallization of glasses

7. Ceramic powder synthesis from gas, liquid and solid phases

8. Sintering and microstructure control of ceramics

8. Midterm examination

9. Microstructure and physical properties of ceramics: Characteristics and control of microstructures such as crystal particles, grain boundaries and pores in ceramics

10. Mechanical properties of Ceramics

11. Ceramic dielectrics: classification of dielectrics, properties, and applications

13. Ceramic based ioinc conductores

14. Ceramic-based optical materials

15. Ceramics-based Luminescence materials: phosphors, scintillators, laser materials.

16. Examination

準備学習 (予習・復習)等の内容と分量 Homework

Students are expected to read relevant contents in the text beforehand. After class, students are also requested to understand the lecture by reading additionally the related bibliography and solving problems provided there.

成績評価の基準と方法 Grading System

50%: reports, 50%: examination

他学部履修の条件 Other Faculty Requirements

テキスト・教科書 Textbooks

Sol-Gel Science: The Physics and Chemistry of Sol-Gel Processing / C.J. Brinker and G.W. Scherer: Academic Press, 1990 Synthesis of Inorganic Materials 2nd ed / U. Schubert and N. Husing: Wiley-VCH, 2004 Physical Ceramics -Principles for Ceramic Science and Engineering/Y-M. Chiang, D. Birnie III, and W. D. Kingery: John Wiley & Sons, 1997

講義指定図書 Reading List

参照ホームページ Websites

研究室のホームページ Websites of Laboratory

https://www.eng.hokudai.ac.jp/labo/inorgsyn/

備考 Additional Information

Basic understanding of Physical chemistry, Inorganic chemistry, Solid state chemistry and Inorganic materials chemistry is required.

科目名 Course Title	エネルギー材料特	論[Materials for Energy Conversion and S	Storage]
講義題目 Subtitle			
責任教員 Instructor	幅崎 浩樹[HABA	AZAKI Hiroki] (大学院工学研究院)	
担当教員 Other Instructors	KITANO Sho (エキ	学研究院)	
科目種別 Course Type			
開講年度 Year	2024	時間割番号 Course Number	094064
期間 Semester	Summer	単位数 Number of Credits	1
授業形態 Type of Class	Lecture	対象年次 Year of Eligible Student	~
対象学科・クラス Eligible Depa	rtment/Class		
ナンバリングコード Numbering	Code	CHEM_REQEL 5162	
補足事項 Other Information			
授業実施方式 Class Method		1 対面授業科目《対面のみ》	
キーワード Key Words			
•	rage, ionic conducti	vity, solar energy conversion, electrochem	ical devices
授業の目標 Course Objectives			
		of importance for efficient energy conversi	on and storage for realizing cabon
		about functional materials such as ion	
		on and energy storage, focusing on the rel	
		design of energy conversion and energy s	
到達目標 Course Goals			
- Understand phenomena such	as semiconductor e	lectrode reactions, ionic conduction, and	electrocatalytic reactions from the
viewpoint of material chemistry			
- Understand the principles of	various solar cells,	fuel cells, and rechargeable batteries, an	d the material properties required
to achieve high performance in			
- Get necessary knowledge or	n materials design	for energy conversion and storage through	ugh understanding the correlation
between the structural character			
授業計画 Course Schedule			
1. Materials for fuel cells: Char	acteristics of variou	s fuel cells and materials used in the fuel o	cells will be discussed.
2. Semiconductor electrodes: 1	Based on a band me	odel, fundamentals of photoenergy conver	sion on semiconductor electrodes
will be discussed.			
3. Ion conductors: Structural d	esign and mechanisr	n of ion conduction in inorganic solids will	be introduced and discussed.
4. Electrocatalysts: Structural	and electronic desig	n of electrocatalysts for hydrogen evolutio	on and oxygen evolution/reduction
will be introduced and discusse	d.		
5. Presentations: Characteristi	cs of several electro	ochemical energy storage and conversion	devices and their materials will be
presented by individual student	s and discussed.		
準備学習 (予習・復習)等の内容	容と分量 Homework		
Students are requested to prep	are presentations of	specific topics allocated to each student.	
成績評価の基準と方法 Gradin	g System		
Presentations (50%) and exam (50%)		
他学部履修の条件 Other Faci	ulty Requirements		
テキスト・教科書 Textbooks			
教科書は使用しない。必要に応	いし, プリントを配布す	トる。	
講義指定図書 Reading List			
参照ホームページ Websites			
研究室のホームページ Websit	es of Laboratory		
備考 Additional Information			
Students need basic knowledge	on inorganic chemis	stry and electrochemistry.	

科目名 Course Title 溝義題目 Subtitle		診[Advanced Applied Di1i-+]		
再載版日 SUDUTIE	110月生化子符	論[Advanced Applied Biochemistry]		
責任教員 Instructor	松本 謙一郎 [MATSUMOTO Kenichiro] (大学院工学研究院)			
担当教員 Other Instructors	HACHISUKA	Shin-ichi (工学研究院), FUJITA Masahiro (KIKEN)	
科目種別 Course Type				
開講年度 Year	2024	時間割番号 Course Number	094065	
期間 Semester	Intensive	単位数 Number of Credits	1	
受業形態 Type of Class	Lecture	対象年次 Year of Eligible Student	~	
対象学科・クラス Eligible Depa				
ナンバリングコード Numbering	Code	CHEM_REQEL 5171		
補足事項 Other Information				
授業実施方式 Class Method		1 対面授業科目《対面のみ》		
キーワード Key Words				
Genetic information, protein s	structure, molec	ular mechanism, biosynthetic mechanism, a	animal cells, secondary metabolites	
piopolymers, bioremediation, p	hysical chemistry	У		
受業の目標 Course Objective	S			
=		novel engineering subjects on of biomole	ecules in the fields of life science	
nformation, medicine, and env		never engineering subjects on or bioliloic	sector in the helds of me selence	
到達目標 Course Goals				
	derstand deeply	the topics of genetic information, protein	structure animal cell cultivation	
•		ean environments in the fields of life so		
environment.	nymers, and ch	ean environments in the news of me so	cience, information, medicine, an	
受業計画 Course Schedule				
	nalytical methods	s of RNA and other biomolecules		
5–8: Strategies of metabolic pa				
7 0. Strategies of metabolic pa	aniways, and prin	leiples of enzymatic reactions		
準備学習 (予習・復習)等の内	중노스를 비amau	work		
		xt time. Students submit a report after the l	ooturo	
成績評価の基準と方法 Gradir		xt time. Students submit a report after the r	ecture.	
Active class participation and i	-			
The attendance rate must be o				
他学部履修の条件 Other Fac	ulty Requiremen	ts		
テキスト・教科書 Textbooks				
講義指定図書 Reading List				
参照ホームページ Websites				
This course will be provided	as part of the	Hokkaido Summer Institute., For more inf	formation (invited lecturers, cours	
letails, etc.),	please	e visit the	website below:	
//1 11 • 1	ute.oia.hokudai.a	c.jp/en/courses/CourseDetail=G046		
https://hokkaidosummerinstitu				
nttps://hokkaidosummerinstitu 研究室のホームページ Websit	tes of Laborator	y		
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講義相目 Subtite				
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5. Functional polymer materials via advanced synthetic strategy 6. Report preparation 準備学習(予習・復習)等の内容と分量 Homework Carefully reading handouts distributed in advance, if available. 成績評価の基準と方法 Grading System Attendance of 70% or more of the number of class hours shall be the condition of the grade evaluation. The grade is evaluated by (1) attitude in the class (20%) and (2) the report assignments (80%). To pass, students must earn at least 60 points out of 100 points. 他学部履修の条件 Other Faculty Requirements テキスト・教科書 Textbooks 特に指定はない。授業時に資料を配付する。 Reference materials will be distributed as necessary. 講義指定図書 Reading List 参照ホームページ Websites 研究室のホームページ Websites of Laboratory	• •			
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Reference materials will be distributed as necessary. 講義指定図書 Reading List 参照ホームページ Websites 研究室のホームページ Websites of Laboratory		斗を配付する。		
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研究室のホームページ Websites of Laboratory	講義指定図書 Reading List			
研究室のホームページ Websites of Laboratory				
	参照ホームページ Websites			
		<u> </u>		
		-		
https://poly-ac.eng.hokudai.ac.jp/index_e.html		c.jp/index_e.html		
调号 Additional Information	備考 Additional Information			

科目名 Course Title	化学計測学特	論[Instrumentation Chemistry]	
講義題目 Subtitle			
責任教員 Instructor	長谷川 靖哉	[HASEGAWA Yasuchika] (大学院工学研究	院)
担当教員 Other Instructors			17 - 1
科目種別 Course Type			
開講年度 Year	2024	時間割番号 Course Number	094067
期間 Semester	Intensive	単位数 Number of Credits	1
授業形態 Type of Class	Lecture	対象年次 Year of Eligible Student	~
対象学科・クラス Eligible Dep	oartment/Class		
ナンバリングコード Numberir	ng Code	CHEM_REQEL 5191	
補足事項 Other Information			
授業実施方式 Class Method		1 対面授業科目《対面のみ》	
キーワード Key Words			
Chemical Information, elemen	ntal analysis, condi	itional analysis, structural analysis in nano-	and micro–area.
授業の目標 Course Objectiv	/es		
Grounding in physical, organi	ic and inorganic ch	nemistry.	
In this course, instrumentati	on chemistry cont	aining elemental analysis, configurational a	nalysis, structural analysis in nano-

In this course, instrumentation chemistry containing elemental analysis, configurational analysis, structural analysis in nanoand micro-area are introduced. Based on their studies, students learn fundamental knowledges and various information about chemical analysis of organic and inorganic materials.

到達目標 Course Goals

Students learn principle, variety and characterization of instrumentation chemistry for material analysis. Based on instrumentation chemistry containing elemental analysis, configurational analysis, structural analysis in nano- and micro-area, students make the most of their knowledges for construction of their chemical research.

授業計画 Course Schedule

1-2. introduction of instrumentation chemistry: importance for structural analysis on the material surface, classification of chemical instruments, grounding in high vacuum engineering

3. configurational analysis (TEM, SEM, AFM, STM)

4. elemental analysis (AES, EPMA, XPS, XRF)

5. structural analysis (XRD, EXAFS, HEED, LEED, SAXS)

6. photo-physical analysis (UV-Vis absorption spectra, fluorescence and phosphorescence spectra, emission lifetime, Raman spectra)

7. MS spectral analysis (EI-MS, CI-MS, ESI-MS, MALDI-MS, SIMS)

8. examination

準備学習 (予習・復習)等の内容と分量 Homework

Pre-examination for review of instrumentation chemistry

成績評価の基準と方法 Grading System

The attendance rate must be over 70% to be qualified to take the final exam. Evaluations will be made based on (1) learning attitude (20%), (2) exercise (10%), (3) final examination scores (70%).

他学部履修の条件 Other Faculty Requirements

テキスト・教科書 Textbooks

講義指定図書 Reading List

参照ホームページ Websites

This course will be provided as part of the Hokkaido Summer Institute., For more information (invited lecturers, coursedetails,etc.),pleasevisitthewebsitebelow:,https://hokkaidosummerinstitute.oia.hokudai.ac.jp/en/courses/CourseDetail=G051

研究室のホームページ Websites of Laboratory

https://www.eng.hokudai.ac.jp/labo/amc/en/index.html

科目名 Course Title	利受公理之合性	Advanced Ethics and Safety for Science	a and Engineering]
構義題目 Subtitle	科子倫理女王符詞	侖[Advanced Ethics and Safety for Scienc	
	+火→→ =米 . 白7 「\/	ATSUMOTO K	百匹之
責任教員 Instructor 担当教員 Other Instructors	中川 浩行(京都)	ATSUMOTO Kenichiro] (大学院工学研タ ナ学)	
科目種別 Course Type	十7月 1日11 (水相)	八十	
開講年度 Year	2024	時間割番号 Course Number	094068
期間 Semester	Intensive	単位数 Number of Credits	1
授業形態 Type of Class	Lecture	对象年次 Year of Eligible Student	\sim
対象学科・クラス Eligible Depa			
ナンバリングコード Numbering		CHEM_REQEL 5210	
補足事項 Other Information			
授業実施方式 Class Method		1 対面授業科目《対面のみ》	
キーワード Key Words			
Engineering Ethics, Safety Eng	ineering		
授業の目標 Course Objectives			
Students will learn fundamental	s of ethics and safe	ty engineering for scientists and engineer	rs. In the ethics education, students
		ence and technology on society and na	
scientists and engineers owes	to the society. In s	safety education, students will learn risk	avoidance, safety related laws and
process safety design methods,	through various ex	amples. By understanding these, student	s will deepen the knowledge to take
responsible judgments and action	ons, that are essent	ial to be a self-independent scientist or e	engineer.
到達目標 Course Goals			
By taking this course, students	-		
-		vith consideration of safty, when a proc	es technology is introduced to the
society to enrich the human so	2		
2. undestand ethics and morals	as a scientist or en	gineer.	
授業計画 Course Schedule			
1. Basis of engineering ethics (
	thics and role of sc	ientists and engineers. Understand techn	ique and structure for taking ethical
behavior.			
2. Sefety engineering and press	an design (6 nonied		
2. Safety engineering and proce		the hazards caused by handled substanc	as and risk control techniques, and
the purpose and outline of safe			es and fisk control techniques, and
Learn basis of process safety d			
準備学習(予習・復習)等の内容		{	
Lecture materials will be distrib			
One credit for a lecture is give	n for 45 hours of st	udy. Since the actual lecture is 90 minute	es (counted as 2 hours) $ imes$ 8 periods
= 16 hours, the credit acquisit	tion requires about	4 hours review per period. Keep in min	d this point and review the lecture
using the lecture materials.			
成績評価の基準と方法 Gradin	g System		
For grade evaluation, students	-		
		shment based on the submitted assignmen	t.
他学部履修の条件 Other Face	ulty Requirements		
テキスト・教科書 Textbooks			
講義指定図書 Reading List			
参照ホームページ Websites			
研究室のホームページ Websit	es of Laboratory		
備考 Additional Information			
	edit of ″Engineer e	thics and safety" of Department of Appl	ied Science and Engineering cannot
take this lecture.			

科目名 Course Title	公本	也道沙[I sharstowy Evension in Chemical So	ionoog and Engineering II]
料日名 Course Title 講義題目 Subtitle	松石化子夫映	指導法[Laboratory Exercise in Chemical So	
請我題日 Subuue 青任教員 Instructor	※ 今 ル 学 陸 体	業長 (十学院総合化学院)	
員任教員 Instructor 担当教員 Other Instructors	芯石化子阮10 Provided by su	議員(大学院総合化学院)	
科目種別 Course Type	TTOVIded by St	thet visor	
開講年度 Year	2024	時間割番号 Course Number	
期間 Semester	Full Year	単位数 Number of Credits	2
授業形態 Type of Class	Experiment	对象年次 Year of Eligible Student	1~2
対象学科・クラス Eligible Depa	*		1 2
ナンバリングコード Numbering		CHEM REQEL 5302	
補足事項 Other Information			
授業実施方式 Class Method		1 対面授業科目《対面のみ》	
キーワード Key Words			
Teaching skills: teaching assist	tant		
授業の目標 Course Objective			
Graduate students are reques	sted to teach ur	ndergraduate-level laboratory experiments.	This course examines how to gain
teaching abilities and skills in a	conducting chemi	ical experiments.	
到達目標 Course Goals			
Through the course, the stu	dents will be ab	ble to gain proper abilities and skills to	teach undergraduate-level chemical
experiments.			
授業計画 Course Schedule			
On the basis of evaluation of s	tudent's achieve	ments, the course offers on-the-job-trainin	g to
– gain fundamental principle/k	nowledge on a gi	ven chemical experiment and abilities/skills	to operate/conduct the experiment
– gain teaching abilities/skills	to undergraduate	e-level students	
– play leadership in teaching a			
準備学習 (予習・復習)等の内		vork	
Daily preparatory works for te	_		
成績評価の基準と方法 Gradin			
Evaluate based on daily achiev		· · · ·	
他学部履修の条件 Other Fac	uity Requiremen	ts	
テキスト・教科書 Textbooks			
講義指定図書 Reading List			
参照ホームページ Websites			
研究室のホームページ Websi	tes of Laborator	y	
備考 Additional Information			
Register this course at the sen	nester of graduat	ion.	

허민성 구매			· · · · · · · · · · · · · · · · · · ·
科目名 Course Title	総合化字実験切	ff究法[Laboratory Exercise in Chemical Sc	lences and Engineering III
講義題目 Subtitle	40 A H MARL 11 -		
責任教員 Instructor		§員(大学院総合化学院)	
担当教員 Other Instructors	Provided by sup	ervisor	
科目種別 Course Type			
開講年度 Year	2024	時間割番号 Course Number	
期間 Semester	Full Year	単位数 Number of Credits	2
授業形態 Type of Class	Seminar	対象年次 Year of Eligible Student	1~2
対象学科 クラス Eligible Depa	rtment/Class		
ナンバリングコード Numbering	Code	CHEM_REQEL 5312	
補足事項 Other Information			
授業実施方式 Class Method		1 対面授業科目《対面のみ》	
キーワード Key Words			
Experimental skills: Teaching s	kills: Presentatior	n skills	
授業の目標 Course Objective			
		ges and experiences on various chemical	experiments and to manage his/her
		o manage various chemical research and t	
both Japanese and English.			-
到達目標 Course Goals			
Through the course, students v	vill be able to		
– gain experimental and presen	tation skills/abilit	ies	
– play leadership in research w	orks		
授業計画 Course Schedule			
On the basis of evaluating stud	ent's achievemen	ts, the course offers the on-the-job-training	ng to
- understand fundamental princ	ciples of chemical	experiments	
- gain experiences in chemical	•		
- gain presentation abilities/sk	-	ese and English	
– play leadership in each resea	-	0	
準備学習 (予習・復習)等の内容		rk	
Daily preparatory works on lab	oratory experimer	nts	
成績評価の基準と方法 Gradin			
Evaluate based on daily achieve	ements (50%) and	seasonal reports (50%)	
他学部履修の条件 Other Face	ulty Requirements	3	
テキスト・教科書 Textbooks			
講義指定図書 Reading List			
参照ホームページ Websites			
研究室のホームページ Websit	es of Laboratory		
備考 Additional Information			
	oston of moducti-	N	
Register this course at the sem	lester of graduatic	011.	

科目名 Course Title 講義題目 Subtitle	公工ル受(生	端物理化学)[Molecular Chemistry (Advance	d Physical Chomistry)		
	- フリ 1 16子 (元	コックローク Liviolecular Chemistry (Advance	eu i nysical Chennistry/]		
青任教員 Instructor	七十十十十二十十二十二十二十二十二十二十二十二十二十二十二十二十二十二十二十二	旧林(05世长記)(十学院理学研究院)			
担当教員 Other Instructors	村越 敬 [MURAKOSHI Kei] (大学院理学研究院) FUKUSHIMA Tomohiro (理学研究院)				
科目種別 Course Type	FOROSTIME	和10111011110(建于制无序)			
開講年度 Year	2024	時間割番号 Course Number	094101		
期間 Semester	Fall	単位数 Number of Credits	1		
授業形態 Type of Class	Lecture	对象年次 Year of Eligible Student	\sim		
対象学科・クラス Eligible Dep					
ナンバリングコード Numbering		CHEM_ELMOL 6002			
補足事項 Other Information	g oode				
授業実施方式 Class Method		2 対面授業科目《一部遠隔》			
キーワード Key Words					
· ·	electronic stru	cture, Surface morphology, Surface spectrosc	copy Catalysis		
授業の目標 Course Objective		etare, carace morphology, carace spectrose			
		lsorption and catalytic reaction occur at the	solid surface due to the interaction		
		c knowledge and latest research to unde			
properties.		Ŭ			
到達目標 Course Goals					
Understand the intermolecula	ar force and the	structure and electronic state of the solid s	surface. Understand the origin of the		
unique physical properties of	the surface / in	nterface. In addition, we also acquire basic ki	nowledge on advanced nanostructure		
analysis methods to understar	nd surface sciend	ce from physicochemical point of view.			
授業計画 Course Schedule					
(1) Structure and electronic st	tate of solid surf				
	care of conta car.	face			
(2) Foundations of atomic and					
	l intermolecular		canning tunneling microscope, etc.)		
(3) Outline of the latest surface 準備学習(予習・復習)等の内	l intermolecular ce / interface ev]容と分量 Home	forces valuation method (atomic force microscope, se	canning tunneling microscope, etc.)		
(3) Outline of the latest surface 準備学習(予習・復習)等の内 Homework will be handed out	l intermolecular ce / interface ev 日容と分量 Home in the class.	forces valuation method (atomic force microscope, se	canning tunneling microscope, etc.)		
(3) Outline of the latest surface 準備学習(予習・復習)等の内 Homework will be handed out 成績評価の基準と方法 Grad	l intermolecular ce / interface ev]容と分量 Home in the class. ing System	forces valuation method (atomic force microscope, so swork	canning tunneling microscope, etc.)		
(3) Outline of the latest surface 準備学習(予習・復習)等の内 Homework will be handed out 成績評価の基準と方法 Grading Grading will be evaluated base	l intermolecular ce / interface ev 日容と分量 Home in the class. ing System ed on attendance	forces valuation method (atomic force microscope, se work e and homeworks.	canning tunneling microscope, etc.)		
(3) Outline of the latest surface 準備学習(予習・復習)等の内 Homework will be handed out 成績評価の基準と方法 Grad	l intermolecular ce / interface ev 日容と分量 Home in the class. ing System ed on attendance	forces valuation method (atomic force microscope, so swork e and homeworks.	canning tunneling microscope, etc.)		
(3) Outline of the latest surface 準備学習 (予習・復習)等の内 Homework will be handed out 成績評価の基準と方法 Gradi Grading will be evaluated base 他学部履修の条件 Other Face	l intermolecular ce / interface ev 日容と分量 Home in the class. ing System ed on attendance	forces valuation method (atomic force microscope, so swork e and homeworks.	canning tunneling microscope, etc.)		
(3) Outline of the latest surface 準備学習(予習・復習)等の内 Homework will be handed out 成績評価の基準と方法 Grading Grading will be evaluated base	l intermolecular ce / interface ev 日容と分量 Home in the class. ing System ed on attendance	forces valuation method (atomic force microscope, so swork e and homeworks.	canning tunneling microscope, etc.)		
(3) Outline of the latest surface 準備学習 (予習・復習)等の内 Homework will be handed out 成績評価の基準と方法 Gradi Grading will be evaluated base 他学部履修の条件 Other Face テキスト・教科書 Textbooks	l intermolecular ce / interface ev 日容と分量 Home in the class. ing System ed on attendance	forces valuation method (atomic force microscope, so swork e and homeworks.	canning tunneling microscope, etc.)		
(3) Outline of the latest surface 準備学習 (予習・復習)等の内 Homework will be handed out 成績評価の基準と方法 Gradi Grading will be evaluated base 他学部履修の条件 Other Face	l intermolecular ce / interface ev 日容と分量 Home in the class. ing System ed on attendance	forces valuation method (atomic force microscope, so swork e and homeworks.	canning tunneling microscope, etc.)		
(3) Outline of the latest surface 準備学習(予習・復習)等の内 Homework will be handed out 成績評価の基準と方法 Gradi Grading will be evaluated base 他学部履修の条件 Other Face テキスト・教科書 Textbooks 講義指定図書 Reading List	l intermolecular ce / interface ev 日容と分量 Home in the class. ing System ed on attendance	forces valuation method (atomic force microscope, so swork e and homeworks.	canning tunneling microscope, etc.)		
(3) Outline of the latest surface 準備学習 (予習・復習)等の内 Homework will be handed out 成績評価の基準と方法 Gradi Grading will be evaluated base 他学部履修の条件 Other Face テキスト・教科書 Textbooks	l intermolecular ce / interface ev 日容と分量 Home in the class. ing System ed on attendance	forces valuation method (atomic force microscope, so swork e and homeworks.	canning tunneling microscope, etc.)		
 (3) Outline of the latest surface 準備学習(予習・復習)等の内 Homework will be handed out 成績評価の基準と方法 Gradi Grading will be evaluated base 他学部履修の条件 Other Face テキスト・教科書 Textbooks 講義指定図書 Reading List 参照ホームページ Websites 研究室のホームページ Webs 	l intermolecular ce / interface ev 3容と分量 Home in the class. ing System ed on attendance culty Requireme ites of Laborato	forces valuation method (atomic force microscope, so work e and homeworks.	canning tunneling microscope, etc.)		
 (3) Outline of the latest surface 準備学習(予習・復習)等の内 Homework will be handed out 成績評価の基準と方法 Gradid Grading will be evaluated base 他学部履修の条件 Other Face テキスト・教科書 Textbooks 講義指定図書 Reading List 参照ホームページ Websites 研究室のホームページ Webs https://wwwchem.sci.hokudai 	l intermolecular ce / interface ev 3容と分量 Home in the class. ing System ed on attendance culty Requireme ites of Laborato	forces valuation method (atomic force microscope, so work e and homeworks.	canning tunneling microscope, etc.)		
 (3) Outline of the latest surface 準備学習(予習・復習)等の内 Homework will be handed out 成績評価の基準と方法 Gradi Grading will be evaluated base 他学部履修の条件 Other Face テキスト・教科書 Textbooks 講義指定図書 Reading List 参照ホームページ Websites 研究室のホームページ Webs 	l intermolecular ce / interface ev 3容と分量 Home in the class. ing System ed on attendance culty Requireme ites of Laborato	forces valuation method (atomic force microscope, so work e and homeworks.	canning tunneling microscope, etc.)		

科目名 Course Title	ハスル労任	有機構造化学特論)[Molecular Chemistry	(Structural and Deviced Organia
	Chemistry)]	月機構垣佔子符論)[Molecular Chemistry	(Structural and Physical Organic
講義題目 Subtitle			
責任教員 Instructor	鈴木 孝紀 [SUZUKI Takanori] (大学院理学研究院)	
担当教員 Other Instructors			
科目種別 Course Type			
開講年度 Year	2024	時間割番号 Course Number	094102
期間 Semester	Winter	単位数 Number of Credits	1
授業形態 Type of Class	Lecture	対象年次 Year of Eligible Student	~
対象学科・クラス Eligible De	partment/Class		
ナンバリングコード Numberi	ing Code	CHEM_ELMOL 6000	
補足事項 Other Information	ו		
授業実施方式 Class Metho	d	4 遠隔授業科目《遠隔のみ》	
キーワード Key Words			
Structural Organic Chemistr	ry		
授業の目標 Course Object	ives		
Various functions of materia	ials can be deriv	ed by proper designing organic pi-electron	systems. This course will provide
students with the two of the	important concep	ots which are necessary to comprehend this a	rea of organic chemistry.
到達目標 Course Goals			
Students will learn the back	ground and basic	idea to understand the various intriguing phe	nomena in the functionalized organic
pi-electron systems/organic	solids.		
授業計画 Course Schedule			
授業計画 Course Schedule Two major topic are as follo	ws:		
授業計画 Course Schedule Two major topic are as follo 1) "Disappearance of poly	ws:	ing behavior of crystallizaion, rapid/reluct	ant phase transition of crystalline
授業計画 Course Schedule Two major topic are as follo 1) "Disappearance of poly materials	ws: morphs": Intrigu		-
授業計画 Course Schedule Two major topic are as follo 1) "Disappearance of poly materials 2) "Orbital interaction thro	ws: morphs": Intrigu	ing behavior of crystallizaion, rapid/reluct gh space": extremely long C-C bond, X-1	-
授業計画 Course Schedule Two major topic are as follo 1) "Disappearance of poly materials	ws: morphs": Intrigu		-
授業計画 Course Schedule Two major topic are as follo 1) "Disappearance of poly materials 2) "Orbital interaction thro optimized structure	ws: morphs": Intrigu pugh bonds/throu	gh space": extremely long C-C bond, X-n	-
授業計画 Course Schedule Two major topic are as follo 1) "Disappearance of poly materials 2) "Orbital interaction thro optimized structure The class instruction will be	ws: morphs": Intrigu ough bonds/throu done in Japanese	gh space": extremely long C-C bond, X-n	-
授業計画 Course Schedule Two major topic are as follor 1) "Disappearance of poly materials 2) "Orbital interaction thro optimized structure The class instruction will be 準備学習 (予習・復習)等の	ws: morphs": Intrigu ough bonds/throu done in Japanese 内容と分量 Home	gh space": extremely long C-C bond, X-n work	-
授業計画 Course Schedule Two major topic are as follor 1) "Disappearance of poly materials 2) "Orbital interaction thro optimized structure The class instruction will be 準備学習 (予習・復習)等の Printed material will be hand	ws: morphs": Intrigu ough bonds/throu done in Japanese 内容と分量 Home led out in the clas	gh space": extremely long C-C bond, X-n work	-
授業計画 Course Schedule Two major topic are as folloo 1) "Disappearance of poly materials 2) "Orbital interaction thro optimized structure The class instruction will be 準備学習 (予習・復習)等の Printed material will be hand 成績評価の基準と方法 Gra	ws: morphs": Intrigu ough bonds/throu done in Japanese 内容と分量 Home led out in the clas	gh space": extremely long C-C bond, X-n work	-
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授業計画 Course Schedule Two major topic are as folloo 1) "Disappearance of poly materials 2) "Orbital interaction thro optimized structure The class instruction will be 準備学習 (予習・復習)等の Printed material will be hand 成績評価の基準と方法 Gra	ws: morphs": Intrigu bugh bonds/throu done in Japanese 内容と分量 Home led out in the clas iding System	gh space": extremely long C-C bond, X-n work s	-
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授業計画 Course Schedule Two major topic are as folloo 1) "Disappearance of poly materials 2) "Orbital interaction thro optimized structure The class instruction will be 準備学習 (予習・復習)等の Printed material will be hand 成績評価の基準と方法 Gra Presentations and reports	ws: morphs": Intrigu ough bonds/throu done in Japanese 内容と分量 Home ded out in the clas iding System Faculty Requireme	gh space": extremely long C-C bond, X-n work s	-
授業計画 Course Schedule Two major topic are as follo 1) "Disappearance of poly materials 2) "Orbital interaction thro optimized structure The class instruction will be 準備学習 (予習・復習)等の Printed material will be hand 成績評価の基準と方法 Gra Presentations and reports 他学部履修の条件 Other F テキスト・教科書 Textbooks	ws: morphs": Intrigu bugh bonds/throu done in Japanese 内容と分量 Home led out in the clas iding System	gh space": extremely long C-C bond, X-n work s	-
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授業計画 Course Schedule Two major topic are as follo 1) "Disappearance of poly materials 2) "Orbital interaction thro optimized structure The class instruction will be 準備学習 (予習・復習)等の Printed material will be hand 成績評価の基準と方法 Gra Presentations and reports 他学部履修の条件 Other F テキスト・教科書 Textbooks 講義指定図書 Reading List	ws: morphs": Intrigu bugh bonds/throu done in Japanese 内容と分量 Home led out in the clas iding System aculty Requireme	gh space": extremely long C-C bond, X-n work s nts	-
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授業計画 Course Schedule Two major topic are as follo 1) "Disappearance of poly materials 2) "Orbital interaction thro optimized structure The class instruction will be 準備学習(予習・復習)等の Printed material will be hand 成績評価の基準と方法 Gra Presentations and reports 他学部履修の条件 Other F テキスト・教科書 Textbooks 講義指定図書 Reading List 構造有機化学 基礎から物 参照ホームページ Websites	ws: morphs": Intrigu bugh bonds/throu done in Japanese 内容と分量 Home led out in the clas iding System faculty Requireme faculty Requireme	gh space": extremely long C-C bond, X-n work s nts チまで/中筋 一弘:東京化学同人, 2020	-

科目名 Course Title	分子化学(高	分子機能科学)[Molecular Chemistry (Macro	omolecular Science)]
講義題目 Subtitle			
責任教員 Instructor	中野 環「NA	KANO Tamaki] (触媒科学研究所)	
担当教員 Other Instructors			
科目種別 Course Type	-		
開講年度 Year	2024	時間割番号 Course Number	094103
期間 Semester	Spring	単位数 Number of Credits	1
授業形態 Type of Class	Lecture	对象年次 Year of Eligible Student	
は来が感 Type of Olass 対象学科・クラス Eligible Depa		对象十次 Tear of Lingible Student	
対象チャークラス Eligible Depa ナンバリングコード Numbering		CLIEM ELMOL 6002	
アンファンシュート Numbering 補足事項 Other Information	Code	CHEM_ELMOL 6002	
授業実施方式 Class Method		2 対面授業科目《一部遠隔》	
キーワード Key Words			
		formation, Optically Active, Chirality, Helix	
授業の目標 Course Objective			
Synthesis, structure, and func	ctions of various	polymers will be introduced. A focus will	be on polymer chirality. Aiming to
understand the basic and adv	anced concepts	of polymer stereochemistry, we will discus	ss examples of polymers and related
small molecules.			
到達目標 Course Goals			
Students aim to learn basic a	and advanced co	oncepts of synthesis, structure, and proper	rties of polymers. In addition, they
understand concepts of genera	al chirality, exten	nd their understanding to polymer chirality,	and obtain in-depth insights into the
relation between chiral functio	ns and chiral str	uctures of polymers.	
授業計画 Course Schedule			
Beginning from the basis aspec	ets of polymer sy	vnthesis and its classification, we discuss pol	lymer structure and functions with an
emphasis on chirality. The pla	nned contents a	re as follows:	
1. Basics of polymer science (1	1)		
2. History of polymer science	(1)		
3. Polymer structure: structu	ral features unio	que to polymers such as mola mass disper-	sity, tacticity (stereoregularity), and
helicity. Nomenclature, classifi	ication, and anal	ytical methodologies (2)	
4. Synthesis of chiral polymers	: asymmetric po	lymerization (2)	
5. Functions of chiral polymers	s: structure-pro	perty relations (2)	
準備学習 (予習・復習)等の内	容と分量 Home	work	
Students are asked to read th	nrough literature	e relevant to polymer synthesis and polyme	r chirality and summarize the points
that they wish to discuss in th	e class. After e	ach class, they are asked to find and read jo	ournal articles that are related to the
contents of class teaching and	discussions.		
成績評価の基準と方法 Gradin	ng System		
Evaluation will be conducted k	based on report	papers submitted after all planned class tea	ching is finished and also on attitude
toward learning.			
他学部履修の条件 Other Fac	ulty Requirement	nts	
テキスト・教科書 Textbooks			
Polymer Chemistry: An Introd	uction (3rd Ed.)	/Malcom P. Stevens:Oxford, 1999	
	NTC 2000		
高分子化学入門/蒲池幹治:	NIS, 2009		
		田清蔵:講談社,1997	
大学院高分子科学/野瀬卓平		田清蔵:講談社,1997	
大学院高分子科学/野瀬卓平 蒲義指定図書 Reading List	2、中浜精一、宮		
大学院高分子科学/野瀬卓平 講義指定図書 Reading List pi-Stacked Polymers and Mole	平、中浜精一、宮 ecules/T. Naka		
大学院高分子科学/野瀬卓马 講義指定図書 Reading List pi-Stacked Polymers and Mole Stereochemistry of Organic Co	平、中浜精一、宮 ecules/T. Nakan ompounds/E. L	no Ed.:Springer, 2014 . Eliel, S. H. Wilen:Wiley, 1994	
大学院高分子科学/野瀬卓斗 講義指定図書 Reading List pi-Stacked Polymers and Mole Stereochemistry of Organic Co NMR Spectroscopy of Polymer	Z、中浜精一、宮 ccules/T. Naka ompounds/E. L rs/K. Hatada, T	no Ed.:Springer, 2014 . Eliel, S. H. Wilen:Wiley, 1994	r, 1997
大学院高分子科学/野瀬卓斗 講義指定図書 Reading List pi-Stacked Polymers and Mole Stereochemistry of Organic Co NMR Spectroscopy of Polymer Macromolecular Design of Poly	平、中浜精一、宮 cules/T. Naka ompounds/E. L rs/K. Hatada, J ymeric Materials	no Ed.:Springer, 2014 . Eliel, S. H. Wilen:Wiley, 1994 F. Kitayama:Springer, 2004	r, 1997
大学院高分子科学/野瀬卓平 講義指定図書 Reading List pi-Stacked Polymers and Mole Stereochemistry of Organic Co NMR Spectroscopy of Polymer Macromolecular Design of Poly Protein Structure and Function	平、中浜精一、宮 ccules/T. Nakap ompounds/E. L rs/K. Hatada, T ymeric Materials n/G. A. Petsko	no Ed.:Springer, 2014 Eliel, S. H. Wilen:Wiley, 1994 Г. Kitayama:Springer, 2004 /K. Hatada, T. Kitayama, O. Vogl:Dekker p, D. Ringe:New Science Press, 2004	r, 1997
大学院高分子科学/野瀬卓平 講義指定図書 Reading List pi-Stacked Polymers and Mole Stereochemistry of Organic Co NMR Spectroscopy of Polymer Macromolecular Design of Poly Protein Structure and Function Circular Dichroism/N. Berov	平、中浜精一、宮 ccules/T. Nakap ompounds/E. L rs/K. Hatada, T ymeric Materials n/G. A. Petsko	no Ed.:Springer, 2014 Eliel, S. H. Wilen:Wiley, 1994 f. Kitayama:Springer, 2004 ⁄K. Hatada, T. Kitayama, O. Vogl:Dekker	r, 1997
大学院高分子科学/野瀬卓平 講義指定図書 Reading List pi-Stacked Polymers and Mole Stereochemistry of Organic Co NMR Spectroscopy of Polymer Macromolecular Design of Poly Protein Structure and Function Circular Dichroism/N. Berow	平、中浜精一、宮 ccules/T. Nakap ompounds/E. L rs/K. Hatada, T ymeric Materials n/G. A. Petsko	no Ed.:Springer, 2014 Eliel, S. H. Wilen:Wiley, 1994 Г. Kitayama:Springer, 2004 /K. Hatada, T. Kitayama, O. Vogl:Dekker p, D. Ringe:New Science Press, 2004	r, 1997
NMR Spectroscopy of Polymer Macromolecular Design of Poly Protein Structure and Function Circular Dichroism/N. Berov 参照ホームページ Websites	Z、中浜精一、宮 cules/T. Nakar ompounds/E. L rs/K. Hatada, ∩ ymeric Materials n/G. A. Petsko a, K. Nakahishi,	no Ed.:Springer, 2014 . Eliel, S. H. Wilen:Wiley, 1994 F. Kitayama:Springer, 2004 /K. Hatada, T. Kitayama, O. Vogl:Dekker o, D. Ringe:New Science Press, 2004 R. W. Woody:Wiley-VCH, 2000	r, 1997
大学院高分子科学/野瀬卓马 講義指定図書 Reading List pi-Stacked Polymers and Mole Stereochemistry of Organic Co NMR Spectroscopy of Polymer Macromolecular Design of Poly Protein Structure and Function Circular Dichroism/N. Berov 参照ホームページ Websites 研究室のホームページ Websit	平、中浜精一、宮 ecules/T. Nakap ompounds/E. L rs/K. Hatada, T ymeric Materials n/G. A. Petsko a, K. Nakahishi, tes of Laborator	no Ed.:Springer, 2014 ∴ Eliel, S. H. Wilen:Wiley, 1994 Γ. Kitayama:Springer, 2004 ∕K. Hatada, T. Kitayama, O. Vogl:Dekker p, D. Ringe:New Science Press, 2004 R. W. Woody:Wiley-VCH, 2000 γ	r, 1997
大学院高分子科学/野瀬卓马 講義指定図書 Reading List bi-Stacked Polymers and Mole Stereochemistry of Organic Co NMR Spectroscopy of Polymer Macromolecular Design of Poly Protein Structure and Function Circular Dichroism/N. Berov 参照ホームページ Websites	平、中浜精一、宮 ecules/T. Nakap ompounds/E. L rs/K. Hatada, T ymeric Materials n/G. A. Petsko a, K. Nakahishi, tes of Laborator	no Ed.:Springer, 2014 ∴ Eliel, S. H. Wilen:Wiley, 1994 Γ. Kitayama:Springer, 2004 ∕K. Hatada, T. Kitayama, O. Vogl:Dekker p, D. Ringe:New Science Press, 2004 R. W. Woody:Wiley-VCH, 2000 γ	r, 1997

科目名 Course Title	分子化学(触频	集理論)[Molecular Chemistry (Catalysis The	eory)]		
講義題目 Subtitle					
責任教員 Instructor	長谷川 淳也	[HASEGAWA Junya] (触媒科学研究所)			
担当教員 Other Instructors	IIDA Kenji (触	媒科学研究所), SHROTRI Abhijit (触媒科学	ž研究所),		
	MIYAZAKI Ra	y (触媒科学研究所)			
科目種別 Course Type					
開講年度 Year	2024	時間割番号 Course Number	094104		
期間 Semester	Winter	単位数 Number of Credits	1		
授業形態 Type of Class	Lecture	対象年次 Year of Eligible Student	~		
対象学科・クラス Eligible Depa	artment/Class				
ナンバリングコード Numbering	ing Code CHEM_ELMOL 6002				
補足事項 Other Information					
授業実施方式 Class Method		1 対面授業科目《対面のみ》			
キーワード Key Words					

Catalysis, Catalytic chemistry, Theoretical and computational chemistry of catalysis

授業の目標 Course Objectives

Catalysts are key materials for the effective utilization of resources and energy and for the resolution of environmental issues. On the other hand, the challenge is in the difficulty in developing catalysts and controlling catalytic reactions due to the various complexities such as active site structure, components of reactions, reaction mechanism, reaction mode, etc. In this lecture, you will learn about methods for understanding basic catalytic phenomena using theoretical and computational chemistry. Although solid catalysts are mainly dealt with, we will theoretically consider the general catalytic principles including molecular catalysis and biocatalysis.

到達目標 Course Goals

By the end of this course, you are able to apply your theoretical perspective to the catalytic phenomena. Specifically, you can theoretically recognize the energetics, kinetics, equilibrium, electronic theory, and properties of catalyst materials of catalytic phenomena. In addition, students understand theoretical calculation approaches that are useful for developing catalyst materials to optimize catalytic reactions. On the other hand, you will gain deeper knowledges through the presentations and question-and-answer sessions in a round lecture format.

授業計画 Course Schedule

Students read through an English textbook to learn the basics of catalyst theory. This course will be held in a round lecture format. Each student summarize the content and give a presentation. Teachers give additional explanation and supplements. The specific contents are as follows.

Part 1 Introduction, potential energy diagrams: Adsorption, reaction, diffusion, surface dependence (chapters 1 and 2)

Part 2 Chemical equilibrium on surfaces: Adsorption isotherms, free energy diagrams (chapter 3)

Part 3 Rate constant: Time scale of chemical reactions, transition state theory (Chapter 4)

Part 4 Kinetics: Microscopic kinetics, application to ammonia synthesis reaction (Chapter 5)

Part 5 Energy trends in catalysis, mapping of catalytic activity: Scaling relationships, activity maps, selectivity maps, Sabatier analysis (chapters 6 and 7)

Part 6 Electronic factors: band structure, d-band model, relation between reaction and electronic structure, ensemble effect, ligand effect (chapters 8 and 12)

Part 7 Catalyst structure, catalyst poisons and promoters: Structure of real catalysts, poisoning and promotion of catalysts (chapters 9 and 10)

Part 8 Surface Electrocatalysis: Solid-electrolyte interface, interfacial electron transfer, hydrogen electrode model, potential dependence of rate, overpotential, limiting potential (Chapter 11)

準備学習 (予習・復習)等の内容と分量 Homework

Read the relevant sections of the textbook for each class in advance to acquire an overview of the knowledge required for discussion. Create a presentation file summarizing the content for the part you are responsible for, print out copies for each person, and present using a computer. One or two practice problems from the textbook may be assigned as a review.

成績評価の基準と方法 Grading System

The learning achievement will be evaluated comprehensively based on (1)presentation (levels of presentation skill, logic, comprehension, etc.), (2) discussion (positivity and quality of comments, etc.), and (3) learning attitude.

他学部履修の条件 Other Faculty Requirements

テキスト・教科書 Textbooks

Fundamental Concepts in Heterogeneous Catalysis/Jens K. Nørskov, Felix Studt, Frank Abild-Pedersen, Thomas Bligaard: Wiley, 2014

講義指定図書 Reading List

参照ホームページ Websites

研究室のホームページ Websites of Laboratory

https://www.cat.hokudai.ac.jp/hasegawa/

備考 Additional Information

Basic knowledges of physical chemistry, inorganic chemistry, and organic chemistry are prerequisite for taking this course.

科目名 Course Title	分子化学(光化学)[Molecular Chemistry (Photochemistry)]				
講義題目 Subtitle					
責任教員 Instructor	上野 貢生 [UENO Kosei] (大学院理学研究院)				
担当教員 Other Instructors					
科目種別 Course Type					
開講年度 Year	2024	時間割番号 Course Number	094105		
期間 Semester	Spring	単位数 Number of Credits	1		
授業形態 Type of Class	Lecture	対象年次 Year of Eligible Student	\sim		
対象学科・クラス Eligible Depar					
ナンバリングコード Numbering					
補足事項 Other Information					
授業実施方式 Class Method		1 対面授業科目《対面のみ》			
キーワード Key Words					
Electronically Excited State	o. Fluorosconco/I	Phosphorescence: Nonradiative Process	es: Photophysical Processes:		
Photochemical Reactions: Spec		-nosphorescence. Noniadiative Frocess	es. Fliotophysical Flocesses.		
授業の目標 Course Objectives					
		and the physicochemical processes from t	he excited states which are the		
basis of photochemistry of orga			he excited states which are the		
到達目標 Course Goals	file indicedies are st	uuleu.			
	ical reactions and	physicochemical phenomena are studied	by learning the nature of the		
*		various physicochemical precesses from th	0		
usage of related spectroscopy a		various physicochemical processes from th	le excited states. I finciples and		
授業計画 Course Schedule	ite also learneu.				
	phomical and phot	ophysical processes of organic compound	ls Fundamental background of		
		nain topics of the course is as follows.	is. Fundamental background of		
		glet and triplet states 3) Radiative (fluo	rosconco/phosphorosconco) and		
		tersystem crossing) 4) Characteristics			
		hysicochemical information obtained from			
		ectrum, emission yield, lifetime, and dynamic			
		transfer 8) State-of-the-art of photochemi			
Thotochemical reactions () The	to induced electroi	i transfer of state of the art of photochem	lear researches		
進供学習 (圣羽・復羽)年の内2	첫 나스 목 Hamawark				
準備学習(予習・復習)等の内容			in an latical characteristics at the		
	courses on physi	cal chemistry and instrumental methods	in analytical chemistry at the		
undergraduate school.					
成績評価の基準と方法 Grading System Assignments in classes (30 %), attitude to learning in classes (20 %), and term-end report (homework) (50 %)					
他学部履修の条件 Other Faculty Requirements					
B于印度多00米IT Outer Faculty Requirements					
テキスト・教科書 Textbooks					
テイスト・教科書 Textbooks					
講義指定図書 Reading List					
参照ホームページ Websites					
研究室のホームページ Websites of Laboratory					
https://www.chem.sci.hokudai.ac.jp/~bunseki/					
備考 Additional Information					
Recommended textbook 1) "Principles of Molecular Photochemistry: An Introduction", N. J. Turro et al., University Science					

Books, 2009. 2)「光化学 I」, 井上晴夫他著, 丸善, 1999.

科目名 Course Title	ハスル営(ル)	学员内创成学校验》[Malaaulan Chamiatmy ()	Advanced Chemical Reaction Design
14 H A Course little	分子化学(化学反応創成学特論)[Molecular Chemistry (Advanced Chemical Reaction Design		
	and Discovery)		
講義題目 Subtitle			
責任教員 Instructor	陳 旻究 [JIN Mingoo] (創成研究機構化学反応創成研究拠点)		
担当教員 Other Instructors	Min Gao (ICReDD), HUANG Chung-Yang (ICReDD), SIDOROV Pavel (ICReDD),		
	AKAMA Tomoko (ICReDD), LIST Benjamin (ICReDD)		
科目種別 Course Type			
開講年度 Year	2024	時間割番号 Course Number	094106
期間 Semester	Fall	単位数 Number of Credits	1
授業形態 Type of Class	Lecture	対象年次 Year of Eligible Student	~
対象学科・クラス Eligible Depa	artment/Class		
ナンバリングコード Numbering Code		CHEM_ELMOL 6201	
補足事項 Other Information			
授業実施方式 Class Method		1 対面授業科目《対面のみ》	
+ K Kaya Wanda			

Design of Chemical reaction and molecular assembly with functions, Chemoinformatics, Computational Chemistry

授業の目標 Course Objectives

This course introduces a brand-new research way for investigating molecular chemistry. Especially, advanced examples of the design of new chemical reactions and molecular assembly systems with photo-physical functions will be gently introduced, and the methodology for these research subjects will be described. Also, the advanced level of computational chemistry and chemoinformatics to solve chemical problems will be introduced. Totally four sessions will be delivered to introduce these contents.

1. Advanced course: Introducing Photocontrol to Molecular Systems:

In these lectures, we will describe technologies that allow researchers to control the molecular systems by light and photochemical methods. Specifically, the course will focus on photoredox catalysis and photoswitches.

2. Design of Molecular Dynamics in Crystals and Evaluation Methodology:

In this session, molecular dynamics in crystalline media will be described with recent research. Especially, crystalline molecular rotors system will be focused as well as how to investigate the molecular motion in solid state.

3. Chemoinformatics in advanced topics:

The class covers the advanced topics in the field of chemoinformatics. Chemoinformatics provides useful tools for chemical search, rational design of compounds with desired properties, synthesis prediction, etc. This part is dedicated to topics such as representation of chemical reactions and modeling their properties, as well as current machine learning techniques, focusing on Deep Learning.

4. Advanced Course for Computational Catalysis

The lectures related to advanced computational catalysis are aimed to deeply understand how does the computational chemistry establish catalytic concept and how computational results explain experimental phenomena of catalytic reactions.

到達目標 Course Goals

The main goal of this course is "Knowing the molecular chemistry research fields with experimental and computational methodologies".

Especially, students will know "the photoredox catalysis and photoswitches on molecular system", "the basic ideas to design molecular dynamics in crystal", "Chemical reaction in Chemoinformatics with current machine learning techniques" and "Applications in Computational Catalysis".

授業計画 Course Schedule

The entire course contains four sessions as below;

1. Advanced course: Introducing Photocontrol to Molecular Systems:

- Photoredox Catalysis
- Photoswitches

2. Design of Molecular Dynamics in Crystals and Evaluation Methodology:

- General Introduction of Crystalline Molecular Rotors and Structural Design
- Application and Evaluation for the Molecular Motions in solid state

3. Chemoinformatics in advanced topics:

• Chemical reactions in Chemoinformatics;

Current machine learning techniques.					
Advanced Course for Computational Catalysis:					
 Applications in Computational Catalysis Challenges in Computational Catalysis 					
asic knowledge of chemistry at the undergraduate level might be required. And, the students who got the introduction cou 七学反応創成学入門: CHEM_ELCOM 5271) would be encouraged to have this advanced course to boost their skills.					
沈績評価の基準と方法 Grading System					
/e will give a take-home exam with several open-answer questions for each session, that students have to submit before so					
eadline.					
b学部履修の条件 Other Faculty Requirements					
・キスト•教科書 Textbooks					
i義指定図書 Reading List					
▶照ホームページ Websites					
T究室のホームページ Websites of Laboratory					
ttps://www.icredd.hokudai.ac.jp/all-members/the-huang-lab					
ttps://www.icredd.hokudai.ac.jp/the-jin-group					
ttps://www.icredd.hokudai.ac.jp/the-sidorov-group					
ttps://www.icredd.hokudai.ac.jp/the-gao-group					

科目名 Course Title	分子化学A(分子理論化学)[Molecular Chemistry A (Theoretical Chemistry)]			
講義題目 Subtitle				
責任教員 Instructor	武次 徹也 [TAb	武次 徹也 [TAKETSUGU Tetsuya] (大学院理学研究院)		
担当教員 Other Instructors	HASEGAWA Jur	nya (触媒科学研究所), MAEDA Satoshi	(理学研究院), IIDA Kenji (触媒科学	
	研究所), KOBA	研究所), KOBAYASHI Masato (理学研究院), IWASA Takeshi (理学研究院), Min Gao (ICReDD), MIYAZAKI Ray (触媒科学研究所)		
	(ICReDD), MIYA			
科目種別 Course Type				
開講年度 Year	2024	時間割番号 Course Number	094107	
期間 Semester	Spring/Summer	単位数 Number of Credits	2	
授業形態 Type of Class	Lecture	対象年次 Year of Eligible Student	~	
対象学科・クラス Eligible Dep	oartment/Class			
ナンバリングコード Numberir	ig Code	CHEM_ELMOL 6012		
補足事項 Other Information				
授業実施方式 Class Method		1 対面授業科目《対面のみ》		
キーワード Key Words				
• • • •	ecular orbital, Hart		self-consistent field theory, density	
Schroedinger equation, mol		ree-Fock theory, multiconfigurational s		
Schroedinger equation, mol functional theory, potential	energy surface,	ree-Fock theory, multiconfigurational s geometry optimization, intrinsic reacti		
Schroedinger equation, mole functional theory, potential approximation, Rotational-vil	energy surface, prational state, Reac	ree-Fock theory, multiconfigurational s geometry optimization, intrinsic reacti		
Schroedinger equation, mol functional theory, potential approximation, Rotational-vil 授業の目標 Course Objectiv	energy surface, prational state, Reac 'es	ree-Fock theory, multiconfigurational s geometry optimization, intrinsic reacti	on coordinate, Born-Oppenheime	
Schroedinger equation, mol functional theory, potential approximation, Rotational-vil 授業の目標 Course Objectiv This course aims to provide	energy surface, prational state, Reac res elementary ideas and	ree-Fock theory, multiconfigurational s geometry optimization, intrinsic reacti tion dynamics	on coordinate, Born-Oppenheime the basics of the electronic structure	
Schroedinger equation, mole functional theory, potential approximation, Rotational-vil 授業の目標 Course Objectiv This course aims to provide theory will be addressed. See	energy surface, prational state, Reac res elementary ideas and cond, potential energy	ree-Fock theory, multiconfigurational s geometry optimization, intrinsic reacti tion dynamics d concepts in quantum chemistry. First, t	on coordinate, Born-Oppenheime the basics of the electronic structure ion-path-based dynamics, molecula	
Schroedinger equation, mole functional theory, potential approximation, Rotational-vil 授業の目標 Course Objectiv This course aims to provide theory will be addressed. See vibrational theory, reaction of	energy surface, prational state, Reac res elementary ideas and cond, potential energy lynamics, and theore	ree-Fock theory, multiconfigurational s geometry optimization, intrinsic reacti tion dynamics d concepts in quantum chemistry. First, t gy surface will be explained. Third, react	on coordinate, Born-Oppenheime the basics of the electronic structure ion-path-based dynamics, molecula	
Schroedinger equation, mole functional theory, potential approximation, Rotational-vil 授業の目標 Course Objectiv This course aims to provide theory will be addressed. See vibrational theory, reaction o in modern computational che	energy surface, prational state, Reac res elementary ideas and cond, potential energy lynamics, and theore	ree-Fock theory, multiconfigurational s geometry optimization, intrinsic reacti tion dynamics d concepts in quantum chemistry. First, t gy surface will be explained. Third, react	on coordinate, Born-Oppenheime the basics of the electronic structure ion-path-based dynamics, molecula	
Schroedinger equation, mole functional theory, potential approximation, Rotational-vil 授業の目標 Course Objectiv This course aims to provide theory will be addressed. See vibrational theory, reaction of in modern computational che 到達目標 Course Goals	energy surface, prational state, Reac res elementary ideas and cond, potential energy lynamics, and theore mistry.	ree-Fock theory, multiconfigurational s geometry optimization, intrinsic reacti tion dynamics d concepts in quantum chemistry. First, t gy surface will be explained. Third, react	on coordinate, Born-Oppenheime the basics of the electronic structure ion-path-based dynamics, molecula ill be given to learn the methodology	
Schroedinger equation, mole functional theory, potential approximation, Rotational-vil 授業の目標 Course Objectiv This course aims to provide theory will be addressed. See vibrational theory, reaction of in modern computational che 到達目標 Course Goals Students are expected to un	energy surface, prational state, Reac res elementary ideas and cond, potential energy lynamics, and theore mistry.	ree-Fock theory, multiconfigurational s geometry optimization, intrinsic reacti tion dynamics d concepts in quantum chemistry. First, t gy surface will be explained. Third, react etical approaches to condensed phases wi	on coordinate, Born-Oppenheime the basics of the electronic structure ion-path-based dynamics, molecula ill be given to learn the methodology such as Schroedinger equation, wave	
Schroedinger equation, mole functional theory, potential approximation, Rotational-vil 授業の目標 Course Objectiv This course aims to provide theory will be addressed. See vibrational theory, reaction of in modern computational che 到達目標 Course Goals Students are expected to un function, molecular orbital,	energy surface, prational state, Reac res elementary ideas and cond, potential energy lynamics, and theore mistry. derstand the basic c angular momentum	ree-Fock theory, multiconfigurational s geometry optimization, intrinsic reacti tion dynamics d concepts in quantum chemistry. First, t gy surface will be explained. Third, react etical approaches to condensed phases wi	on coordinate, Born-Oppenheime the basics of the electronic structure ion-path-based dynamics, molecula ill be given to learn the methodolog such as Schroedinger equation, wave ational self-consistent field theory	
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Schroedinger equation, mole functional theory, potential approximation, Rotational-vil 授業の目標 Course Objectiv This course aims to provide theory will be addressed. See vibrational theory, reaction of in modern computational che 到達目標 Course Goals Students are expected to un function, molecular orbital, density functional theory. Stu the mechanism of chemical re	energy surface, prational state, Reac res elementary ideas and cond, potential energy lynamics, and theore mistry. derstand the basic of angular momentum udents are also expe- eactions and reaction	ree-Fock theory, multiconfigurational s geometry optimization, intrinsic reacti tion dynamics d concepts in quantum chemistry. First, t gy surface will be explained. Third, react etical approaches to condensed phases wi concepts in electronic structure theory, s a, Hartree-Fock theory, multi-configura cted to achieve the basic ideas on the po	ion coordinate, Born-Oppenheime the basics of the electronic structure ion-path-based dynamics, molecula ill be given to learn the methodolog such as Schroedinger equation, wave ational self-consistent field theory otential energy surface to understan	
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Schroedinger equation, mole functional theory, potential approximation, Rotational-vil 授業の目標 Course Objectiv This course aims to provide theory will be addressed. See vibrational theory, reaction of in modern computational che 到達目標 Course Goals Students are expected to un function, molecular orbital, density functional theory. Stu the mechanism of chemical re- reaction coordinate, and reac 1. Scientific papers that desc	energy surface, prational state, Reac elementary ideas and cond, potential energy lynamics, and theore mistry. derstand the basic of angular momentum idents are also expe- eactions and reaction stion path dynamics. ribes quantum chemi	ree-Fock theory, multiconfigurational s geometry optimization, intrinsic reacti tion dynamics d concepts in quantum chemistry. First, t gy surface will be explained. Third, react etical approaches to condensed phases wi concepts in electronic structure theory, s t, Hartree-Fock theory, multi-configura cted to achieve the basic ideas on the po- n dynamics, such as potential energy surf As a result, students understand	ion coordinate, Born-Oppenheime the basics of the electronic structur- ion-path-based dynamics, molecula ill be given to learn the methodolog such as Schroedinger equation, wav- ational self-consistent field theory otential energy surface to understan face, geometry optimization, intrinsi and chemical reactions	
Schroedinger equation, mole functional theory, potential approximation, Rotational-vil 授業の目標 Course Objectiv This course aims to provide theory will be addressed. See vibrational theory, reaction of in modern computational che 到達目標 Course Goals Students are expected to un function, molecular orbital, density functional theory. Stu the mechanism of chemical reaction coordinate, and react 1. Scientific papers that desc 2. Knowledges to design, per	energy surface, prational state, Reac elementary ideas and cond, potential energy lynamics, and theore mistry. derstand the basic of angular momentum idents are also expe- eactions and reaction stion path dynamics. ribes quantum chemi	ree-Fock theory, multiconfigurational s geometry optimization, intrinsic reacti tion dynamics d concepts in quantum chemistry. First, t gy surface will be explained. Third, react etical approaches to condensed phases wi concepts in electronic structure theory, s a, Hartree-Fock theory, multi-configura cted to achieve the basic ideas on the po- n dynamics, such as potential energy surf As a result, students understand ical computations of electronic structures	ion coordinate, Born-Oppenheime the basics of the electronic structur- ion-path-based dynamics, molecula ill be given to learn the methodolog such as Schroedinger equation, wav- ational self-consistent field theory otential energy surface to understan face, geometry optimization, intrinsi and chemical reactions	
Schroedinger equation, mole functional theory, potential approximation, Rotational-vil 授業の目標 Course Objectiv This course aims to provide theory will be addressed. See vibrational theory, reaction of in modern computational che 到達目標 Course Goals Students are expected to un function, molecular orbital, density functional theory. Stu the mechanism of chemical reac 1. Scientific papers that desc 2. Knowledges to design, per 授業計画 Course Schedule	energy surface, prational state, Reac provide a state, Reac elementary ideas and cond, potential energy lynamics, and theored mistry. derstand the basic of angular momentum idents are also expe- eactions and reaction stion path dynamics. ribes quantum chemin form, and understand	ree-Fock theory, multiconfigurational s geometry optimization, intrinsic reacti tion dynamics d concepts in quantum chemistry. First, t gy surface will be explained. Third, react etical approaches to condensed phases wi concepts in electronic structure theory, s a, Hartree-Fock theory, multi-configura cted to achieve the basic ideas on the po n dynamics, such as potential energy surf As a result, students understand ical computations of electronic structures d the result of quantum chemical calculati	ion coordinate, Born-Oppenheime the basics of the electronic structur- ion-path-based dynamics, molecula ill be given to learn the methodolog such as Schroedinger equation, wav- ational self-consistent field theory otential energy surface to understan face, geometry optimization, intrinsi and chemical reactions	
Schroedinger equation, mole functional theory, potential approximation, Rotational-vil 授業の目標 Course Objectiv This course aims to provide theory will be addressed. See vibrational theory, reaction of in modern computational che 到達目標 Course Goals Students are expected to un function, molecular orbital, density functional theory. Stu the mechanism of chemical re- reaction coordinate, and reac 1. Scientific papers that desc	energy surface, prational state, Reac prational state, Reac elementary ideas and cond, potential energy lynamics, and theore mistry. derstand the basic of angular momentum idents are also expe- eactions and reaction ction path dynamics. ribes quantum chemis form, and understand	ree-Fock theory, multiconfigurational s geometry optimization, intrinsic reacti tion dynamics d concepts in quantum chemistry. First, t gy surface will be explained. Third, react etical approaches to condensed phases wi concepts in electronic structure theory, s a, Hartree-Fock theory, multi-configura cted to achieve the basic ideas on the po n dynamics, such as potential energy surf As a result, students understand ical computations of electronic structures d the result of quantum chemical calculati	ion coordinate, Born-Oppenheime the basics of the electronic structur ion-path-based dynamics, molecula ill be given to learn the methodolog such as Schroedinger equation, wav ational self-consistent field theory otential energy surface to understan face, geometry optimization, intrinsi and chemical reactions	

- 4. Electron correlations, Density functional theory
- 5. Potential energy surface, Vibrational analysis, Geometry optimization
- 6. Transition state, Intrinsic reaction coordinate
- 7. Born-Oppenheimer approximation
- 8. Theory of molecular vibration and rotation
- 9. Reaction Path dynamics
- 10. Transition state theory

11. Ab initio Molecular dynamics approach

準備学習 (予習・復習)等の内容と分量 Homework

Students are expected to derivate the equations introduced in the class and to solve some exercises.

成績評価の基準と方法 Grading System

The attitude at the lecture (30%) and report scores (70%) are evaluated.

テキスト・教科書 Textbooks

講義指定図書 Reading List

分子理論の展開/永瀬茂、平尾公彦:岩波書店,2002

新版すぐできる量子化学計算ビギナーズマニュアル/平尾公彦(監修)、武次徹也(編集):講談社サイエンティフィク, 2015

参照ホームページ Websites

研究室のホームページ Websites of Laboratory

科目名 Course Title	分子化学A(有機金属化学)[Molecular Chemistry A (Organometallic Chemistry)]				
講義題目 Subtitle					
責任教員 Instructor	澤村 正也 [SAWAMURA Masaya] (大学院理学研究院)				
担当教員 Other Instructors	ITOH Hajime (工学研究院), SHIMIZU Yohei (理学研究院)				
科目種別 Course Type					
開講年度 Year	2024	時間割番号 Course Number	094108		
期間 Semester	Spring/Summer	単位数 Number of Credits	2		
授業形態 Type of Class	Lecture	対象年次 Year of Eligible Student \sim			
対象学科・クラス Eligible Department/Class					
ナンバリングコード Numbering Code		CHEM_ELMOL 6212			
補足事項 Other Information					
授業実施方式 Class Method		2 対面授業科目《一部遠隔》			
キーワード Key Words					

Organometallic Chemistry, Catalysts for Organic Synthesis, Design of Reactions, Mechanisms of Organometallic Reactions, Structures of Organometallic Complexes, Asymmetric Synthesis, Hydrogenation, Cross-coupling

授業の目標 Course Objectives

Spring Term: Objectives of this course is to acquire the basis for designing new metal-catalyzed organic reactions. To this end, students learn that organometallic chemistry is playing important roles to solve problems in synthetic organic chemistry in various aspects and gain a systematic understanding on how organometallic complexes participate in organic reactions.

Summer Term: This course is intended to familiarize the student with advanced concepts in organometallic chemistry. This course mainly focuses on the organic synthetic reactions where the transition-metal-catalyzed process is a key step.

到達目標 Course Goals

Spring Term: The goal of this course is understand various modes of metal?carbon bonds and their reactivities in a systematic way based on molecular orbital considerations.

Summer Term: The goal of this course is to provide graduate students with comprehensive understandings of organometallic chemistry. Students will be familiar with various transition-metal-catalyzed reactions, reaction mechanisms, application of catalysis, basic concept of catalyst design.

授業計画 Course Schedule

Spring Term: The course goes forward along the recommended reading (Hegedus, Chapter 1, Chapter 2, Chapter 9). Summer Term:

1. Synthetic Applications of Transition Metal Hydrides I

2. Synthetic Applications of Transition Metal Hydrides II

3. Synthetic Applications of Complexes Containing Metal–Carbon sigma–Bonds I

- 4. Synthetic Applications of Complexes Containing Metal-Carbon sigma-Bonds II
- 5. Synthetic Applications of Complexes Containing Metal–Carbon sigma–Bonds III
- 6. Synthetic Applications of Transition Metal Carbene Complexes

7. Synthetic Applications of Transition Metal Carbene Complexes II

準備学習 (予習・復習)等の内容と分量 Homework

Students will be expected to have read the assigned materials prior to each class period.

成績評価の基準と方法 Grading System

Attendence rate over 70% is mandatory.

Spring Term: Evaluation is performed based on the score of final exam.

Summer Term: Midterm (30%) and final exam (70%).

他学部履修の条件 Other Faculty Requirements

テキスト・教科書 Textbooks

ヘゲダス遷移金属による有機合成 第3版/L.S. Hegedus 著・村井真二訳:東京化学同人, 2011

講義指定図書 Reading List

参照ホームページ Websites

研究室のホームページ Websites of Laboratory

https://www.chem.sci.hokudai.ac.jp/~orgmet/index.php?id=25 https://itogrouphp.eng.hokudai.ac.jp/en.html https://www.icredd.hokudai.ac.jp

備考 Additional Information

It is advisable to take all lectures and experiments on organic chemistry in the undergraduate study. Moodle in ELMS will be used in case of distance learning.

科目名 Course Title	応用分子化学(化学エネルギー変換)[Applied Molecula	ar Chemistry (Chemical Energy	
	Conversion)]			
講義題目 Subtitle				
責任教員 Instructor	坪内 直人 [TSUBOUCHI Naoto] (大学院工学研究院)			
担当教員 Other Instructors	〒1 电八[130D000111 Na010](八子阮上子刎九阮)			
科目種別 Course Type				
	0004		004100	
開講年度 Year	2024	時間割番号 Course Number	094109	
期間 Semester	Winter	単位数 Number of Credits	1	
授業形態 Type of Class	Lecture	対象年次 Year of Eligible Student	~	
対象学科・クラス Eligible Depa				
ナンバリングコード Numbering	Code	CHEM_ELMOL 6102		
補足事項 Other Information				
授業実施方式 Class Method		1 対面授業科目《対面のみ》		
キーワード Key Words				
_	lance, Chemical Eq	uilibrium, Reaction Rate, Combustion, Stea	am Reforming, Energy Efficiency,	
Cold Gas Efficiency, Heat Loss				
授業の目標 Course Objectives				
-		s on oil, coal and natural gas, and this depe	ndency will be almost unchanged	
		cent IEA (International Energy Agency) v		
		ssil fuels is the best way to reduce CO2 en		
		asic theories about chemical energy convers		
		ormer for methane steam reforming.		
到達目標 Course Goals				
	s of chemical rea	ction engineering, such as material balar	nce, enthalpy balance, chemical	
equilibrium and reaction rate.			,	
-	ning in a fixed hed a	reformer at adiabatic conditions.		
Endenate methane steam reform	inno in a incu beu i	control at adiabatic conditions.		
All students are also required t	o present and discu	es their own research subjects from a view	of reactor designing	
All students are also required t 授業計画 Course Schedule	o present and discl	iss their own research subjects from a view of	or reactor designing.	
	actor theory M-+	ial balance calculation method		
1. Fundamentals of chemical re	-			
2. Fundamentals of chemical re	-			
	-	ical equilibrium calculation method		
4. Fundamentals of chemical re	-		ofmothene	
		d reformer: Steam reforming and combustion		
6. Simulator development: Homogeneous gas phase reaction, gas-solid reaction, gas-solid catalytic reaction				
準備学習(予習・復習)等の内容				
Students are expected to read relevant contents in the text beforehand. After class, students are also requested to understand				
the lecture by reading additionally the related bibliography and solving problems provided there.				
成績評価の基準と方法 Grading System				
Grades are awarded based on regular assignments, presentation and discussion in the class.				
他学部履修の条件 Other Faculty Requirements				
テキスト・教科書 Textbooks				
教科書は特に指定せず, Lecture 時にプリントを配布する。				
Handout made by the instructo	r will be delivered.			
講義指定図書 Reading List				
参照ホームページ Websites				
研究室のホームページ Websites of Laboratory				
https://chemeng-hokudai.jp/en/				
備考 Additional Information				
Students are required to understand the basic knowledge of related Chemical Engineering Stoichiometry, Thermodynamics and				
Reaction Kinetics in advance.				

科目名 Course Title	応用分子化学(分離プロセス工学 I)[Applied Molecular	Chemistry (Separation Process	
	Engineering I)]		v · x	
講義題目 Subtitle				
責任教員 Instructor				
担当教員 Other Instructors				
科目種別 Course Type				
開講年度 Year	2024	時間割番号 Course Number	094110	
期間 Semester	Intensive	単位数 Number of Credits	1	
授業形態 Type of Class	Lecture 対象年次 Year of Eligible Student \sim			
対象学科・クラス Eligible Depa	rtment/Class			
ナンバリングコード Numbering	Code	CHEM_ELMOL 6101		
補足事項 Other Information				
授業実施方式 Class Method		1 対面授業科目《対面のみ》		
キーワード Key Words				
Porous Materials, Adsorption				
授業の目標 Course Objective	S			
	the basic principles	of separation processes with a particular	focus on processes using porous	
materials such as adsorption.				
到達目標 Course Goals				
By the end of this course, a su				
1. Understand the mechanisms	-			
2. Understand methods to obta isotherm	in adsorption isothe	erms, and become able to describe the chara	acteristics of the material from its	
	ion theories and ad	sorption equations, and become able to an	alvza adsorption isotherms using	
them	ion theories and ad	solption equations, and become able to an	alyze ausorption isotherms using	
授業計画 Course Schedule				
This course will be held as an i	n-person class at Sa	apporo Campus.		
1. Overview of Adsorption Phe	nomena and Adsorb	ents		
2. Adsorption Phenomena				
3. Typical Adsorbents and The	ir Production Proce	sses		
4. Adsorption Mechanisms				
5. Adsorption Isotherms				
		(Henry Equation, Freundlich Equation, Lan	gmuir Equation)	
7. Adsorption Theories and Ad	sorption Equations	(BET Equation)		
8. Examination	<u> </u>			
準備学習 (予習・復習)等の内容と分量 Homework				
Students are encouraged to read relevant materials ahead of time and review what they have been taught, especially the				
contents of quizzes after classes to deepen their understanding.				
成績評価の基準と方法 Grading System The attendance rate must be over 70% to be qualified to take the final project. Evaluations will be made based on (1) learning				
attitude (20%), (2) quiz scores (20%) and final examination scores (60%). Quizzes will be used to evaluate the level of				
understanding of each class and examinations will be used to evaluate the achievement level of this course.				
他学部履修の条件 Other Faculty Requirements				
テキスト・教科書 Textbooks				
講義指定図書 Reading List				
-				
参照ホームページ Websites				
This course will be provided as part of the Hokkaido Summer Institute., For more information (invited lecturers, course				
details, etc.), please visit the website below:,				
https://hokkaidosummerinstitute.oia.hokudai.ac.jp/en/courses/CourseDetail=G059				
研究室のホームページ Websit	es of Laboratory			
備考 Additional Information				

Prerequisite courses include undergraduate-level mathematics, transport phenomena, thermodynamics, statistical thermodynamics, and separation process

科目名 Course Title	応用分子 ル学()	分離プロセス工学Ⅱ)[Applied Molec	ular Chemistry (Separation Process		
	Engineering II)]	Mum	Care Chemistry (Deparation 1100855		
講義題目 Subtitle	Sugmeeting II/J				
責任教員 Instructor					
担当教員 Other Instructors					
科目種別 Course Type	Ron C. Runnebaum (University of California, Davis)				
料白裡別 Course Type 開講年度 Year	2024	時間割番号 Course Number	004111		
	2024		094111		
期間 Semester	Intensive	単位数 Number of Credits 対象年次 Year of Eligible Student	1		
授業形態 Type of Class	Lecture	刘家平次 fear of Eligible Student	\sim		
	対象学科・クラス Eligible Department/Class				
· · · · ·	ンバリングコード Numbering Code CHEM_ELMOL 6101				
	足事項 Other Information				
授業実施方式 Class Method		1 対面授業科目《対面のみ》			
キーワード Key Words					
Porous Materials, Adsorption,		on, Chromatography			
授業の目標 Course Objective	s				
_		rocesses with a particular focus on pro-	cesses using porous materials such as		
adsorption and membrane sepa	ration.				
到達目標 Course Goals					
1. Understand the roles of sepa	•	-			
		esses in terms of rate and equilibrium			
	-	ransport phenomena relevant to the de	· · ·		
4. Understand the fundamental	principles of indust	rial adsorption and membrane separation	on processes and perform basic design		
of these processes					
	devices and product	s equipped with adsorption and membra	ane-separation functions		
授業計画 Course Schedule					
1. Roles of industrial separation	-				
2. Thermodynamics and transp	ort phenomena rele	vant to separation processes			
3. Adsorption process					
4. Case study 1					
5. Case study 2					
6. Membrane separation proces	SS				
7. Case study 3					
8. Project					
準備学習 (予習・復習)等の内	容と分量 Homework	ζ.			
Students are encouraged to re	ad the textbook and	l relevant materials ahead of time. Stud	dents are required to submit assigned		
homework.					
成績評価の基準と方法 Gradir	ng System				
		nigher to be eligible for the final project			
three factors: learning attitud	e (20%), which incl	udes engagement and participation, as	ssignment scores (30%), which assess		
understanding of class mater	ial and separation	principles, and the final project sco	ore (50%), which evaluates practical		
application of skills learned.					
テキスト・教科書 Textbooks					
1. Separation Process Principles: With Applications Using Process Simulators, 4th Edition / J. D. Seader, Ernest J. Henley, D.					
Keith Roper: John Wiley & Sons, Inc., 2016					
2. Product and Process Design Principles: Synthesis, Analysis and Evaluation, 4th Edition/Warren D. Seider, Daniel R.					
Lewin, J. D. Seader, Soemantri Widagdo, Rafiqul Gani, Ka Ming Ng:Wiley, 2016					
講義指定図書 Reading List					
現代化学工学/橋本健治、荻	野文丸 編:産業図	書,2001			
参照ホームページ Websites					
This course will be provided as part of the Hokkaido Summer Institute., For more information (invited lecturers, course					
details, etc.), please visit the website below:,					
https://hokkaidosummerinstitute.oia.hokudai.ac.jp/en/courses/CourseDetail=G060					
備考 Additional Information					
Prerequisite courses include undergraduate-level mathematics, transport phenomena, thermodynamics, statistical					
thermodynamics, and separation process					
It is desirable for students to b	e able to understan	d numerical methods to solve differentia	al equations.		

科目名 Course Title	応用分子化学A	A(触媒設計)[Applied Molecular Chemis	try A (Catalyst Design)]		
溝義題目 Subtitle					
責任教員 Instructor	清水 研一 [SHIMIZU Kenichi] (触媒科学研究所)				
旦当教員 Other Instructors	TOYAO Takashi (触媒科学研究所)				
科目種別 Course Type					
鼎講年度 Year	2024	時間割番号 Course Number	094112		
閉間 Semester	Fall/Winter	単位数 Number of Credits	2		
受業形態 Type of Class	Lecture	対象年次 Year of Eligible Student	~		
オ象学科・クラス Eligible De∣	partment/Class				
⁻ンバリングコード Numberir	ıg Code	CHEM_ELMOL 6112			
甫足事項 Other Information					
受業実施方式 Class Method	1	3 遠隔授業科目《一部対面》			
テーワード Key Words					
atalysis, surface chemistry, と業の目標 Course Objectiv		alysis, kinetics, industrial chemistry			
• • • • •		sm and design concept of heterogeneous	catalysts, students should understar		
		. The goal of this lecture is to un			
hermodynamics in terms of	catalysis and use th	nese basic knowledge for catalyst design	and catalysis research. In addition, w		
iscuss design concept and	practical role of he	eterogeneous catalysis in current cataly	rtic processes for automotive emission		
ontrol and organic synthesis	3.				
削達目標 Course Goals					
		tics and thermodynamics in terms of cat			
		entation on it. We will also learn impo			
		control. In the presentation, students e			
	nologies. Presentat	ion techniques of students will be improv	ved.		
受業計画 Course Schedule					
. Geometry of solid surface	• •				
. Evaluation of catalytic act					
. Characterization of cataly					
. Characterization of cataly	st II				
. Design of solid catalyst					
. Catalyst preparation					
Computational chemistry	or catalysis				
. Intermediate exam					
. Environmental catalysis					
0. Catalysis for fossil fuel c					
1. Catalysis for industrial p		als			
2. Catalysis for green chem	istry				
3. Presentation					
4. Presentation					
隼備学習(予習・復習)等のに			S prior to each leature. Attended my		
準備学習 (予習・復習)等の For the former-half lectures	, PDF files to be us	sed in the lectures are uploaded in ELM			
準備学習 (予習・復習)等の For the former-half lectures rint and bring it for each le	, PDF files to be us ecture. Students sh	sed in the lectures are uploaded in ELM nould understand basic physical chemist	ry, reading textbooks. Using scienti		
準備学習 (予習・復習)等の For the former-half lectures rint and bring it for each l lectronic calculator, studen	, PDF files to be us ecture. Students sh ts' laptop, they solv	sed in the lectures are uploaded in ELM	ry, reading textbooks. Using scienti		
準備学習 (予習・復習)等のF For the former-half lectures wint and bring it for each lectoric calculator, studen 改績評価の基準と方法 Grad	, PDF files to be us ecture. Students sh ts' laptop, they solv ding System	sed in the lectures are uploaded in ELM nould understand basic physical chemist we kinetic problems, draw solid surface as	ry, reading textbooks. Using scientine nd create a presentation file.		
準備学習 (予習・復習)等のF For the former-half lectures, wint and bring it for each lectronic calculator, studen dectronic calculator, studen 成績評価の基準と方法 Grad ntermediate exam (50%), fina	, PDF files to be us ecture. Students sh ts' laptop, they solv ding System	sed in the lectures are uploaded in ELM nould understand basic physical chemist	ry, reading textbooks. Using scientine nd create a presentation file.		
print and bring it for each le electronic calculator, studen 或績評価の基準と方法 Grad ntermediate exam (50%), fina テキスト・教科書 Textbooks	, PDF files to be us ecture. Students sh ts' laptop, they solv ding System	sed in the lectures are uploaded in ELM nould understand basic physical chemist we kinetic problems, draw solid surface as	ry, reading textbooks. Using scienti nd create a presentation file.		
準備学習(予習・復習)等のF For the former-half lectures print and bring it for each le electronic calculator, studen 或績評価の基準と方法 Grad ntermediate exam (50%), fina テキスト・教科書 Textbooks 講義指定図書 Reading List	, PDF files to be us ecture. Students sh ts' laptop, they solv ding System	sed in the lectures are uploaded in ELM nould understand basic physical chemist we kinetic problems, draw solid surface as	ry, reading textbooks. Using scientine nd create a presentation file.		
準備学習 (予習・復習)等のF For the former-half lectures orint and bring it for each le electronic calculator, studen 式績評価の基準と方法 Grad ntermediate exam (50%), fina テキスト・教科書 Textbooks	, PDF files to be us ecture. Students sh ts' laptop, they solv ding System	sed in the lectures are uploaded in ELM nould understand basic physical chemist we kinetic problems, draw solid surface as	ry, reading textbooks. Using scienti nd create a presentation file.		

科目名 Course Title	物質化学(固体物性化学)[Materials Chemistry (Organic Solid State Chemistry)]				
講義題目 Subtitle					
責任教員 Instructor	原田 潤 [HARADA Jun] (大学院理学研究院)				
担当教員 Other Instructors					
科目種別 Course Type			Y		
開講年度 Year	2024	時間割番号 Course Number	094201		
期間 Semester	Spring	単位数 Number of Credits	1		
授業形態 Type of Class	Lecture	対象年次 Year of Eligible Student	~		
対象学科・クラス Eligible Depa					
ナンバリングコード Numbering	Code	CHEM_ELMAT 6000			
補足事項 Other Information					
授業実施方式 Class Method		1 対面授業科目《対面のみ》			
キーワード Key Words					
molecular materials, crystal str	uctures, symmetry,	intermolecular interactions, charge-transfe	er interactions, hydrogen bonding,		
band structures, electrical cone		e reactions, molecular motions			
授業の目標 Course Objectives	6				
		materials, which are aggregates of molecul			
		interactions and the structures/functions,			
		lectronic structures of molecular materia			
	notions and reaction	s in crystals can be understood in terms of	crystal structures.		
到達目標 Course Goals		1 11 .			
After successful completion of					
1. Understand the principles by		-			
	p between structur	al features of molecular crystals and the	ir physical properties, molecular		
motions, and reactivities.		Gran mala sular dation to amontal dation			
3. Acquire basic idea of functional material design: from molecular design to crystal design.					
授業計画 Course Schedule The following topics will be lectured in order:					
1. Molecular structures and symmetry of crystals					
The relationship between the shapes of molecules and the structure/symmetry of their crystals will be discussed.					
2. Intermolecular interactions and molecular arrangements in crystals					
The influence of charge-transfer interaction and hydrogen bonding on crystal structures will be discussed. Guidelines for					
controlling the molecular arran					
3. Electronic structures of mole					
	The electronic structures (band structures) of radical crystals will be discussed. Neutral-to-ionic transitions and formal				
		complex crystals will be explained.			
4. Chemical reactions and mole	ecular motions in cry	rstals			
Chemical reactions and molecu	lar motions in cryst	als will be illustrated and explained in terms	s of the crystal structures.		
準備学習 (予習・復習)等の内容	容と分量 Homework				
You are expected to have basic knowledge of physical chemistry and need to review it beforehand. The lecture materials are					
available in Moodle in advance. Reports will be assigned.					
成績評価の基準と方法 Grading System					
Unless there are special circumstances, more than 70% class attendance is required for the grade evaluation. The grade will be					
evaluated based on the reports assigned during the course.					
他学部履修の条件 Other Faculty Requirements					
テキスト・教科書 Textbooks					
講義指定図書 Reading List					
参照ホームページ Websites					
研究室のホームページ Websites of Laboratory					

科目名 Course Title	物質化学(ナノデバイス材料特論)[Materials Chemistry (Materials for Nanodevice)]			
講義題目 Subtitle				
責任教員 Instructor	松尾 保孝 [MATSUO Yasutaka] (電子科学研究所)			
担当教員 Other Instructors	NAGASHIMA Ka	NAGASHIMA Kazuki (電子科学研究所), YOMOGIDA Yohei (電子科学研究所)		
科目種別 Course Type				
開講年度 Year	2024	時間割番号 Course Number	094202	
期間 Semester	Summer	単位数 Number of Credits	1	
授業形態 Type of Class	Lecture	対象年次 Year of Eligible Student	~	
対象学科・クラス Eligible Depa	artment/Class			
ナンバリングコード Numbering Code		CHEM_ELMAT 6000		
補足事項 Other Information				
授業実施方式 Class Method 2 対面		2 対面授業科目《一部遠隔》		

Photonics materials, Photonic devices, Subwavelength Optics, Plasmonics, Electronic materials, Electron transport, Charge storage, IoT, energy conversion, sensors.

授業の目標 Course Objectives

The purpose of this lecture is to understand the relationships between functions and structures of several materials and devices. Especially, this lecture focuses on the fabrication and analysis of photonics materials, electronic materials and you will learn photonics and electronics properties and their applications.

到達目標 Course Goals

1. Understanding on he fundamentals of nanophotonics and nanoelectronic devices from the electronic state, fabrication method, property and evaluation of their functions

2. Acquiring basic knowledge of plasmonics and photonics materials such as interaction between photons and electrons, light wave propagation, diffraction, reflection and polarisation control, in order to understand the correlation between structure and function of photonics materials based on oxides and metal-based plasmonics materials

3. Understanding the electron transport properties and interface properties of silicon-based semiconductors and the charge storage properties of dielectrics, mainly oxides, and learning design guidelines from advanced electronic materials and devices

授業計画 Course Schedule

This lecture will review the photonic and electronic materials, the device applications, and the engineering innovations in the advanced information society.

(1) Photonics materials

(2) Fundamentals on refraction, diffraction and interference,

- (3) Phtonic devices using diffraction and phase of light and materials supporting them
- (4) Plasmonics and its application for analysis
- (5) Fundamentals of electronics materials

(6) Fundamentals and evaluation methods for electron transport properties, physical properties of interface and charge storage properties.

(7) Advanced electronics materials and devices

(8) New trends in electronics for IoT, environmental and medical applications

準備学習 (予習・復習)等の内容と分量 Homework

The outline can be understood from the deliverd documents in each lectures.

The report works will be given at end of each section.

成績評価の基準と方法 Grading System

As a general rule, attendance of 70% or more of the lectures is a requirement for the evaluation.

The evaluation is based on the total score of the mini-examination for each lecture.

他学部履修の条件 Other Faculty Requirements

テキスト・教科書 Textbooks

講義指定図書 Reading List

半導体デバイス―基礎理論とプロセス技術/S.M. Sze:産業図書 固体の電子構造と化学/P.A. Cox:技報堂出版 物質構造と誘電体入門(物性科学入門シリーズ)/高重正明:裳華房 はじめての光学(KS物理専門書)/川田善正:講談社

参照ホームページ Websites

https://www.es.hokudai.ac.jp

研究室のホームページ Websites of Laboratory

https://sites.google.com/view/nagashima-lab/ https://nanostructure.es.hokudai.ac.jp/

備考 Additional Information

When you want to study the fundamental of optics including lens, grating, hologram, or you use optical microscope or laser, feel free to attend this lecture.

科目名 Course Title	物質化字(材料化	公学)[Materials Chemistry (Introduction	to Material Science)	
講義題目 Subtitle	· · · · · · -			
責任教員 Instructor	高橋 啓介 [TAKAHASHI Keisuke] (大学院理学研究院)			
担当教員 Other Instructors	Lauren TAKAHASHI (理学研究院)			
科目種別 Course Type				
開講年度 Year	2024	時間割番号 Course Number	094203	
期間 Semester	Fall	単位数 Number of Credits	1	
授業形態 Type of Class	Lecture	対象年次 Year of Eligible Student	~	
対象学科・クラス Eligible Depar				
ナンバリングコード Numbering Code		CHEM_ELMAT 6002		
補足事項 Other Information				
授業実施方式 Class Method		4 遠隔授業科目《遠隔のみ》		
キーワード Key Words				
Data science, machine learning	, materials informat	tics, statistics, visualization		
授業の目標 Course Objectives	5			
This course introduces the fund		-		
	-	nformatics, environment construction fo		
	and catalyst scie	ence data, scientific data visualization	n and analysis, and supervised and	
unsupervised learning,				
		o the design and knowledge extraction	of materials and catalysts from data,	
with a focus on supervised and	-			
		ython language, explaining data science	e and technology that can be started	
from zero without any program	ming experience.			
到達目標 Course Goals				
		aster basic data science techniques in n	naterials informatics and to be able to	
design materials and catalysts a	and extract knowled	lge from the data.		
授業計画 Course Schedule	1 1 0 1			
Lecture 1: Overview of Materia				
Lecture 2 Data and Data Prepr	ocessing			
Lecture 3 Data Visualization				
Lecture 4 Machine Learning Ba				
Lecture 5 Machine Learning 1 Lecture 6 Machine Learning 2:	-			
Lecture 7 Machine Learning 3	•			
Lecture 8 Examination and Rep	-			
準備学習(予習・復習)等の内容		4		
		▶ he content explained in class, students	are encouraged to review the material	
after class and before the exam		ne content explained in class, students	are checkinged to review the material	
成績評価の基準と方法 Gradin				
Grading will be based on report				
他学部履修の条件 Other Facu				
テキスト・教科書 Textbooks				
An Introduction: Materials Info	rmatics and Cataly	st Informatics∕Keisuke Takahashi:Spri	nger, 2024	
テキスト、参考書使用しない。				
No text book in the class.				
講義指定図書 Reading List				
参照ホームページ Websites				
		ata.org/, https://scikit-learn.org/stable	e/	
研究室のホームページ Websit	es of Laboratory			
備考 Additional Information				

科目名 Course Title	物質化学(現代	化学反応理論)[Materials Chemistry(Advanced Chemical Reaction Rate	
	Theory)]			
講義題目 Subtitle				
責任教員 Instructor	小松﨑 民樹 [KOMATSUZAKI Tamiki] (電子科学研究所)			
担当教員 Other Instructors				
科目種別 Course Type				
開講年度 Year	2024	時間割番号 Course Number	094204	
期間 Semester	Winter	単位数 Number of Credits	1	
授業形態 Type of Class	Lecture	対象年次 Year of Eligible Student	\sim	
対象学科・クラス Eligible Depa	rtment/Class			
ナンバリングコード Numbering Code		CHEM_ELMAT 6002		
補足事項 Other Information				
授業実施方式 Class Method		1対面授業科目《対面のみ》		
キーワード Key Words				

chemical reactions, nonequilibrium, collective motion, dynamical systems theory, machine learning, AI

授業の目標 Course Objectives

Chemical reactions inevitable for maintaining living systems correspond to the change of rearrangement of atoms constituting molecules. Even though the corresponding scale differs from that of the molecular level at the order of 10²⁰, the motion of the planets in our universe is also regarded as that of molecules at the same footing. However, because the motions of particles are interacting with each other in complicated fashions nonlinearly, the prediction of the future is apparently almost impossible due to arbitrary small uncertainty at the initial condition. People will understand that mathematical science enables us to provide a special route in the phase space along which one can predict the fate of reactions, and actually is utilized for controlling reactions and designing a route of a spacecraft to travel different plants with the minimum cost.

到達目標 Course Goals

We will understand the motion of particles from the viewpoint of the geometry of the phase space composed of the coordinates and the conjugate momenta of particles. We will understand the history of the development of chemical reaction theories from the viewpoint of not chemistry but Hamiltonian systems, and learn a set of problems forgotten in the history of chemistry. Then, we learn the so-called normal form and that even under the existence of chaos there exists a deterministic regularized route in the phase space. We will learn the question of whether such deterministic regularized route exists or not will shed light on the question of why reactions occur, i.e., chance and necessity of the changes, which has been asked from the day of alchemy. Furthermore, we will learn the applications of quantum computing and machine learning to chemical reaction design and discovery.

授業計画 Course Schedule

The lecture will be organized for students who have not learned chemical reactions theory and Hamiltonian dynamical systems more than Newton's law.

We will take an ample of time to accept questions from students and ask students to write a short report on which you must write what you learned at each lecture and what the most difficult to follow were.

- 0: An overview of the history of chemical reactions: from dynamical system viewpoint.
- 1: Universal chemical reaction theories based on high-dimensional phase space geometry

2: Breakdown of normally hyperbolic invariant manifolds: alternation of degree of freedom between reactive and nonreactive degrees of freedom

- 3: Reinforcement learning on chemical reaction design and discovery
- 4: Quantum Computing on chemical reaction design and discovery

準備学習 (予習・復習)等の内容と分量 Homework

I make a timeslot of Q&A, and ask a report to write any questions he/she feel during each lecture.

成績評価の基準と方法 Grading System

Grading is mainly based on the report on the exercises that were given in the class, and based on activity (how much he/she made questions as a report) in a class as well.

他学部履修の条件 Other Faculty Requirements

テキスト・教科書 Textbooks

I do not supply any books, but hope that all students learn how the interdisciplinary research between chemistry and mathematics is potentially deeper than the design of a spacecraft pathway, and that students actively imagine and dig what type of new research may exist in between chemical reactions and the other research arena.

講義指定図書 Reading List

参照ホームページ Websites

http://mlns.es.hokudai.ac.jp/

研究室のホームページ Websites of Laboratory

https://mlns.es.hokudai.ac.jp/english.html

科目名 Course Title	物質化学A(ナノ物質化学)[Materials Chemistry A (Mesoscopic Material Chemistry)]			
講義題目 Subtitle				
責任教員 Instructor	佐田 和己 [SADA Kazuki] (大学院理学研究院)			
担当教員 Other Instructors	MATSUOKA Keit	MATSUOKA Keitaro (理学研究院), TSUTSUMI Takuro (理学研究院)		
科目種別 Course Type				
開講年度 Year	2024	時間割番号 Course Number	094205	
期間 Semester	Spring/Summer	単位数 Number of Credits	2	
授業形態 Type of Class	Lecture	対象年次 Year of Eligible Student	~	
対象学科・クラス Eligible Depa	artment/Class			
ナンバリングコード Numbering Code		CHEM_ELMAT 6012		
補足事項 Other Information				
授業実施方式 Class Method 2 対面授業科目《一部遠隔》				

Polymer Chemistry, Self-organization, Molecular Networks, Molecular Assembly, Supermolecular Chemistry, Gel, Nanoporous Materials, Crystals, Radiation Chemistry, Computational Chemistry

授業の目標 Course Objectives

Based on the understanding of the essence of materials, this lecture will provide specialized knowledge for designing and developing the functions of inorganic, metallic, organic, and biomaterials and composite materials. In particular, students will learn the fundamentals to consider the physical properties design and application guidelines for materials ranging from hard materials such as crystals to soft artificial materials such as gels and biomacromolecules such as proteins and nucleic acids. In particular, three topics, molecular network materials, astatine-based radiotherapy, and chemical reaction analyses based on computational chemistry, will be reviewed and their applications to nanotechnology and other fields will be introduced based on cutting-edge research.

In addition, we will discuss what research is, reflect on our own research, and discuss how to solve problems or deepen our own research through PBL or presentation-style exercises that transcend the boundaries of engineering and science.

(I) Material Design

Preparation, structure, and function of materials with network structures such as supramolecular chemistry, gels, crystals, and MOFs will be reviewed and their applications will be introduced.

(II) Astatine-based radiotherapy

Organic chemistry using astatine, an α -ray emitting nuclide emitted by accelerators for radiotherapy, and its applications will be introduced.

(III) Reaction analysis theory based on computational chemistry Several methodologies for analyzing chemical reaction mechanisms based on potential energy surfaces and actual applied research will be reviewed.

(IV) Deepening research through PBL (Problem-Based Learning)

Students will discuss how to solve problems or deepen their own research by using their own research as a subject matter.

到達目標 Course Goals

Firstly this course reviews fundamentals of molecular network structures and bio-molecular machines with respect to selforganization.

Students will be able to acquire basic knowledge both on preparation and molecular design of network structures and on biomolecular machines, understand their construction and working principle in advanced applications of physical chemistry and material science. Students will be able to discuss problem solving or deepening their own research using their own research as a subject. Students will be able to know organic chemistry of alpha-ray emitting astatine prepared by using an accelerator for radiotherapy.Students will be able to understand the advantages of various reaction analysis theories based on potential energy surfaces.

授業計画 Course Schedule

(Topic I) Material Design provided by K. S.

(Topic II) Astatine-based radiotherapy by K. M.

(Topic III) Reaction analysis theory by T. T.

(Topic IV) Problem Based Learning (PBL) for deepening of own research by K. S., K. M. and T. T. Based on your own research as the subject, discuss how to solve problems or deepen your own research.

準備学習 (予習・復習)等の内容と分量 Homework

Students will read reviews and the primary literature on each topic, and submit questions for instrutor after every classes and some written reports on the topics.

成績評価の基準と方法 Grading System

Attendance more than 11 times among 15 classes is essential for evaluation of the credit.

The grade is evaluated in the following three items;(1) learning attitude (15%), (2) report/homework or presentation (70%), (3) term paper (15%). Understanding for each class is evaluated by report/homework/presentation, and the basic knowledge for whole subjects is confirmed by term examination.

他学部履修の条件 Other Faculty Requirements

テキスト・教科書 Textbooks

講義指定図書 Reading List

「科学的思考」のレッスン:学校で教えてくれないサイエンス/戸田山和久:NHK 出版, 2011

参照ホームページ Websites

研究室のホームページ Websites of Laboratory

https://www.chem.sci.hokudai.ac.jp/~matchemS/english/index.html

科目名 Course Title	応用物質化学(有	機物性化学)[Applied Materials Chemis	try (Physical Chemistry of Organic		
	Materials)]				
講義題目 Subtitle					
責任教員 Instructor					
担当教員 Other Instructors					
科目種別 Course Type					
開講年度 Year	2024	時間割番号 Course Number	094206		
期間 Semester	Fall	単位数 Number of Credits	1		
授業形態 Type of Class	Lecture	対象年次 Year of Eligible Student	\sim		
対象学科・クラス Eligible Depa					
ナンバリングコード Numbering		CHEM_ELMAT 6100			
補足事項 Other Information	0000				
授業実施方式 Class Method		1 対面授業科目《対面のみ》			
技味実施方式 Glass Method キーワード Key Words		1 对面设来杆白《对面》207//			
	had Salitan Andar	non localization Domadation Machanian			
		rson localization, Degradation Mechanism	1		
授業の目標 Course Objectives		important mathed for understanding the	atmicture and driving machanisms of		
		important method for understanding the			
		ar cells and organic electroluminescent			
-		properties of several organic molecules	s, their aggregates, ongomers, and		
	neoretical approact	nes such as quantum chemistry.			
到達目標 Course Goals	hauld he able to				
By the end of the lecture, you		anghin hatwaan malagulan functions and	nhusical properties and		
-		onship between molecular functions and			
- Develop the ability to discove	er problems in actua	al materials chemistry and to solve them	using a theoretical approach.		
 (1) Charge-transfer complex (P (2) Thiophene system (degradat (3) Silane system (sigma-Hucket (4) Graphene-based system (pc (5) Polyacetylenes (solitons) (6) Spectroscopy (spectroscopie) (7) Current topics 	i-stacking) cion mechanism) el, Anderson localiza lycyclic aromatic co c approach)	ompounds)	upproach. (in no particular order)		
準備学習 (予習・復習)等の内容					
		ntum chemistry from the undergraduate o	course.		
成績評価の基準と方法 Gradin					
	or more of the cla	ass sessions is a requirement for grading	. The attitude at the lecture (20%)		
and report (30%) are evaluated.					
他学部履修の条件 Other Facu	ily requirements				
テキスト・教科書 Textbooks Lecture 用資料は、適宜配布す 講義指定図書 Reading List 有機半導体のデバイス物性(K 有機エレクトロニクス入門/筒ታ 参照ホームページ Websites	S 物理専門書)/安				
参照小一ムハーン Websites					
研究室のホームページ Websites of Laboratory					
備考 Additional Information					

科目名 Course Title	応用物質化学(界	面電子化学)[Applied Materials Chemistr	y (Interfacial Electrochemistry)]
講義題目 Subtitle			
責任教員 Instructor	伏見 公志[FUSH	HIMI Koji] (大学院工学研究院)	
担当教員 Other Instructors			
科目種別 Course Type			
開講年度 Year	2024	時間割番号 Course Number	094207
期間 Semester	Summer	単位数 Number of Credits	1
授業形態 Type of Class	Lecture	対象年次 Year of Eligible Student	~
対象学科・クラス Eligible Depa	rtment/Class		
ナンバリングコード Numbering	Code	CHEM_ELMAT 6102	
補足事項 Other Information			
授業実施方式 Class Method	授業実施方式 Class Method 1 対面授業科目《対面のみ》		
キーワード Key Words			
Electrode structure, interfacia	l reaction, charge t	transfer process, mass transport process	, electrochemical methods, micro-
electrochemistry	electrochemistry		
授業の目標 Course Objectives			
		urring at interfaces between electrolyte a	
learn electrode reactions from	views of interfacial	thermodynamics, charge transfer kinetics,	and mass transport process at the
interface, and then proceed to	principle and appl	ication using electrochemical methods as	well as physical chemistry at the
interface. You are finally required to present and discuss electrochemical or interfacial subjects as well as your own research			
subjects.			
到達目標 Course Goals			
By the end of this course, a su	ccessful learner will		
1 . be able to discuss basic as	pects of electrocher	nistry, mainly for electrode structure incl	ading atomic level surface, electric
double layer, electrode potenti	al, etc.		

2. be able to fullfil to interfacial reaction such as charge transfer process and mass transfer process.

3. be able to understand details of electrochemical methods both to evaluate and to apply electrochemical reaction.

授業計画 Course Schedule

1-3. Fundamentals of electrochemistry; electrode structure, electrode potential, non-Faradaic and Faradaic processes, energy conversion, electrolyte

4. Outline of electrochemical methods; apparatus, electrochemical cell, and electric circuit used in electrochemistry

5-6. Polarization technique; controlling processes of interfacial reaction (charge transfer process and mass transfer process), cyclic voltammetry, hydrodynamic method, microelectrode technique

7. Transient technique; potentiometry, ammerometry, coulometry, AC impedance spectroscopy, electrochemical sensor

8. Presentation; electrochemical theory and methods in newest research topics are introduced and discussed.

準備学習 (予習・復習)等の内容と分量 Homework

You are requested to read relevant contents in the documents beforehand. You are also expected to study journal articles in interfacial electrochemistry and prepare presentation materials to be used in class discussions.

You are requested to submit a report about class.

成績評価の基準と方法 Grading System

Students will be evaluated by presentations (50%) and reports (50%).

他学部履修の条件 Other Faculty Requirements

テキスト・教科書 Textbooks

Electrode Dynamics/A.C. Fisher: Oxford University Press, 1996

講義指定図書 Reading List

電気化学測定法(上)/藤嶋昭,相澤益男,井上徹:技報堂出版, 1984 Electrochemical Methods, Fundamentals and Applications, 2nd ed./Allen J. Bard, Larry R. Faulkner:Wiely, 2001 Analytical and Physical Electrochemistry/Hubert H. Girault:EPFL Press

参照ホームページ Websites

研究室のホームページ Websites of Laboratory

https://elechem.eng.hokudai.ac.jp/

科目名 Course Title	六田暢所化学(無機暢耕化学)[A==1:-1] (1-==:1:-1)					
	応用物質化学(無機物性化学)[Applied Materials Chemistry (Inorganic Solid State					
講義題目 Subtitle	Chemistry)]					
責任教員 Instructor	鱒渕 友治 [MASUBUCHI Yuji] (大学院工学研究院)					
担当教員 Other Instructors						
科目種別 Course Type			T			
開講年度 Year	2024	時間割番号 Course Number	094208			
期間 Semester	Fall	単位数 Number of Credits	1			
授業形態 Type of Class	Lecture	対象年次 Year of Eligible Student	~			
対象学科 クラス Eligible Depa	rtment/Class					
ナンバリングコード Numbering	Code	CHEM_ELMAT 6102				
補足事項 Other Information						
授業実施方式 Class Method		1 対面授業科目《対面のみ》				
キーワード Key Words						
Sintering, Thin film, Single crys	stal, Nano materials	s, Morphology				
授業の目標 Course Objectives	S					
Inorganic solids are known to	o show various p	roperties depending on their constituent	elements and crystal structure.			
Additionally, from the viewpo	int of "material"	their morphology and microstructure mu	st be optimized to achieve their			
applications. This lecture will b	e dealing with prep	paration process of sintered body, thin film,	, single crystal, and nano materials			
for inorganic materials. We will	also discuss how t	heir physical properties relate to their mor	phology and micro structure.			
到達目標 Course Goals						
To explain a relationship betwe	en various propert	ies and microstructures in functional inorga	nic solids.			
To explain preparation methods	s of sintered body,	thin film, single crystal, and nano materials	of functional inorganic solids.			
To explain fundamental mechan	nisms of diffusion, n	ucleation, crystal growth, and grain growth	in functional inorganic solids.			
授業計画 Course Schedule						
1. Introduction: properties and morphology of inorganic solids						
2. Sintering: solid and liquid phase diffusion, sintering of metal nitrides						
3. Thin film: deposition process, vacuum deposition, vapor and liquid phase deposition						
4. Single crystal: crystal growth	n mechanism, vario	us crystal growth process				
5. Nano material: properties, n	ano particles, com	posites, assemblage				
準備学習 (予習・復習)等の内容	容と分量 Homewor	k				
In order to improve the learn	ing, students are	encouraged to prepare for and review the	e topics in "Course Schedule" by			
referring to the appropriate se	ctions of the hand	outs and scientific papers, etc., in the time	allotted by the regulations of the			
Faculty of Engineering.						
成績評価の基準と方法 Gradin	g System					
Comprehensively evaluate the	degree of "Course	e Goals" from the results of the exercise d	uring the class and a final report.			
Breakdown of the evaluation shall be exercise: 30%, final report: 70%, a total of more than 60 points are required to obtain the						
credit.						
他学部履修の条件 Other Faculty Requirements						
テキスト・教科書 Textbooks						
適宜、資料を配付する。						
講義指定図書 Reading List						
参照ホームページ Websites						
https://www.eng.hokudai.ac.jp						
研究主の小一ムヘーン Websit	es ot ladoratory		研究室のホームページ Websites of Laboratory			

科目名 Course Title	応用物質化学(電子材料化学特論)[Applied Materials	Chemistry (Physical Chemistry of		
	Electronic Materials)]				
講義題目 Subtitle		/_			
責任教員 Instructor	青木 芳尚 [AOKI Yoshitaka] (大学院工学研究院)				
担当教員 Other Instructors					
科目種別 Course Type					
開講年度 Year	2024	時間割番号 Course Number	094209		
期間 Semester	Winter	単位数 Number of Credits	1		
授業形態 Type of Class	Lecture	対象年次 Year of Eligible Student	\sim		
対象学科・クラス Eligible Depa					
ナンバリングコード Numbering		CHEM ELMAT 6102			
補足事項 Other Information	Code	CHEM_EEMAT 0102			
授業実施方式 Class Method		 1 対面授業科目《対面のみ》			
		1 对面这条杆百《对面切夺//			
キーワード Key Words		hat we have a first the second second			
		heterojunctions, defect thermodynamics			
授業の目標 Course Objective		l ovido fuel colla, ell colid state battery, by	thrid color colla		
到達目標 Course Goals	ices including solid	l oxide fuel cells, all solid state battery, hy			
可定日標 Course Goals Fundamentals of all solid state	alactrochomical de	Nicos			
To understand the phenomena					
Band structures at meta/semic					
Interplays between ion and elec					
授業計画 Course Schedule					
1. Introduction of band theory					
	c activity and elec	tronic properties of Pt ORR catalysts			
3. Fundamental of electrochem					
4. Correlation between elecron	-				
5. Design of solid state ionics of					
		-plays between ion and electron carriers.			
準備学習(予習・復習)等の内					
Students are encouraged o une	derstand the worki	ng principals of fuel cells and all solid sta	te batteries, fundamentals of defect		
thermodynamics and basic cond	cepts of electronic	properties at hetero-interfaces.			
成績評価の基準と方法 Gradin	ig System				
The scores are determined by	(1) questions or di	scussion after lectures (30%), (2) learning a	attitude (10%) and (3) reports at end		
of semester					
他学部履修の条件 Other Faculty Requirements					
テキスト・教科書 Textbooks					
Physics of semiconductor devices/S. M. Sze					
電極化学 上/佐藤教男					
講義指定図書 Reading List					
参照ホームページ Websites					
研究室のホームページ Websit	-				
https://ionics.eng.hokudai.ac.	p/index-e.html				
備考 Additional Information					

科目名 Course Title	応用物質化学(機能固体材料化学)[Applied Materials Chemistry (Functional So					
	Materials Ch	emistry)]				
構義題目 Subtitle						
任教員 Instructor		島田 敏宏 [SHIMADA Toshihiro] (大学院工学研究院)				
目当教員 Other Instructor	S					
科目種別 Course Type	0004		004010			
鼎講年度 Year 別間 Semester	2024 Intensive	時間割番号 Course Number 単位数 Number of Credits	094210			
設業形態 Type of Class	Lecture	本位				
大学校・クラス Eligible D	L					
ーンバリングコード Numbe		CHEM_ELMAT 6101				
都足事項 Other Information						
受業実施方式 Class Meth		4 遠隔授業科目《遠隔のみ》				
Fーワード Key Words						
	and the "heart" of rials. The second g	chemistry and physics of solid state function goal is to understand what is written in literat to achieve this goal.				
. Introduction to solid sta 2. Semiconductors focused 3. Transparent conductors 4. Advanced ligand field the 5. Interfaces: work function 6. Phase memory materials 7. Ferroelectrics and liquid 8. Thermography and stron Related theoretical concep 集備学習 (予習・復習)等の Preparation: read the hand	of solid state mate e about the related wing list can also b te chemistry / phy on solar cells (oxides, nanowires eory and basics of a and chemistry of (DVD-R/W, shap crystal agly correlated elect ts will be introduce D内容と分量 Home out posted on the	rials. topics. e lectured according to request. sics and thermoelectricity s, graphene) photophysics - lasers, nonlinear optics, optic semiconductor junction devices e memory alloys) etron systems ed every time. ework website (URL will be given at the first lecture				
Homework: solve the probl	em given in the lea	cture and write a brief final report.				
成績評価の基準と方法 Gr						
Grading is based on the qu 也学部履修の条件 Other						
テキスト・教科書 Textbook Handout will be given prior 講義指定図書 Reading Lis	to the lecture via	website				
参照ホームページ Website	NC					
W 552/1) 14 14 10 16 17 10 17	10					

科目名 Course Title	目名 Course Title 応用物質化学(先端材料化学)[Applied Materials Chemistry (Advanced Materials Chemistry)]			
講義題目 Subtitle				
責任教員 Instructor	北川 裕一 [KITAGAWA Yuichi] (大学院工学研究院)			
担当教員 Other Instructors				
科目種別 Course Type				
開講年度 Year	2024	時間割番号 Course Number	094211	
期間 Semester	Summer	単位数 Number of Credits	1	
授業形態 Type of Class	Lecture	対象年次 Year of Eligible Student	~	
対象学科・クラス Eligible Depa	rtment/Class			
ナンバリングコード Numbering	Code	CHEM_ELMAT 6102		
補足事項 Other Information				
授業実施方式 Class Method		1 対面授業科目《対面のみ》		
キーワード Key Words				
Molecular photochemistry, ligh	t absorption, lumine	escence, organic compound, metal complex		
授業の目標 Course Objective	S			
In this course, advanced photo	ofunctional material	s and fundamental principles of photocher	nistry are presented. This course	
enhances the understanding of	advanced photofund	ctional research and the ability to design ph	notofunctional materials.	
到達目標 Course Goals				
Students will be able to unders	stand basic concept	s of photochemistry such as electronic ene	ergy in materials, light absorption,	
and excited state dynamics	to understand the	e basic principles of designing photofur	nctional materials and advanced	
photofunctional material studi	es. The goal of thi	s course is to provide students with suf	ficient background to understand	
photofunctional studies in varie	ous research fields.			
授業計画 Course Schedule				
1–2. Fundamentals of photoche	emistry			
3. Light absorbing materials				
4–5. Luminescent materials				
6. Polarized absorbing and lum				
7. Photo-induced electron tran	sfer•Photochemical	reaction		
8. Examination				
淮井尚辺(ス辺、復辺)年の市。	ㅎ 니 / 르 니	-		
準備学習(予習・復習)等の内容				
		the lecture slide. To enhance a learning eff	lect, the students are expected to	
review and prepare for about tw 成績評価の基準と方法 Gradin		outs.		
成績計画の基準と方法 Grading According to the class attitude		will be calculated		
According to the class attitude	and test, the score	will be calculated.		
他学部履修の条件 Other Faculty Requirements				
世子印度修び末件 Other Faculty Requirements				
テキスト・教科書 Textbooks				
講義指定図書 Reading List				
参照ホームページ Websites				
研究室のホームページ Websites of Laboratory				
	町先主の小一ムハーン Wedsites of Laboratory			
備考 Additional Information				

科目名 Course Title	応用物質化学(応用材料化学 I)[Applied Materials Chemistry (Applied Inorganic Materials			
	Chemistry I)]			
講義題目 Subtitle				
責任教員 Instructor	忠永 清治[]	「ADANAGA Kiyoharu」(大学院工学研究院)		
担当教員 Other Instructors	KIJIMA Norih	ito (AIST), SUE Kiwamu (AIST)		
科目種別 Course Type				
開講年度 Year	2024	時間割番号 Course Number	094212	
期間 Semester	Fall	単位数 Number of Credits	1	
授業形態 Type of Class	Lecture	対象年次 Year of Eligible Student	~	
対象学科・クラス Eligible Dep	artment/Class			
ナンバリングコード Numbering Code		CHEM_ELMAT 6100	CHEM_ELMAT 6100	
補足事項 Other Information				
授業実施方式 Class Method		1 対面授業科目《対面のみ》		
キーワード Key Words				

Functional inorganic materials, Secondary batteries, Nanostructural analysis, High-temperature/high-pressure solvent, Flow production, Data-driven materials development

授業の目標 Course Objectives

The relation between the functionality of materials, especially inorganic functional materials, and their nanostructure and macroscopic form such as bulk will be lectured.

The appearance mechanism of various functionality obtained by controlling the composition and microstructure of the materials, the process development method for production of functional materials with desired structures, and feature of high-temperature / high-pressure solvent properties and controlling methods of the properties, will be addressed.

The prospects for industrial application in the future will be discussed by taking up some topics, including the synthesis and characterization of electrode and electrolyte materials for lithium ion batteries, and the production of nanoparticles and nanocomposites.

到達目標 Course Goals

The relation between the materials properties and nano/micro-structures drawing the required functions will be understood. In addition, the basic science and skills for materials processing and analysis will be mastered. The task to find a topic for oneself and investigate it will be given to the students as training to select and treat information scientifically.

授業計画 Course Schedule

Lectures will be given by Professor Professor Norihito KIJIMA and Kiwamu SUE(AIST).

The following contents will be lectured using the documents edited for the class by the lecturers:

1. Materials chemistry of secondary batteries (Rechargeable Batteries): Overview of secondary batteries, component materials for secondary batteries, and situation surrounding storage batteries.

2. Materials chemistry for secondary batteries: Preparation and characterization of materials for batteries

3. Fundamentals for development of functional materials production process: high-temperature / high-pressure solvent properties, controlling method of the properties such as flow production, application to functional materials production

4. Data-driven method for development of materials production process: how to develop apparatus, how to collect and analyze data, application to functional materials production

準備学習 (予習・復習)等の内容と分量 Homework

Review the distributed documents and blackboard demonstration contents, and ask any questions at the next class.

成績評価の基準と方法 Grading System

Your attitude in classes (20%) and reports (80%) will affect your final grade.

他学部履修の条件 Other Faculty Requirements

テキスト・教科書 Textbooks

なし。適宜資料を配布する。

None. Materials will be distributed as appropriate.

講義指定図書 Reading List

参照ホームページ Websites

https://www.aist.go.jp/

研究室のホームページ Websites of Laboratory

備考 Additional Information

Materials will be distributed as appropriate.

科目名 Course Title	応用物質化学(応	応用物質化学(応用材料化学Ⅱ)[Applied Materials Chemistry (Applied Inorganic Materials			
, and the second se	Chemistry II)]				
講義題目 Subtitle					
責任教員 Instructor	忠永 清治 [TAD	ANAGA Kiyoharu] (大学院工学研究院)			
担当教員 Other Instructors	KUWATA Naoaki	(NIMS), KUBOTA Kei (NIMS)			
科目種別 Course Type					
開講年度 Year	2024	時間割番号 Course Number	094213		
期間 Semester	Fall	単位数 Number of Credits	1		
授業形態 Type of Class	Lecture	対象年次 Year of Eligible Student	~		
対象学科・クラス Eligible Depar	rtment/Class				
ナンバリングコード Numbering Code		CHEM_ELMAT 6100			
補足事項 Other Information					
授業実施方式 Class Method		1 対面授業科目《対面のみ》			
キーワード Key Words					
Inorganic solid materials, mate	erials processing, b	attery materials, materials analysis, diffus	sion in solids, thermodynamics of		
batteries, ion dynamics measure	ements				

授業の目標 Course Objectives

For various materials, mainly inorganic materials, you will learn about the functionality obtained by controlling the composition, crystal structure and microstructure, and the mechanism by which they are expressed. For example, solid-state battery materials will be subjected to understand the fundamentals of material properties through a physicochemical approach. Ion dynamics measurement methods including nuclear magnetic resonance will be introduced too. Students will also learn about processing and characterization methods (instrumental analysis, spectroscopy, etc.) for fabricating functionally designed materials.

到達目標 Course Goals

Students will be able to understand the relationship between various properties of materials and nano-micro-macro structures through the example of battery materials and the mechanism by which these various physical properties are expressed. In addition, you will be able to consider what microstructure should be designed to maximize the desired function, and what kind of method should be used to obtain such a structure in terms of "materials processing".

授業計画 Course Schedule

The following contents will be lectured using the distributed materials.

- 1. Introduction $\ddot{}$ About the structure and function development of materials.
- 2. Synthesis: Synthesis theory and process chemistry for grinding, sintering, and microstructure control.

3. Characteristic evaluation "Relationship between nano-micro-macro structure of materials and electrochemical properties. Also, about their evaluation methods.

4. Summary: Industrial application and future prospect of material sciences and materials technology. About the role and potential of materials in a sustainable society.

準備学習 (予習・復習)等の内容と分量 Homework

Review the distributed lecture materials and contents, and ask questions in the next class.

成績評価の基準と方法 Grading System

Attendance of 75% or more of the number of classes is a condition for grade evaluation. Grades are evaluated based on (1) learning attitudes (20%) and (2) reports (80%).

他学部履修の条件 Other Faculty Requirements

テキスト・教科書 Textbooks

なし。適宜資料を配布する。

No textbook required. Materials will be distributed each time.

講義指定図書 Reading List

参照ホームページ Websites

https://www.nims.go.jp/

研究室のホームページ Websites of Laboratory

科目名 Course Title	生物化学A(I)[Biochemistry A (I)]	
溝義題目 Subtitle			
責任教員 Instructor	髙橋 正行 [TA	KAHASHI Masayuki] (大学院理学研究院	2)
旦当教員 Other Instructors	ABE Kazuhiro	(理学研究院)	
科目種別 Course Type			
開講年度 Year	2024	時間割番号 Course Number	094301
朝間 Semester	Fall/Winter	単位数 Number of Credits	2
受業形態 Type of Class	Lecture	対象年次 Year of Eligible Student	~
対象学科・クラス Eligible Dep	artment/Class		
トンバリングコード Numbering	g Code	CHEM_ELBIO 6012	
甫足事項 Other Information			
受業実施方式 Class Method		1 対面授業科目《対面のみ》	
キーワード Key Words			
· · · · · · · · · · · · · · · · · · ·	transporters, ion	channels, structural biology, cryo-EM,	drug design, motor protein, musc
ontraction, cell motility, cyto			
受業の目標 Course Objective			
s an important basis of liv	ving systems, var	ious substrates including cations are as	symmetrically distributed across th
nembrane, which is generate	d and maintained	by ATP-driven primary active transporte	rs; Free energy derived from ATP
lso utilized for muscle cont	raction and variou	s cellular processes. In this lecture, stu	dents will learn about the molecula
nechanisms of membrane pro	oteins that genera	te asymmetric distribution of materials,	and also for the function of variou
roteins that involved in the r	naintenance and al	teration of cellular morphology.	
创達目標 Course Goals			
tudents are expected to dee	ply understand th	e molecular mechanism of membrane prot	teins, including primary transporter
com the viewpoint of structur	al biology, and the	e molecular mechanism of muscle contracti	on and various cell motile processes
受業計画 Course Schedule			
) How to "look at" the prote	in shape		
2) Understand protein function	ns in terms of Che	mistry	
3) Membrane proteins			
4) Molecular mechanism of pri	imary transporters		
5) Structural physiology of pri	mary transporters		
6) Structural based drug deve	lopment 1		
7) Structural based drug deve	lopment 2		
3) Molecular mechanism of mu		nd its regulation	
9) Structure and mechanism o	-		
0) Dynamics of cytoskeletal p	proteins		
1) Molecular mechanism of c	ell migration		
2) Molecular mechanism of c			
3) Morphological changes of	neuronal cells		
隼備学習 (予習・復習)等の内			
宇備子自てア自己返自)寺の Students are expected to revi			
或績評価の基準と方法 Gradi			
		class is a requirement for grading. Evalua	tion is based on (1) attitude toward
		inations (60%). The reports will assess the	
		determine the ability to apply the knowled	
也学部履修の条件 Other Fac			Se Serree.
テキスト・教科書 Textbooks			
時にもうけない。			
溝義指定図書 Reading List			

研究室のホームページ Websites of Laboratory

科目名 Course Title	生物化学A(Ⅱ	生物化学A(II)[Biochemistry A (II)]			
講義題目 Subtitle	生体システムの	生体システムのシグナル伝達―形態形成と生体防御[Signal Transduction for Biological			
	Morphogenesis	Morphogenesis and Host Defense Systems]			
責任教員 Instructor	茂木 文夫 [M	OTEGI Fumio] (遺伝子病制御研究所)			
担当教員 Other Instructors	TAKAOKA Ak	TAKAOKA Akinori (遺伝子病制御研究所)			
科目種別 Course Type					
開講年度 Year	2024	時間割番号 Course Number	094302		
期間 Semester	Fall/Winter	単位数 Number of Credits	2		
授業形態 Type of Class	Lecture	対象年次 Year of Eligible Student	~		
対象学科・クラス Eligible Depa	artment/Class				
ナンバリングコード Numbering Code		CHEM_ELBIO 6012			
補足事項 Other Information					
授業実施方式 Class Method		1 対面授業科目《対面のみ》			
キーロード Key Words					

Signal transduction, genetic mutations, cellular architectures and morphogenesis, host defense, molecular mechanisms of diseases, immunology, basic medicine, infectious diseases, cancer, basic skills for scientific writing, experimental techniques of cell biology/molecular biology/immunology

授業の目標 Course Objectives

Life system can be considered as an orchestral unit, which is composed of multiple biomolecular components. Each part of the orchestra exquisitely responds and/or adjusts to various external and internal stresses to keep the whole harmony, which is important for the homeostasis of life system. On the other hand, dysfunction of some part makes the orchestra play an inharmonic music as a whole, leading to breakdown in the homeostasis of life system, that is, developing a disease. Thus, the life system is spatiotemporally regulated under the molecular networks, which are controlled by the biochemical machinery. This course aims at the stepwise understanding about the static organization of life system and its dynamic changes in response to stresses, from molecular to cellular, and further to individual aspects. In particular, we focus on the signaling transduction system that underlies the host defense against pathogen invasion and cancer development.

到達目標 Course Goals

The final goal of this course is to foster an integrative understanding and research view of chemistry by learning physiological functions of biomolecular components and their dysfunctions as a pathogenic condition (i.e., a disease development) from an interdisciplinary view of chemistry and medicine. We hope that this course contributes to rearing a researcher with a broad-based knowledge below the chemistry as well as the ability of creative and imaginative thinkings. This course will also support students to learn the general process of paper publication, how to write a "fascinating" research paper as well as how to prepare the accessory documents, and basic skills to make better use of molecular biological or immunological approaches.

授業計画 Course Schedule

This module will cover two major topics in (1) Cellular and tissue morphogenesis and (2) The host defense system during normal and disease contexts. A major challenge in biology is how to comprehend the enormous complexity underlying biological systems, and how to translate this knowledge into biomedical technologies. A set of lectures aims to understand how molecules interact with each other to produce the signals that orchestrate complex physiological functions. The detailed schedule will be informed in the first day of the course. Lecture contents, such as color-printed synopsis and/or lecture slide notes, will be provided in advance of each lecture.

Section 1: Molecular mechanisms of cellular and tissue morphogenesis

This section aims to understand the basic principles in animal body plan by introducing molecular mechanisms underlying "morphogenesis of cells and tissues" during normal development and in disease conditions. Growth and form are fundamental to all living organisms, and crucial to health and diseases. Development in methods and tools for molecular imaging has transformed biological and biomedical sciences. In particular, this section will introduce several basic concepts in molecular imaging with light microscopy and their applications. Introduction of each imaging technology will be linked with a set of biological problems of fundamental interests and biomedical implications. With a quantitative and holistic understanding of how molecular functions are ensured during normal development and how they are maladapted in disease, students will gain better insights into biomedical implications that effectively protect and regenerate organ functions and to better treat human diseases.

Section 2: The host defense system against pathogen invasion and cancer development

The second section will introduce the physiological functions of human organs from a macroscopic viewpoint, zooming them down to the functions at cellular and molecular levels. Next, we provide additional explanation about host responses to external and internal stresses to keep homeostasis in human living organism. Particularly, we focus on host defense against microbial infection that is an external stress, i.e., immunity. Students explore the following issues from the point of view of signal transduction as a cascade of intracellular chemical reactions: How does a living organism recognize invasion by microbes? What is the molecular mechanism for the specific elimination of the invading pathogens? In addition, we further review a mechanism for host defense against oncogenesis, which can be considered as an internal stress.

Students will also learn about the molecular mechanism for disease pathogenesis, particularly in terms of immunodeficiency that develops as a result of impairment of the immune system due to genetic abnormalities. Here, students further explore the outcome of abnormalities of biochemical events in human at the molecular to cellular and in vivo levels. This is helpful to students to acquire the relevant knowledge about basic medicine and to deepen their understanding of not only disease pathogenesis, but also the molecular-based strategy for disease treatment.

Section 3: Practical techniques to write "attractive" scientific papers.

This course will provide students the opportunity to learn research-based knowledge and skills in a more practical way. Students have a 10-min explanation about basic experimental techniques of molecular biology/immunology at the end of every class, by introducing a scientific paper published in a major journal such as Nature and Science. In addition, this part covers an overview of manuscript process (from submission to revision, resubmission, and acceptance) and also conveys essentials to a high-quality paper, by introducing actual examples of a manuscript and its related documents (Cover letter, Review comments, Rebuttals, Proof, etc.), which were successfully accepted in high-profile journals.

Section 1: Cellular and tissue morphogenesis in normal and disease contexts

(1) Visualization of biological molecules in vivo

(2) Visualization of biochemical reactions in vivo

(3) Cell and tissue morphogenesis (I)

(4) Cell and tissue morphogenesis (II)

(5) Basics of scientific presentation

(6) How to make effective scientific presentation (I)

(7) How to make effective scientific presentation (II)

(8) Practical exercise of scientific presentation

Section 2: The host defense system against pathogen invasion and cancer development

§ 2.1: Physiology of life system

(1) Macroscopic presentation of human body including a anatomical structure, biochemical, physiological functions of each organs

(2) General introduction of host defense (innate immunity and adaptive immunity)

(3) Basic knowledge of antibody molecules and their clinical application

(4) Roles of immune cells (e.g., dendritic cells, lymphocytes) and molecular mechanisms whereby these cells are functionally activated.

(5) Soluble factors that regulate the immune system, and their mechanisms of action

§ 2.2: Pathology of life system

(6) Fundamental knowledge of pathogenic microbes (e.g., viruses and bacteria)

(7) Diseases and pathological conditions as a result of breakdown in life system (e.g., infectious diseases, cancers)

(8) Molecular mechanisms for disease pathogenesis (genetic abnormalities and immunodeficiency)

(9) Therapeutic principle that is based on a molecular abnormality responsible for a disease (e.g., gene therapy)

Section 3: Basic knowledge of research and its practical application

(1) Basic and application of experimental methods that are often used in the research field of molecular biology/immunology

(2) Overview of manuscript process (from submission to revision, resubmission, and acceptance)

(3) Essentials to a high-quality paper

準備学習 (予習・復習)等の内容と分量 Homework

There is no obligatory assignment that students have to prepare or review during this course. We think that what is essential is that students can maximally concentrate their attentions on each class and find something interesting to move them to spontaneously explore it further. We therefore will make maximum efforts to make each class attractive and to support students to learn the topic of each class in an extended manner.

成績評価の基準と方法 Grading System

During our interactive classes, we consider it as one of the important factors for assessment how actively students participate in each class (PARTICIPATION). In this respect, for example, to spontaneously participate in Q&A activity and to think logically are much more important than to simply make a correct answer. Basically, there is no written exam for assessment, but students will be required to submit a report once at the end of this course. This report is regarded as a main factor for assessment (REPORT). As a theme of report, students can make a choice of one topic that they have become most interested in during this course. No specific format of report, and students are recommended to describe a topic of interest together with more detailed information that they additionally investigate by themselves, and to include some content of their research themes in a way that creates a link to a topic of their choice.

The course employs the grading system that is generally used for the student assessment in this university: Excellent plus, excellent, very good, good, and not good. In addition to the basic score of ATTENDANCE, the following major factors are

considered to comprehensively and fairly make the final assessment:1) Course assignment such as presentation and report: 70%2) Active attitude to lectures and learning: 30%

他学部履修の条件 Other Faculty Requirements

テキスト・教科書 Textbooks

講義指定図書 Reading List

参照ホームページ Websites

研究室のホームページ Websites of Laboratory

Motegi lab homepage: https://www.motegilab.com Takaoka lab homepage: https://www.igm.hokudai.ac.jp/sci/

備考 Additional Information

Feel free to contact us for further information.

Fumio Motegi, Ph. D. Division of Developmental Physiology, Institute for Genetic Medicine, Hokkaido University Tel:011-706-5527; ext. 5527 E-mail: motegi@igm.hokudai.ac.jp

Akinori Takaoka, M.D., Ph.D., Division of Signaling in Cancer and Immunology, Institute for Genetic Medicine, Hokkaido University Phone 011-706-5020; ext. 5020 E-mail takaoka@igm.hokudai.ac.jp

科目名 Course Title	生物化学A(Ⅲ)	Biochemistry A (III)]	
講義題目 Subtitle	T.W.17.1.(III)[
責任教員 Instructor	内田 毅「UCHII	DA Takeshi] (大学院理学研究院)	
担当教員 Other Instructors	TIH 3X LUUIII	パーローのにに、「「「「「「「」」」」、「「「」」」、「「」」、「「」」、「「」」、「」」、	
科目種別 Course Type			
開講年度 Year	2024	時間割番号 Course Number	094303
期間 Semester	Spring/Summer	単位数 Number of Credits	2
授業形態 Type of Class	Lecture	対象年次 Year of Eligible Student	\sim
対象学科・クラス Eligible Depa			
ナンバリングコード Numbering		CHEM ELBIO 6012	
補足事項 Other Information			
授業実施方式 Class Method		1 対面授業科目《対面のみ》	
文条 ジェン Class Method キーワード Key Words		1 对面段来行口《对面9907//	
-	rod Sportroscopy	Fluorescence Spectroscopy, Raman Sc.	attoring Magnetic Resonance Single-
nolecular Detection	area spectroscopy,	Fuorescence spectroscopy, Raman Sc	attering, Magnetic Resonance, Single
授業の目標 Course Objective	e		
		uctures of biological molecules such a	as proteins nucleic acids and othe
		its with basic theories of spectroscopie	-
applications.	provide studel	the man basic meeties of specificscopie	e and knowledge about their biologica
到達目標 Course Goals			
	ound and basic theo	ories of various kinds of spectroscopies	for analyzing structures and function
of biological molecules.			
授業計画 Course Schedule			
[1st Half]			
Explain the basic theory of som	ne spectroscopies.		
Week 1: Orientation and Introd			
Week 2: Basic Theory of Mass		iochemistry	
Week 3: Basic Theory of Absor	rption Spectroscop	y in Biochemistry	
Week 4: Basic Theory of Infrar	ed Spectroscopy in	Biochemistry	
Week 5: Basic Theory of Rama	n Spectroscopy in	Biochemistry	
Week 6: Basic Theory of Fluor	escence Spectrosc	opy in Biochemistry	
Week 7: Basic Theory of Circu	lar Dichroism Spec	troscopy in Biochemistry	
Week 8: Basic Theory of Nucle	ear Magnetic Reson	ance Spectroscopy in Biochemistry	
		on and Other Spectroscopic Technique	s in Biochemistry
Week 10: Presentation by stud	ents		
[2nd Half]			
Explain the application of the s			
Week 11: Biological Application			
Week 12: Biological Application		-	
Week 13: Biological Application			
Week 14: Biological Application	n of Nuclear Magne	etic Kesonance	
Week 15: Exercise 維備營習 (系習-復習)等の中国	あし八月 ロ	1.	
準備学習 (予習・復習)等の内容		ĸ	
Assignment is required for even deなでのすまた。	-		
成績評価の基準と方法 Gradin Quiz & Assignment, 70%; exam			
Quiz & Assignment, 70%; exam 他学部履修の条件 Other Fac			
テキスト・教科書 Textbooks			
講業也中國書 Daa June 199			
講義指定図書 Reading List	an Condunt I M	御·Cambridge 2007	
Methods in Molecular Biophysi アトキンフ 物理化学(下) 笠 10			
アトキンス 物理化学(下) 第 10 生体公子公光学 1 胆 / 尾崎 5			
生体分子分光学入門/尾崎 雪 参照ホームページ Websites	产任、石憰 労大:尹	<⊥山版,1992	
❷ポパハームハーン Websites			
研究室のホームページ Websit	-		
nttps://www.chem.sci.hokudai.a			

On-site classes are expected. In principle, attendance of 70% or more of the class sessions is required for credit.

科目名 Course Title	生物化学A(IV)[F	Biochemistry A (IV)]		
講義題目 Subtitle				
責任教員 Instructor	谷野 圭持 [TAN]	INO Keiji] (大学院理学研究院)		
担当教員 Other Instructors	SUZUKI Takahiro	(理学研究院)		
科目種別 Course Type				
開講年度 Year	2024	時間割番号 Course Number	094304	
期間 Semester	Fall/Winter	単位数 Number of Credits	2	
授業形態 Type of Class	Lecture	対象年次 Year of Eligible Student	~	
対象学科・クラス Eligible Depar	rtment/Class			
ナンバリングコード Numbering	Code	CHEM_ELBIO 6012		
補足事項 Other Information				
授業実施方式 Class Method		1 対面授業科目《対面のみ》		
キーワード Key Words				
Carbocation, Lewis acid, Enc	ol silyl ether, Allyl	silane, Electrophilic addition reaction, Ca	arbon radical, Radical reduction,	
Radical addition reaction, Radical cyclization reaction				
授業の目標 Course Objectives				

The chemistry of enol silyl ethers as well as allylsilanes provides powerful methods in modern organic synthesis. The reactions of these compounds usually proceed through cationic intermediates, and it is very important to know the properties of carbocation species. This course increases students' understanding of useful carbon-carbon forming reactions mediated by Lewis acids.

This lecture also discuss about the features of carbon radicals such as the relationship between structure and stability, typical generation methods, and the addition to the multiple bonds. The attainment target is acquirement of practical knowledge, which enable comprehension of the complex synthetic schemes.

到達目標 Course Goals

At the end of the course each student should be able to:

1. explain the relationship between structure, stability, and reactivity of various carbocation species.

2. explain the "scope and limitations" in reactions involving carbocation intermediates, comparing with those in reactions of carbanion species or organometallic compounds.

3. discuss the mechanism of the reactions mediated by a Lewis acid by assuming appropriate reactive intermediates.

4. explain the relationship between structure, stability, and reactivity about various carbon radicals.

5. explain the "scope and limitations" about the reactions using radicals in comparison with the common ionic reactions.

6. discuss the mechanism of the cascade reactions involving radical species.

7. suggest appropriate schemes involving a several-step transformation for the synthesis of small organic molecules.

授業計画 Course Schedule

1. general properties of carbocation species

2. methods for generating carbocation species

3. preparation and reactions of enol silyl ethers

4. preparation and reactions of allylsilanes

5. reactions of vinylsilanes and alkynylsilanes

6. Prince reaction and carbonyl-ene reaction

7. alkylation reaction using organometallic reagents

8. basic property and generation of radical species

9. radical reduction by using alkyltin hydrides

10. radical reduction by using low valent metal salts

11. addition reactions of carbon radical with alkenes

12. radical cyclization leading to carbocycles or heterocycles

準備学習 (予習・復習)等の内容と分量 Homework

Students are requested to take enough time to go over each subject noted down during the lecture. A full understanding of the reaction mechanisms is especially important.

成績評価の基準と方法 Grading System

based on a mid-term examination (50%) and a term examination (50%)

他学部履修の条件 Other Faculty Requirements

テキスト・教科書 Textbooks

Textbooks are not assigned. 講義指定図書 Reading List

参照ホームページ Websites

研究室のホームページ Websites of Laboratory

https://www.chem.sci.hokudai.ac.jp/~oc2/

科目名 Course Title	応用生物化学	(生合成工学)[Applied Biochemistry (Biosy	vnthetic and Metabolic Engineering)]
講義題目 Subtitle			
責任教員 Instructor	大利 徹 [DAI!	RI Toru] (大学院工学研究院)	
担当教員 Other Instructors	OGASAWARA Yasushi (工学研究院), SATOH Yasuharu (工学研究院)		
科目種別 Course Type			
開講年度 Year	2024	時間割番号 Course Number	094305
期間 Semester	Fall	単位数 Number of Credits	1
授業形態 Type of Class	Lecture	対象年次 Year of Eligible Student	\sim
対象学科・クラス Eligible Depa	rtment/Class		
ナンバリングコード Numbering	Code	CHEM_ELBIO 6102	
補足事項 Other Information			
授業実施方式 Class Method		1 対面授業科目《対面のみ》	
キーワード Key Words			
-	ts, biosynthesis,	genes, enzymes, bioinformatics	
授業の目標 Course Objective		<u> </u>	
		es essential for biotechnology/bioengineer	ing with microorganisms 1 Principle
-	· ·	of enzyme reaction, 3. Outline of primary/	· · · ·
	Menten kinetics	of enzyme reaction, 5. Outline of primary/	secondary metabolites and metabolic
pathways. 到達目標 Course Goals			
	orstand papers al	pout the microbial metabolites/metabolic p	athways and the enzymes responsible
		hnology to their own research subjects.	attiways and the enzymes responsible
授業計画 Course Schedule		infology to their own research subjects.	
反来計画 Course Schedule 1. Introduction			
2. Principle of bioinformatics			
 Principle of bioinformatics Michaelis Menten kinetics o 	f anguma pasatio	n_1_	
4. Michaelis Menten kinetics o			
5. Review of the primary metal			
6. Diversity of the primary metal		microorganisms	
		bolites and their biosynthetic pathways	
		ompounds based on biosynthetic engineerin	g and motabolic onginooring
準備学習(予習・復習)等の内	-		g and metabolic engineering
		elated to biochemistry and summarize its co	ntents concisely
成績評価の基準と方法 Gradir		stated to blochemistry and summarize its co	intents concisely.
		is evaluated by learning volition (20%) and	the quality of reports (80%)
他学部履修の条件 Other Fac	-		the quality of reports (00%).
テキスト・教科書 Textbooks			
適宜資料を配布する。下記の	絵老書を堆将す7	いが教科書は毎日したい	
講義指定図書 Reading List			
	ケミカルバイオロ・	ジー理解のために/John McMurry, Tadhg	Beglev 著・浦野泰昭 [ほか] 訳・東
京化学同人, 2007		Jent Victor John Memory, Taung	
	osistanco / Chris	topher Walsh:ASM Press, 2003	
		ネルソン,コックス[著];中山和久編集:廣川	川圭店 2010
		1 W. Mount 監訳:岡崎康司、坊農秀雅 :	
ターナショナル,2005	Lanton / David		
<u>参照ホームページ</u> Websites			
研究室のホームページ Websit	tes of Laborator	,	
https://www.eng.hokudai.ac.jp			
備考 Additional Information			
	o bosio knowloda	of high pristry	

Students are requested to have basic knowledge of biochemistry.

科目名 Course Title	応用生物化学	応用生物化学(生命システム工学)[Applied Biochemistry (Biosystem Engineering)]			
講義題目 Subtitle					
責任教員 Instructor	菊川 寛史[K	KIKUKAWA Hiroshi] (大学院工学研究院)			
担当教員 Other Instructors	HIRAISHI Tor	nohiro (RIKEN)			
科目種別 Course Type					
開講年度 Year	2024	時間割番号 Course Number	094306		
期間 Semester	Fall	単位数 Number of Credits	1		
授業形態 Type of Class	Lecture	対象年次 Year of Eligible Student	~		
対象学科・クラス Eligible Dep	oartment/Class				
ナンバリングコード Numbering Code		CHEM_ELBIO 6100			
補足事項 Other Information					
授業実施方式 Class Method		1 対面授業科目《対面のみ》	1 対面授業科目《対面のみ》		
ナーロード Kau Wanda					

gene, protein, transcription, translation, enzyme, biochemistry, evolutionary engineering, molecular design, protein engineering, physical chemistry, bioplastic, biodegradation, biotechnology, genome, omics, metabolic engineering, synthetic biology, bacteria, fungi

授業の目標 Course Objectives

Organisms have an excellent synthetic mechanism to produce complex molecules and to properly degrade and utilize them. This system contributes to the natural element cycle. Enzymes play a central role in this cycle. Biotechnologies that apply such advanced biological functions to engineering applications are used in a wide range of fields, such as the synthesis of chemicals and pharmaceuticals, and environmental protection. In this lecture, we aim to learn about examples of applied engineering research, with a chemical understanding of life systems at the molecular level, centering on the function of enzyme molecules. The first stage explains the mechanism of the underlying gene and transcription / translation system, analysis methods, and further, taking bioplastics as an example, the molecular mechanism of biosynthesis and biodegradation, artificial modification technology of enzyme molecules, structure and Learn about function, how to measure activity, and the thermodynamic understanding of biological reactions. In the next stage, we aim to understand the modification and construction of the cell.

到達目標 Course Goals

Understand the mechanism and methodology for synthesizing and functioning the target protein (enzyme molecule) by genetic engineering. Understanding enzyme reactions and metabolism based on chemistry and thermodynamics. Understand the structure and function of enzyme molecules and their analysis methods. To understand how to use exhaustive analysis methods and data such as genomic DNA sequences, transcriptomes, proteomes and metabolome. Using these biological functions, we will learn the ideas for making them applicable to the field of engineering.

授業計画 Course Schedule

Microbial Engineering: metabolic engineering, enzymatic transformation

Protein Engineering: Protein mutagenesis, engineering

Genetic engineering: genetic modification, genome editing

Synthetic biology: molecular design, modeling

準備学習 (予習・復習)等の内容と分量 Homework

It is desirable to have at least 2 hours of preparation and 2 hours of review for each course content.

成績評価の基準と方法 Grading System

The minimum standard for evaluation is attendance rate 70% or more. The degree of achievement is comprehensively evaluated by the questions and answers during the lecture, and the contents of the report given as appropriate. Failure to submit a report will also not meet the evaluation criteria. The attendance of intensive lectures is essential to be credited.

他学部履修の条件 Other Faculty Requirements

テキスト・教科書 Textbooks

講義指定図書 Reading List

参照ホームページ Websites

研究室のホームページ Websites of Laboratory

전 미友 Assura Title	一卡田上版化学(用)	- 婚小伝(N学)「Alind Disphomiatury (Analy	·: 1.D:1)]				
科目名 Course Title 講義題目 Subtitle	心用生物16子(3	E物分析化学)[Applied Biochemistry (Analy	tical Biocnemistry)				
請報題日 Sublue 責任教員 Instructor	谷 博文「TANI I	Hirofumi] (大学院工学研究院)					
員任教員 Instructor 担当教員 Other Instructors	台 時入 [1AINI]	llrorum」(人子阮二子卯元元)					
科目種別 Course Type							
開講年度 Year	2024	時間割番号 Course Number	094307				
期間 Semester	Fall	単位数 Number of Credits	1				
授業形態 Type of Class	Lecture	对象年次 Year of Eligible Student	\sim				
対象学科・クラス Eligible Depa	1						
ナンバリングコード Numbering		CHEM ELBIO 6102					
補足事項 Other Information							
授業実施方式 Class Method		1 対面授業科目《対面のみ》					
キーワード Key Words							
-	assay Immunoass	ay, Biomolecular interaction, Analytical bioc	phemistry				
授業の目標 Course Objective		iy, Diomorecular metraction, r mary rear size					
		and applications, students will understand	d how in-vivo reactions such as				
		have sophisticated molecular recognition					
		andings, students will be able to construct					
depending on the target to be	measured when nec	essary to obtain substance information in th	neir future research.				
到達目標 Course Goals							
The goals of this course are to	be able to;						
		n biological and biochemical processes, a	nd the applications to analytical				
chemistry exploiting biomolecu	•						
- Design a suitable bioanalytic	al system for a targ	et molecule.					
授業計画 Course Schedule							
		in analytical chemistry: Chemical analysis, r					
and biochemical reactions, biomimetics, biochemical and biological analyses, selectivity and sensitivity, spectrophotometry,							
fluorometry, bioluminescence 2. Enzyme assay: Structure and activity of enzyme, kinetics, and equilibrium of enzyme reaction, assays for enzyme activity							
		enzymes, and equilibrium of enzyme real enzymes, and enzymatic cycling method	action, assays for enzyme activity				
		oody, antigen, hapten, epitope, immunopre	cipitation immuno-enzymometric				
assay, labels in immunoassay	fullor caction, and	ouy, antigen, napten, epitope, ministropie	compitation, minune enzymemetrie				
	lamental and type	of nucleic-acid hybridization, Detection te	echniques of nucleic acid probes,				
Analysis of nucleic acid sequer		, , , , , , , , , , , , , , , , , , ,	······································				
	· •	nts will be divided into multiple teams, and	then team discussion to propose				
new bioanalytical methods and							
準備学習 (予習・復習)等の内	容と分量 Homewor	k					
Students are expected to read	the handouts that	are given at least a week ahead. Students	are also requested to review each				
lecture and study the journal a							
	The total time for preparation and review is approximately 4 hours.						
成績評価の基準と方法 Gradir							
A comprehensive evaluation is based on the degree of achievement judged from the learning status and understanding of the							
analytical methods using/of in-vivo reactions. Specifically, the term-end report, presentation in the class, and the							
	contribution to the class (remarks in discussions, answers to question during class) will be assessed. 他学部履修の条件 Other Faculty Requirements						
テキスト・教科書 Textbooks							
フェスト・波科書 Textbooks テキストは指定せず, 適宜 Lecture 資料を配布する。その他, 参考となる文献を適宜紹介する。							
Not specify texts. Handouts will be distributed. In addition, reference documents will be introduced as appropriate.							
講義指定図書 Reading List							
参照ホームページ Websites							
研究室のホームページ Websit	tes of Laboratory						

It is advisable to master biochemistry, analytical chemistry, and instrumental analysis in advance.

講教題目 Subitie 責任教員 Instructor 満任教員 Cher Instructor 描述のCourse Type 相目ない Course Type 相関いの Mitsue (工学研究院)、ISHIDA Akhiko (工学研究院)、 HIBINO Mitsue (工学研究院) 利用 Senseter Fall 単位教 Number of Credits 2 とeture 対象年文 Year of Eligible Student ~ サガタギ科・クラス Eligible Department/Olass フィン(リングー)	科目名 Course Title	古田生物化学	A (マイカロシステト化学) [Applied Pieche	mistry A (Migrogystom Chomistry)]	
算任教員 Instructor 選要人 第一日 2目当教員 Other Instructors MAEKI Masstoniii (二学研究院), ISHDA Akhiko (二学研究院), 新聞書生、Year 2024 時間割雪号 Course Number 094308 期間Senester Feld 単位教 Number of Oradits 2 資素が癒 Type of Class Leture 対象年次 Year of Eligible Student ~ 対象学科・クラス Eligible Department/Class ガス Akhiko CLE Particle Student ~ オレンジノブード Numbering Code CHEM ELBIO 6112 ~ ポロスティット Kow Worda A ~ 新していていなり anapysis system. Microfluidic device, Microanalytical device, Micro medical diagnostic device オーフード Kow Worda A MEET devices and their application to bichemical analysis and medical diagnostic applications. In addition, acquire the latest knowledge and bicas regarding the development on increanalytical devices and their application to bichemical analysis and medical diagnostic applications. In addition, acquire the latest knowledge and bicases regarding the development on increanalytical devices and their application to bichemical analysis and medical diagnostic application to bicatione and stude application to bichemical analysis and medical diagnostic application to bicatione application to bichemical analysis and medical analysis. The gable of thic curves are to be able		心历生初化于	A(()) L) A L +) [Applied Bloche	emistry A (Microsystem Chemistry)	
相当教員 Other Instructore MAEKI Mesatoski (工学研究院), ISHIDA Akhiko (工学研究院), HIBINO Mitsue (工学研究院), SHIDA Akhiko (工学研究院), HIBINO Mitsue (T学研究院), HIBINO Mitsue (TPGM?), HIBINO MITSUE (HIBINO MITSUE), HIBINO MITSUE,		海南海 兴 [五			
IHBNO Misue (工学研究院) 利目違力(Course Type) 開業年度 Year 2024 時間報告 Course Number 094308 別目 Semester Fall 単位数 Number of Gredits 2 対象学科 グラス Eligible Department/Class 2 2 2 プタネット Thumbering Code CHEM_ELBIO 6112 ~ ~ 横足鼻洞 Other Information 1 3/m E 案内 7 This course Sull understand the principles of microfluidic device, Micro medical diagnostic device 受用の一下 Key Words Micro total analysis system, Microfluidic device, Microanaltysical device, Micro medical diagnostic applications. In addition, acquire the latest, knowledge and ledues regarding the development on microandytical devices and their application to blochemical analysis and medical diagnosts. Through these, it becomes possible to contract an appropriate measurement system according to the measurement target. ##### Course Coals The goals of this course are to be able to; - Fighting Hourse Schedule - Explain the fundamentals and techniques of the microfluidic devices for drug design and therapy. ###### Course Coals - Segan attraction analysis using microdevices: - - - Separation analysis using microdevices: - - - Separation analysis using microdevices					
科目観別 Course Type 開課年度 Year 2 024 時間割番号 Course Number 994308 301 Senseter Fall 単位数 Number of Credits 2 2 2 2 2 2 2 2	但当教員 Other Instructors				
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期間 Senester Full 中位数 Number of Credits 2 浸泉形態 Type of Class Lecture 対象年次 Year of Eligible Student ~ ~		2024	時間割悉早 Course Number	00/308	
授業移動 Type of Class Lecture 対象年次 Year of Eligible Student ~ 分類 学科・プラス Eligible Department/Class デンパシブラード Numbering Code CHEN_ELBIO 6112 横足事項 Other Information 授業実施方式 Class Method 1 対面技業科目《対面のみか) オーワード Key Words Micro trait analysis system, Microfituidic device, Microanaltyical device, Micro medical diagnostic device 授業の目職 Course Objectives This course will understand the principles of microfluidic device development and biochemical analysis, drug discovery an medical diagnostic applications. In addition, acquire the latest knowledge and idease regarding the development on microanalytical devices and their application to biochemical analysis and medical diagnosis. Through these, it becomes possibl to construct an appropriate measurement system according to the measurement target. 新算目長 Course Objectives The goals of this course are to be able to; Explain the fundamentals and techniques of the microfluidic devices for biochemical analyses. Design a suitable micro analysis system for a target molecule. Explain the fundamentals and techniques of the microfluidic devices for drug design and therapy. 授業計画 Course Schedue This course will be held twice a week by multiple lecturers. 1. Concept of analysis using microdevices: immunoussay, circulating tumor cells, cell—free DNA S. Separation analysis and microdevices 1. Spatiant double-cubes, particle separation method and devices 5. Poper-based analytical device S. Poper-based analytical separation system etarge forg forg diagnophysic Physical Betweenet Sudents are expected to reade the fundamentals support the separation system Explain Biochemical System Concept Gas Method System Concept Gas Gas Ophysics of Laboratory Kupteredin Gas Gas Ophysics of Laboratory Kupteredin Gas Gas Cond					
対象学科・グラス Eligible Department/Class ナンパリングコード Numbering Code イロスリングコード Numbering Code イオーワード Key Worls Micro total analysis system, Microfluidic device, Microanaltyteal device, Micro medical diagnostic device 授東の目標 Course Objectives This course will understand the principles of microfluidic device development and biochemical analysis, drug discovery an medical diagnostic applications. In addition, acquire the latest knowledge and ideas regarding the development o microanalytical devices and their application to biochemical analysis and medical diagnosis. Through these, it becomes possible to construct an appropriate measurement system according to the measurement target. 3) all ff Course Ocals The goals of this course are to be able to; - Explain the fundamentals and techniques of the microdevices for biochemical and biomedical analyses. - Design a suitable micro analysis system for a target molecule. - Explain the fundamentals and techniques of the microfluidic devices for drug design and therapy. 授果計圖 Course Schedule This course will be held twice a week by multiple lecturers. 1. Concept of analysis using microdevices: immunoassay, circulating tumor cells, cell=free DNA 3. Separation analysis of biomolecules, particle separation method and devices 5. Paper-based analytical device 5. Paper-based analytical devices imicrodroplet, nanoparticles, drug delivery system and genome editing structure analysis of biomolecules, particle separation method and devices 5. Paper-based analytical systems 8. Portable analytical systems and wearable sensing systems ###37 (75 (36) %00,007 ab) ab-000 					
			为家中次 Tear of Eligible Student		
Reparence of the information			CLIEM EL DIO 6112		
技術実施方式 Class Method		Code			
 キーワード Key Words Micro total analysis system, Microfluidic device, Microanaltyical device, Micro medical diagnostic device			1 対面授業科目(対面のみ)		
Micro total analysis system, Microfluidic device, Microanaltyical device, Micro medical diagnostic device			1 从面段来有百《从面9999/		
接来の目標 Course Objectives This course will understand the principles of microfluidic device development and biochemical analysis, drug discovery and medical diagnostic applications. In addition, acquire the latest knowledge and ideas regarding the development o microanalytical devices and their application to biochemical analysis and medical diagnosis. Through these, it becomes possible to construct an appropriate measurement system according to the measurement target. JJzelf & Course Goals The goals of this course are to be able to: - Explain the fundamentals and techniques of the microfluidic devices for biochemical and biomedical analyses Design a suitable micro analysis system for a target molecule Explain the fundamentals and techniques of the microfluidic devices for drug design and therapy. 授業計画 Course Schedule This course will be held twice a week by multiple lecturers. 1. Concept of analysis using microdevices 2. Blood analysis system using microdevices 3. Separation analysis using microdevices 4. Drug design and therapy using microfluidic devices: microfluidic, durices 5. Paper-based analytical device 6. Microfluidic-based separation system 7. Electrochemical biosensors 8. Portable analytical systems and wearable sensing systems ####21 (P8 'd83)%onget/d9 Homework Students are expected to read the handouts that are given at least in a week ahead. Students are also requested to revieve ach lecture and study the journal articles quoted in the lecture. d###able for therefores: prr> Alphe Textbooks prr> r> r , r	-	icrofluidic device	e. Microanaltyical device. Micro medical d	liagnostic device	
This course will understand the principles of microfluidic device development and biochemical analysis, drug discovery and medical diagnostic applications. In addition, acquire the latest knowledge and ideas regarding the development on incroanalytical devices and their application to biochemical analysis and medical diagnosis. Through these, it becomes possible to construct an appropriate measurement system according to the measurement target. JB2E1# Course Goals The goals of this course are to be able to; - Explain the fundamentals and techniques of the microfluidic devices for drug design and therapy. 授業計画 Course Schedule This course will be held twice a week by multiple lecturers. 1. Concept of analysis using microdevices: 2. Blood analysis using microdevices: immunoassay, circulating tumor cells, cell-free DNA 3. Separation analysis using microdevices 4. Drug design and therapy using microfluidic devices: microfluidic nanoparticles, drug delivery system and genome editing structure analysis of biomolecules, particle separation method and devices 5. Paper-based analytical device 6. Microfluidic-based separation system 7. Electrochemical biosensors 8. Portable analytical systems and wearable sensing systems ###2 (75 d*2) **Ony OR2 **Onework Students are expected to read the handouts that are given at least in a week ahead. Students are also requested to review each lecture and study the journal articles quoted in the lecture. Zphaty Betwoods Cyt-thate Textbooks Cyt-			,		
medical diagnostic applications. In addition, acquire the latest knowledge and ideas regarding the development o microanalytical devices and their application to biochemical analysis and medical diagnosis. Through these, it becomes possible to construct an appropriate measurement system according to the measurement target. JJE1# Course Goals The goals of this course are to be able to; - Explain the fundamentals and techniques of the microfluidic devices for drug design and therapy. JZ#shim the fundamentals and techniques of the microfluidic devices for drug design and therapy. JZ#shim the fundamentals and techniques of the microfluidic devices for drug design and therapy. JZ#shim the fundamentals and techniques of the microfluidic devices for drug design and therapy. JZ#shim the fundamentals and techniques of the microfluidic devices for drug design and therapy. JZ#shim the fundamentals and techniques of the microfluidic devices for drug design and therapy. JZ#shim the fundamentals and techniques of the microfluidic devices for drug design and therapy. JZ shim the fundamentals and techniques of the microfluidic devices: 1. Concept of analysis using microdevices 2. Blood analysis using microfluidic devices: microfloplet, nanoparticles, drug delivery system and genome editing structure analysis of biomolecules, particle separation method and devices 5. Paper-based analytical device 6. Microfluidic-based separation system 7. Electrochemical biosensors 8. Portable analytical systems and wearable sensing systems J#dp=10 (79 : d2) S@mopde2::dy Hornework Students are expected to read the handouts that are given at least in a week ahead. Students are also requested to review ach leture and study the journal articles quoted in the lecture. d#faff@d2#cb2 fcrading System Learning attitude and report Mz#faff@d2 = Reading List Spt Art.HaffEct : f , aig Lecture ğaf4e配ft-fo_8, &coft, seq5e&c3cyttkeaigtaft-fo_8, Not specify texts. Handouts will be distributed. In addition,	·· · · · ·		microfluidic device development and bio	chemical analysis, drug discoverv and	
microanalytical devices and their application to biochemical analysis and medical diagnosis. Through these, it becomes possible to construct an appropriate measurement system according to the measurement target. Jjäelf KOurse Goals The goals of this course are to be able to; - Explain the fundamentals and techniques of the microfluidic devices for biochemical and biomedical analyses Design a suitable micro analysis system for a target molecule Explain the fundamentals and techniques of the microfluidic devices for drug design and therapy. Ky Sthill Course Schedule This course will be held twice a week by multiple lecturers Concept of analysis using microdevices: immunoassay, circulating tumor cells, cell-free DNA - Separation analysis using microfluidic devices: microdroplet, nanoparticles, drug delivery system and genome editing structure analysis of biomolecules, particle separation method and devices - S. Paper-based analytical device - Microfluidic-based separation system - Electrochemical biosensors - Separation analytical systems and wearable sensing systems - Weither are alstudy the journal articles quoted in the lecture. Rd###@1678*4273; Grading System Learning attitude and report - dte*#atgewosch Cher Faculty Requirements - 5+27-1: %Atf = Textbooks - 7+27-1: %Desites - Microfluid: - Chips.jp/en/ - Microfluid: - Micro					
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到達目様 Course Goals The goals of this course are to be able to; Explain the fundamentals and techniques of the microdevices for biochemical and biomedical analyses. Design a suitable micro analysis system for a target molecule. Explain the fundamentals and techniques of the microfluidic devices for drug design and therapy. 提素計画 Course Schedule This course will be held twice a week by multiple lecturers. Decomposition of the analysis using microdevices? Blood analysis system using microdevices? Blood analysis using microfluidic devices: microdroplet, nanoparticles, drug delivery system and genome editing structure analysis of biomolecules, particle separation method and devices Faper-based analytical device Blootforthuidic-based separation system Reforbluidic-based to read the handouts that are given at least in a week ahead. Students are also requested to review each lecture and study the journal articles quoted in the lecture. Ref # ##@foff CF3 · {#1} * Goaly System Learning attitude and report Md*#Abditodits (that are given at least in a week ahead. Students are also requested to review each lecture and study the journal articles quoted in the lecture. Ref # CF3 · {#1} * Goaly System Learning attitude and report Md*#Abditodite Students are given at least in aweek ahead. Students are also requested to review each lecture of the distributed. In addition, reference documents will be introduced as appropriate. ###ECT * : *At * Textbooks * *At * # Textbooks * *At * * * * * * * * * * * * * * * * * 					
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接葉計画 Course Schedule This course will be held twice a week by multiple lecturers. 1. Concept of analysis using microdevices 2. Blood analysis system using microdevices: immunoassay, circulating tumor cells, cell-free DNA 3. Separation analysis using microdevices 4. Drug design and therapy using microdidic devices: microdroplet, nanoparticles, drug delivery system and genome editing structure analysis of biomolecules, particle separation method and devices 5. Paper-based analytical device 6. Microfluidic-based separation system 7. Electrochemical biosensors 8. Portable analytical device biomework Students are expected to read the handouts that are given at least in a week ahead. Students are also requested to review each lecture and study the journal articles quoted in the lecture. nd#評価の基準と方法 Grading System Learning attitude and report dm学部履修の条件 Other Faculty Requirements 7年スト・教科書 Textbooks 7卡スト・教科書 Textbooks 7卡スト・教科書 Textbooks 7卡スト・社指定せず, 適宜 Lecture 資料を配布する。その他, 参考となる文献を適宜紹介する。 Not specify texts. Handouts will be distributed. In addition, reference documents will be introduced as appropriate. mis#指定図書 Reading List 8 Ørsezoホームページ Websites of Laboratory https://microfluidic.chips.jp/en/ 備考 Additional Information				d therapy	
This course will be held twice a week by multiple lecturers. 1. Concept of analysis using microdevices 2. Blood analysis system using microdevices: immunoassay, circulating tumor cells, cell-free DNA 3. Separation analysis using microdevices 4. Drug design and therapy using microfluidic devices: microdroplet, nanoparticles, drug delivery system and genome editing structure analysis of biomolecules, particle separation method and devices 5. Paper-based analytical device 6. Microfluidic-based separation system 7. Electrochemical biosensors 8. Portable analytical systems and wearable sensing systems 準備学習(予習・復習)等の内容と分量 Homework Students are expected to read the handouts that are given at least in a week ahead. Students are also requested to review each lecture and study the journal articles quoted in the lecture. 成績評価の基準と方法 Grading System Learning attitude and report 他学部履修の条件 Other Faculty Requirements テキスト・教科書 Textbooks アキストは指定せず, 適宜 Lecture 資料を配布する。その他、参考となる文献を適宜紹介する。 Not specify texts. Handouts will be distributed. In addition, reference documents will be introduced as appropriate. 講講指定図書 Reading List 参照ホームページ Websites of Laboratory https://microfluidic.etips.p/en/ 備考 Additional Information		a teeninques of t	the interonatione devices for and design an		
1. Concept of analysis using microdevices 2. Blood analysis system using microdevices: immunoassay, circulating tumor cells, cell-free DNA 3. Separation analysis using microdevices: microdroplet, nanoparticles, drug delivery system and genome editing structure analysis of biomolecules, particle separation method and devices 5. Paper-based analytical device 6. Microfluidic-based separation system 7. Electrochemical biosensors 8. Portable analytical systems and wearable sensing systems 準備学習(予習・復習)等の内容と分量 Homework Students are expected to read the handouts that are given at least in a week ahead. Students are also requested to review each lecture and study the journal articles quoted in the lecture. 成績評価の基準と方法 Grading System Learning attitude and report 他学部履修の条件 Other Faculty Requirements 7+スト・教科書 Textbooks テキスト・教科書 Textbooks テキストは指定せず, 適宜 Lecture 資料を配布する。その他,参考となる文献を適宜紹介する。 Not specify texts, Handouts will be distributed. In addition, reference documents will be introduced as appropriate. 講義指定図書 Reading List 参照ホームページ Websites 研究室のホームページ Websites of Laboratory https://microfluidic.chips.jp/en/ 備考 Additional Information		1- 11+:	1- 1+		
Students are expected to read the handouts that are given at least in a week ahead. Students are also requested to review each lecture and study the journal articles quoted in the lecture.	 Drug design and therapy us structure analysis of biomolecu Paper-based analytical devi Microfluidic-based separation Electrochemical biosensors Portable analytical systems 	sing microfluidic iles, particle sep ce on system and wearable ser	aration method and devices	g delivery system and genome editing	
each lecture and study the journal articles quoted in the lecture. 成績評価の基準と方法 Grading System Learning attitude and report 他学部履修の条件 Other Faculty Requirements ラキスト・教科書 Textbooks テキストは指定せず、適宜 Lecture 資料を配布する。その他、参考となる文献を適宜紹介する。 Not specify texts. Handouts will be distributed. In addition, reference documents will be introduced as appropriate. 講義指定図書 Reading List 参照ホームページ Websites 研究室のホームページ Websites of Laboratory https://microfluidic.chips.jp/en/ 備考 Additional Information				Cturdents and also respected to accion	
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Learning attitude and report 他学部履修の条件 Other Faculty Requirements テキスト・教科書 Textbooks テキストは指定せず,適宜 Lecture 資料を配布する。その他,参考となる文献を適宜紹介する。 Not specify texts. Handouts will be distributed. In addition, reference documents will be introduced as appropriate. 講義指定図書 Reading List 参照ホームページ Websites 研究室のホームページ Websites of Laboratory https://microfluidic.chips.jp/en/ 備考 Additional Information					
他学部履修の条件 Other Faculty Requirements テキスト・教科書 Textbooks テキストは指定せず,適宜 Lecture 資料を配布する。その他,参考となる文献を適宜紹介する。 Not specify texts. Handouts will be distributed. In addition, reference documents will be introduced as appropriate. 講義指定図書 Reading List 参照ホームページ Websites 研究室のホームページ Websites of Laboratory https://microfluidic.chips.jp/en/ 備考 Additional Information		ig Oystein			
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講義指定図書 Reading List 参照ホームページ Websites 研究室のホームページ Websites of Laboratory https://microfluidic.chips.jp/en/ 備考 Additional Information					
研究室のホームページ Websites of Laboratory https://microfluidic.chips.jp/en/ 備考 Additional Information	講義指定図書 Reading List				
https://microfluidic.chips.jp/en/ 備考 Additional Information	参照ホームページ Websites				
https://microfluidic.chips.jp/en/ 備考 Additional Information	研究室のホームページ Wabai	tes of laborator	M		
備考 Additional Information			У		
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IN A ADVISAGE OF MAXIMUM DISCOUNTING ADALYTICAL PODDUSTRY (ADD DECEMBER 9 909/VEIC IN 90/90/04		mietre endette	al abomistry and instrumental analysis in	advance	

科目名 Course Title	応用生物化学	A(機能性高分子特論)[Applied Bi	ochemistry A (Advanced Functional
	Polymer)]	r (band r let) i i li lim / li ibbiod Di	
講義題目 Subtitle			
責任教員 Instructor	佐藤 敏文「SΔT	OH Toshifumi] (大学院工学研究院)	
担当教員 Other Instructors	· · · · · · · · · · · · · · · · · · ·		なです~
	YAMAMOTO Ta	kuya (工学研究院), LI FENG (工学研9	1.1%し
科目種別 Course Type			
開講年度 Year	2024	時間割番号 Course Number	094309
期間 Semester	Spring/Summer	単位数 Number of Credits	2
授業形態 Type of Class	Lecture	対象年次 Year of Eligible Student	~
対象学科・クラス Eligible Dep	artment/Class		
ナンバリングコード Numbering Code		CHEM_ELBIO 6111	
補足事項 Other Information			
授業実施方式 Class Method 1 対面授業科目《対面のみ》			
キーロード Key Words			

Polymer synthesis, Precise polymerization, Controlled/Living polymerization, Radical polymerization, Cationic polymerization, Anionic polymerization, Coordination polymerization, Functional polymer, Polymer structure, Polymer design, Polymer solution, Phase separation behavior

授業の目標 Course Objectives

To utilize polymeric materials and to design new functionality, the methodology of the polymer synthesis must be understood. To learn various polymerization-methods and the various polymerization mechanisms is mainly studied as the basics of the macromolecular synthesis, and to understand the latest macromolecular synthesis method is a goal. Moreover, students study and understand the precise syntheses based on the free-radical polymerization, the cationic polymerization, the anionic-polymerization, and coordination polymerization, which is used for the design and synthesis of functional macromolecules.

到達目標 Course Goals

Our goals are to learn various polymerization-methods and the various polymerization mechanisms and to understand the latest macromolecular synthetic method. Moreover, students learn the precise synthesis based on the living polymerization, which is used for the design and synthesis of functional polymeric materials as a goal of this course.

授業計画 Course Schedule

- 1. Polymerizations by Zieglar–Natta.
- 2. Polymerizations by metallocene catalysts.
- 3. Metathesis polymerization
- 4. Ring-opening polymerization leading to environment-conscious polymer and medical polymer
- 5. polycondensation and chain-growth polycondensation leading to engineering plastic and electrofunctional polymer
- 6. Radical polymerization: characteristics of radical polymerization and the primary structure of the resulting polymers.
- 7. Anionic polymerization: characteristics of anionic polymerization and the primary structure of the resulting polymers.
- 8. Cationic polymerization: characteristics of cationic polymerization and the primary structure of the resulting polymers.
- 9. Functional materials by assembly of polymers with designed architectures.

準備学習 (予習・復習)等の内容と分量 Homework

Students are required to carefully read distributed handouts (30 min), if any, beforehand and submit reports for assigned problems by specified dates (30 min). Also, students present a report for problems after the class ends (30 min).

成績評価の基準と方法 Grading System

In principle, students who attend 70% or more classes are graded. The final grade is determined by his/her learning attitude (20%) and reports (80%). The reports are evaluated based on the student's understanding on the synthesis and design of polymers and the logic of the writing. A, 100-90; B, 89-80; C, 79-70; D, 69-60; F, ≤ 60 .

テキスト・教科書 Textbooks

特に指定はしないが,「高分子合成化学」(大津隆行著,化学同人)と「大学院高分子科学」(野瀬卓平,中濱精一,宮田清蔵編,講談社サイエンティフィク)を参考にしていただきたい。The documents will be distributed.

講義指定図書 Reading List

大学院 高分子科学/野瀬卓平・中浜精一・宮田清蔵:講談社サイエンティフィック, 2000

参照ホームページ Websites

研究室のホームページ Websites of Laboratory

https://poly-ac.eng.hokudai.ac.jp/index_e.html

https://cma.eng.hokudai.ac.jp/

備考 Additional Information

The class is opened by face-to-face. Please carefully see ELMS.

科目名 Course Title	総合化学研究	先端講義[Internship]	
講義題目 Subtitle	心口口子如刀	Jumm 我[Internship]	
責任教員 Instructor	汕北 人典 [56	NBOKU Hisanori] (大学院工学研究院)	
担当教員 Other Instructors			
科目種別 Course Type			
開講年度 Year	2024	時間割番号 Course Number	094401
期間 Semester	Fall	単位数 Number of Credits	1
授業形態 Type of Class	Internship	対象年次 Year of Eligible Student	~
対象学科・クラス Eligible Depar			
ナンバリングコード Numbering	Code	CHEM_ELCOM 6212	
補足事項 Other Information			
授業実施方式 Class Method		1 対面授業科目《対面のみ》	
キーワード Key Words			
Internship (domestic and overse	ea)		
授業の目標 Course Objectives	3		
Students improve their skill and	l knowledge by b	peing engaged in an actual work relating th	eir future career.
For overseas internship, stud	ents develop g	lobal vision by their experience overse	as, gain expertise and experimental
techniques which seem to be ha	ard to obtain in J	apan.	
到達目標 Course Goals			
Students start to contact with	where to do in	nternship, then improve skills of commun	ication, language, research practice,
research network and community	ty formation etc,	so that they can raise consciousness as a	n engineer or a researcher.
For overseas internship, stude	ents should try	not to keep the experience at only leve	el of basic studies, try to apply the
experience to collaborative rese	earches with a p	ractical level in the future.	
授業計画 Course Schedule			
The program will be generally c	onducted followi	ng the schedule below.	
1. Announcement			
2. Application (not equal to Re	gistration)		
3. Preparation			
4. Internship for about between	n two weeks and	two months	
5. Submission of a report for the	ie internship, pro	esentation	
準備学習 (予習・復習)等の内羽	容と分量 Homew	ork	
Students need to do preliminary	y search and to p	prepare ecperiments in advance.	
成績評価の基準と方法 Gradin	g System		
Basically, students must submit	a report and do	a presentation (in English language for ov	verseas internship).
They will be evaluated by the a			
他学部履修の条件 Other Facu	ulty Requirement	ts	
テキスト・教科書 Textbooks			
使用しない			
講義指定図書 Reading List			
使用しない			
参照ホームページ Websites			
研究室のホームページ Websit	es of Laboratory	1	
備考 Additional Information			

科目名 Course Title	化学特別講義[Advanced Chemistry]			
講義題目 Subtitle	物理化学特別講義 2024[Physical Chemistry 2024]			
責任教員 Instructor	村越 敬 [MURAKOSHI Kei] (大学院理学研究院)			
担当教員 Other Instructors	羽馬 哲也(東	京大学)		
科目種別 Course Type				
開講年度 Year	2024	時間割番号 Course Number	094411	
期間 Semester	Intensive	単位数 Number of Credits	1	
授業形態 Type of Class	Lecture	対象年次 Year of Eligible Student	~	
対象学科・クラス Eligible Depa	rtment/Class			
ナンバリングコード Numbering	Code	CHEM_ELCOM 6400		
補足事項 Other Information				
授業実施方式 Class Method		1 対面授業科目《対面のみ》		
キーワード Key Words				
Interstellar chemistry, chemica	l kinetics, ion-m	olecule reaction, surface science, quantum	tunneling	
授業の目標 Course Objective	S			
		ses occurring in interstellar space from the	-	
reaction dynamics. In addition	to observational	and theoretical studies, some key laborator	y experiments are also reviewed.	
到達目標 Course Goals				
		n interdisciplinary science covering astror		
	ience, and that	approaches from each discipline are nee	eded to understand the meaning of	
astrochemical phenomena.				
授業計画 Course Schedule	1 .			
1. Introduction to interstellar chemistry				
	 The formation of hydrogen molecules in the early universe Langevin rate coefficient in ion-molecule reactions 			
0				
 Why are laboratory experim Quantum tunneling in chemi 		or studying chemical kinetics?		
6. Dust surface chemistry	.501 y			
7. Future perspective on inters	stellar chemistry			
準備学習(予習・復習)等の内		ork		
		m Chemistry textbooks at undergraduate le	vel is highly recommended.	
成績評価の基準と方法 Gradir				
Comprehensively evaluate cou	rse status includi	ng attendance and report.		
他学部履修の条件 Other Fac	ulty Requirement	ts		
テキスト・教科書 Textbooks				
講義指定図書 Reading List				
参照ホームページ Websites				
研究室のホームページ Websit	-			
http://www.hamalab.c.u-tokyc	.ac.jp/research-	english-ver/		
備考 Additional Information				

科目名 Course Title	化学特別講義[Advanced Chemistry]			
講義題目 Subtitle	無機分析化学特別講義 2024[Inorganic and Analytical Chemistry 2024]			
責任教員 Instructor	上野 貢生 [UENO Kosei] (大学院理学研究院)			
担当教員 Other Instructors	坪井 泰之 (大阪	公立大字)		
科目種別 Course Type		1		
開講年度 Year	2024	時間割番号 Course Number	094412	
期間 Semester	Intensive	単位数 Number of Credits	1	
授業形態 Type of Class	Lecture	対象年次 Year of Eligible Student	\sim	
対象学科・クラス Eligible Depa				
ナンバリングコード Numbering	Code	CHEM_ELCOM 6402		
補足事項 Other Information				
授業実施方式 Class Method		1 対面授業科目《対面のみ》		
キーワード Key Words				
Semiconductor, Materials for e	lectronics, Photoch	emistry, Laser, Plasmonics, Nanomateri	als, Photomedicine	
授業の目標 Course Objective	S			
Recent remarkable advances o	of science and tech	nology are due to the explosive develo	pment of semiconductor engineering.	
The semiconductor engineerin	ig should be closel	y related to chemistry, especially inor	rganic chemistry. Actually, the main	
business of major chemical ir	dustries has a clo	se relation with the conductor engine	ering in Japan. In this meaning the	
chemistry of semiconductors w	vill presumably cent	ered in the field of inorganic chemistry	. On the other hand, nanotechnology	
has been a powerful driving fo	orce of the develop	ment of semiconductor engineering an	d their current and future promising	
applications should be "photon	ics". In the lecture	, the basics and modern/future applicat	ions of nanotechnology and photonics	
will be explained from the view	point of inorganic cl	hemistry.		
到達目標 Course Goals				
Understanding the subjects no	ted below			
1) Fundamentals of light-matte	er interactions			
2) Inorganic chemistry of semic	conducting materials	3		
3) Photonic functions of molect	ular / inorganic mat	erials		
4) Photonics control of Function	onal Materials			
5) Photonic measurements of F	Functional Materials			
6) Photonics functions based o				
7) Modern Applications at the	state-of-art			
授業計画 Course Schedule				
1) Introduction of semiconduct		chemistry		
2) Concepts of nanotechnology	-			
3) Fundamentals of light-matter interactions				
4) Photonics function, control, and measurement of Functional Materials				
5) Photonics functions based on nanostructures				
6) Modern Applications at the	state-of-art			
準備学習 (予習・復習)等の内		< c		
Homework will be handed out in the class.				
成績評価の基準と方法 Grading System Reports (homowark, 80) and Ovin (20%)				
Reports (homework, 80) and Quiz (20 %) 他学部履修の条件 Other Faculty Requirements				
୲ଌ୕ୢ୷୴୶ୡ୲୭୰୵୕୕୕ୖ୷ୖ୲୕ୖ୰ୖୖୖୖୖ୲୲ଌୖ	arcy riequirements			
テキスト・教科書 Textbooks				
ノイハIT 我作者 Textbooks				
講義指定図書 Reading List				
参照ホームページ Websites				
研究室のホームページ Websites of Laboratory				
借書 Additional Informatio				
備考 Additional Information				

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科目名 Course Title		lvanced Chemistry]		
講義題目 Subtitle	有機化学特別講義 2024[Organic Chemistry 2024]			
責任教員 Instructor	澤村 正也 [SAWAMURA Masaya] (大学院理学研究院)			
担当教員 Other Instructors	野崎 京子 (東京)	大字)		
科目種別 Course Type				
開講年度 Year	2024	時間割番号 Course Number	094413	
期間 Semester	Intensive	単位数 Number of Credits	1	
授業形態 Type of Class	Lecture	対象年次 Year of Eligible Student	~	
対象学科・クラス Eligible Depa				
ナンバリングコード Numbering	Gode	CHEM_ELCOM 6400		
補足事項 Other Information		↓ ↓ ナ ☆ 歩 イ \ □ //↓ ↓ ナ ☆ ぇ \ \		
授業実施方式 Class Method		1 対面授業科目《対面のみ》		
キーワード Key Words				
		emistry, Polymer Synthesis, Organic Synt	hesis, Green Chemistry	
授業の目標 Course Objective				
		ion by participating in the breaking and that the presenter is working on and the	0	
two perspectives: (1) social pro		-	development of catalysts for it from	
		producing useful substances from fossil	resources. However, when we think	
		y unused carbon resources. In this lectu		
		ely utilize unused carbon resources ar	-	
environmental impact.			a accord new materials with low	
	olecules with uniqu	e structures may exhibit physical prope	erties due to special orbitals. These	
	-	ing able to appreciate them is a privilege	_	
		g "useful" and consider its significance.		
到達目標 Course Goals				
The goal of this lecture is to u	nderstand the chara	acteristics of transition metal catalysts a	nd typical elements, and acquire the	
ability to think integrative, ta	ke on challenges, a	and come up with ideas that will contril	oute to the development of organic	
chemistry.				
授業計画 Course Schedule				
1. Catalyst development for eff		d carbon resources		
2. Aiming at creating new susta				
3. The appeal of organic molecules with unique structures				
準備学習(予習・復習)等の内				
Preparation is not necessary, but basic knowledge of transition metal catalysts and organic synthetic chemistry is desirable.				
Assign a report assignment related to the lecture content.				
成績評価の基準と方法 Grading System Attendance status and report assignments will be comprehensively evaluated.				
他学部履修の条件 Other Faculty Requirements				
テキスト・教科書 Textbooks				
当日に資料を配布する。				
Materials will be distributed on the day.				
講義指定図書 Reading List				
参照ホームページ Websites				
研究室のホームページ Websites of Laboratory				
https://park.itc.u-tokyo.ac.jp/	/nozakilab/indexE.h	ntml		
備考 Additional Information				

科目名 Course Title	化学特別講義[Advanced Chemistry]		
講義題目 Subtitle	生物化学特別講義 2024[Biochemistry 2024]		
責任教員 Instructor	坂口 和靖 [SAKAGUCHI Kazuyasu] (大学院理学研究院)		
担当教員 Other Instructors	後藤 佑樹 (京都大学)		
科目種別 Course Type			
開講年度 Year	2024	時間割番号 Course Number	094414
期間 Semester	Intensive	単位数 Number of Credits	1
授業形態 Type of Class	Lecture	対象年次 Year of Eligible Student	~
対象学科・クラス Eligible Department/Class			
ナンバリングコード Numbering Code		CHEM_ELCOM 6400	
補足事項 Other Information			
授業実施方式 Class Method		1 対面授業科目《対面のみ》	
キーワード Key Words			
Biological Chemistry, Chemical Biology, Peptide Chemistry, Biosynthesis Engineering, Ribosomal Synthesis			

授業の目標 Course Objectives

In living organisms, ribosomes drive the translation reaction, in which proteins (polypeptide chains) are produced using genetic information as a template. Considering the translation reaction as a synthetic system of chemical compounds, it has many advantages over other synthetic systems. Therefore, there have been many studies to engineer the translation reaction as a synthetic tool to produce artificial peptides. Moreover, by combining with selection techniques such as phage display, ribosome display, and mRNA display, it has been also utilized for the development of functional peptides. In this course, after reviewing the mechanism of translation reaction, we will discuss examples of its engineering and applications, highlighting the significance and potential of chemical biology based on ribosomal synthesis.

到達目標 Course Goals

- · To be able to explain the mechanism of prokaryotic translation reactions.
- To understand the significance of engineered ribosomal synthesis as synthetic tools.
- · To understand the significance of engineered ribosomal synthesis as compound discovery tools.

授業計画 Course Schedule

- 1. Basics of the translation reaction
- 2. Methodologies for the production of artificial proteins/peptides by translation
- 3. Methodologies for the development of functional peptides by engineered translation
- 4. Recent examples of the production of pseudo-natural peptides by in vitro engineered translation
- 5. Recent examples of the discovery of pseudo-natural peptides by in vitro selection

準備学習 (予習・復習)等の内容と分量 Homework

Review your previous knowledge about ribosomal synthesis and peptide chemistry.

成績評価の基準と方法 Grading System

You will be evaluated by active participation including investigation, consideration, and discussion (70%), and assignment on a specified topic (30%, mandatory).

他学部履修の条件 Other Faculty Requirements

テキスト・教科書 Textbooks

適宜、資料を配布する。 Hand out materials.

講義指定図書 Reading List

参照ホームページ Websites

研究室のホームページ Websites of Laboratory

科目名 Course Title	化学特別講義[Advanced Chemistry]		
講義題目 Subtitle		ト特別セミナー[Career Management Speci	ial Seminar
責任教員 Instructor	中富 晶子 [NAKATOMI Akiko] (大学院理学研究院)		
担当教員 Other Instructors	七澤 淳 (理学研究	先院	
科目種別 Course Type	0001		004415
開講年度 Year	2024	時間割番号 Course Number	094415
期間 Semester 授業形態 Type of Class	Intensive Lecture	単位数 Number of Credits 対象年次 Year of Eligible Student	\sim
投業が設てype of Class 対象学科・クラス Eligible Depar		对象十次 Tear of Eligible Student	
ナンバリングコード Numbering		CHEM_ELCOM 6400	
補足事項 Other Information	Oule	CHEM_ELCOM 0400	
授業実施方式 Class Method		1 対面授業科目《対面のみ》	
キーワード Key Words		1 对面及采行口《对面0,507//	
Company, Research and Develo	opmont Caroor Pati	h Management Education	
授業の目標 Course Objectives			
		society expects from its Ph.D.s.	
		uations which help set research themes.	
到達目標 Course Goals			
	ematical thinking w	ith a high level of expertise, students	will be actively involved in non-
specialized social issues.	0	Ç • • •	2
2. Students will connect social	issues with their ow	n specialties (strength), then incorporate	them into a "research theme".
3. Students will actively involve	themselves in grou	p discussions while considering other par	ticipants.
授業計画 Course Schedule			
Classes will be conducted in sm	ıall groups (generall	y around 5 students).	
1: Introduction of economics, p	atents, and corpora	ate research cases by visiting professor A	tsushi Nanasawa, who was active in
the private sector for many year			
		bblem identification and (2) suggest solution	
3: Individual interviews (special	lization / research s	situation / career preference) will be held	around the 1st and 2nd classes. A
follow up meeting will be held in	1 March 2025.		
_	_		
[Themes of lecture / Work shop] Total 6 classes, 120 min each			
		perational issue of group work and worksho	qc
2. What is a patent? / Environm			
3. Research cases of young emp		intelligence	
4. Research cases of veterans /		1	
5. Research cases of responsibl			
6. Summary of workshops / The	eme setting for Com	ipany Consortium	
			11.11
÷ .	-	and the 20th of each month, the dates dec	
		out online (hybrid) participation is also po	ssible depending on the situation.
Slack and Miro online whiteboa 進備学習 (系習・復習)等の内容		~	
準備学習 (予習・復習)等の内容と分量 Homework Choose and read one book on each designated lecture theme in advance so that you can discuss it.			
成績評価の基準と方法 Grading System			
Students will be evaluated by i) attitude toward learning (20%), ii) status of homework efforts (degree of information collection			
and understanding) (30%), iii) contribution to group discussions and group work in each class (e.g., active participation and			
quality of comments) (30%), iv) quality of reports and other submissions (20%).			
他学部履修の条件 Other Faculty Requirements			
		grams and fellowships, and this may limit t	he available places.
テキスト・教科書 Textbooks			
講義指定図書 Reading List			
参照ホームページ Websites			
https://phdiscover.jp/alp/			
https://sites.google.com/eis.hokudai.ac.jp/dxphd-fellow/home			
https://sites.google.com/elms.hokudai.ac.jp/ambitious-phd-fellow/home			
研究室のホームページ Websites of Laboratory			

備考 Additional Information

 \cdot For convenience, this course is offered as a subject at the graduate level, but it is recommended for doctoral course participants.

 \cdot Priority will be given to participants in degree programs and fellowships who are required to complete this course. If there are too many applicants for the course, the instructor in charge will contact you via the ELMS to confirm your place.

*The first class will be held between 4/22-26. Students interested in attending must complete the scheduling form (Deadline for responses: April 10). Information on scheduling is available on the ELMS.

최미성 수	小兴时间; 第三人	1		
科目名 Course Title 講義題目 Subtitle		lvanced Chemistry] だ[Practical Data Science]		
語義題日 Subtrie 責任教員 Instructor	実践的データ科学[Practical Data Science] 中富 晶子 [NAKATOMI Akiko] (大学院理学研究院)			
担当教員 Other Instructors				
科目種別 Course Type				
開講年度 Year	2024	時間割番号 Course Number	094416	
期間 Semester	Intensive	単位数 Number of Credits	1	
授業形態 Type of Class	Lecture	対象年次 Year of Eligible Student	~	
対象学科・クラス Eligible Depar				
ナンバリングコード Numbering	Code	CHEM_ELCOM 6400		
補足事項 Other Information				
授業実施方式 Class Method		1 対面授業科目《対面のみ》		
キーワード Key Words				
Data Science, Social Implement		, Presentation, Career Making		
授業の目標 Course Objectives				
		ment various methods of data science in a he process of social implementation is si		
_	-	who have received an academic research		
		s. The purpose of this class is to a	-	
		of data science, quality control, output		
		to become leaders in various fields by ad	· –	
have cultivated in their current	-			
到達目標 Course Goals				
Students will				
1. be able to understand variou		-		
		ality control when implementing data scien		
	the techniques to	communicate results obtained by data science	ence methods to society.	
授業計画 Course Schedule	internativa laatura	1 Dr. V-i-Line Wede (CEO. DAe Ace	L - /Etive Officer Datada's	
		by Dr. Yoichiro Wada (CEO, D4c Aca Professor, The University of Electro-Co		
as a data scientist for more tha		Professor, the University of Electro Co)MMUNICATIONS) who has been active	
as a uata scientist for more that	II IV years.			
Units 1 to 5: lecture (60 minute	es), exercises (20 m	inutes), and explanations (10 minutes)		
		to perform social implementation role pla	av. Therefore, discussions and data	
analysis are mainly conducted b				
v <u>-</u>	5			
Unit 1: Introduction to Data Sc	cience for Social Imr	plementation, Programming-1 (introductio	on to Python)	
Unit 2: Programming-2 (contro		-		
· ·		e methods-1 (modeling and validation)		
		e methods–2 (various modeling methods a		
	ata science (projec	ct management, program test, output o	check), Communicating to society	
(reporting / presentation)			0 1	
Unit 6: Explanation of the case Unit 7: Performing data analysi		ion by group, presentation of the results o	of each group	
Unit 8: Performing data analysi	· •			
		sion for each group, summary of the lectu	Ire	
準備学習(予習・復習)等の内容				
Advance preparation				
It is necessary to bring your	personal PC. Insta	all the necessary software (all free) befor	re class. Procedure manual will be	
distributed.				
Pre-learning materials will be p	rovided for compute	er language beginners.		
		em home and submit them by the deadline		
	ents of the present	tation in Unit 9, please do so and submi	it it before the deadline announced	
during class.				
	·			
The e-mail address for submiss		ring class.		
成績評価の基準と方法 Gradin		ass by a-mail to the designated address		

(1) Submit assignments that will be given during class by e-mail to the designated address.

Accuracy and logical consistency of the submitted content will be evaluated.

The level of understanding of the lecture will also be evaluated through the the submitted content.

(2) Present the output of the role play that the group worked on.

Then, Submit presentation file by e-mail to the designated address.

Originality of the problem setting, accuracy of the analysis, logical consistency of the presentation and ease of understanding of the submitted content will be evaluated.

The percentage of the total evaluation is 40% for (1) and 60% for (2).

他学部履修の条件 Other Faculty Requirements

Priority will be given to participants in the Ambitious program for smart materials science (SMatS) and other degree programs who are required to complete this course. If there are too many applicants for the course, the instructor in charge will contact you to confirm your place.

テキスト・教科書 Textbooks

講義指定図書 Reading List

参照ホームページ Websites

https://phdiscover.jp/hu/smats/, https://sites.google.com/eis.hokudai.ac.jp/dxphd-fellow/home, https://sites.google.com/elms.hokudai.ac.jp/ambitious-phd-fellow/home

研究室のホームページ Websites of Laboratory

備考 Additional Information

This class will be held on the following dates. Jun 24 (Mon) 3rd-5th period (13:00-18:00) Jun 25 (Tue) 3rd-5th period (13:00-18:00) Jun 26 (Wed) 3rd-5th period (13:00-18:00) The lecture room is Room 2-409, Science Bldg. 2.

科日夕 Course Title	12世界回講美[1]		
科目名 Course Title	化学特別講義[Advanced Chemistry]		
講義題目 Subtitle	Leading and Advanced Molecular Chemistry and Engineering IIB - 2024[Leading and Advanced Molecular Chemistry and Engineering IIB - 2024]		
主け教具 In the state			
責任教員 Instructor 担当教員 Other Instructors		KOSHI Kei](大学院理学研究院) NN (National Yong Ming Chica Tung Univer	-:+\
		AN (National Yang Ming Chiao Tung Univer	Sity)
科目種別 Course Type	0004	ᆎᄪᇸᆇᆸᇬ᠁ᆘᆄᇑ	004401
開講年度 Year 期間 Someotor	2024	時間割番号 Course Number	094421
期間 Semester 授業形態 Type of Class	Intensive Lecture	単位数 Number of Credits 対象年次 Year of Eligible Student	1~
投集形態 Type of Class 対象学科・クラス Eligible Depar		为家牛次 Tear Of Eligible Student	, .
対象子科 ジノス Eligible Depai ナンバリングコード Numbering		CHEM_ELCOM 6401	
インショート Numbering 補足事項 Other Information	Code	CHEM_BECOM 0401	
授業実施方式 Class Method		1 対面授業科目《対面のみ》	
		1 刈山这禾竹口\\刈山>>>ア//	
キーワード Key Words	II -mission Organi	Elvenessense-guided aurgemy EDA-	duanagent duag
Fluorescence techniques, NIK- 授業の目標 Course Objectives		dyes, Fluorescence-guided surgery, FDA-	approved nuorescent uyes
=		bry and evolution of fluorescence technolog	w. They will then delve into the
		s from a molecular perspective. The course	
		of the past decade and discuss the curry	
		les. On the application front, the course v	÷
		in clinical surgery. Finally, the course wil	
fluorescence technology.	arrene appendix	in omnour ourgory, 1, ,	
到達目標 Course Goals			
	urse are to assist s	tudents in: 1) Understanding the working p	principles of fluorescence imaging
		and limitations; 2) Grasping the design conc	
their impact on optical properti	ies; 3) Understandir	g the types of fluorescent dyes available in	the market and future trends.
授業計画 Course Schedule			
1. Introduction to fluorescent techniques			
2. Design of bright organic fluorophores			
3. Development of NIR-emissive dyes			
4. The role of fluorescence technique in biomedical diagnosis			
5. Current advanced of NIR-II fluorescence-guided imaging in clinical and future challenges			
準備学習(予習・復習)等の内容			
		Chemistry textbooks at undergraduate level	is highly recommended.
成績評価の基準と方法 Gradin		- the - mading	
One final written exam will be g 他学部履修の条件 Other Facu		r the grading.	
	lity requirements		
テキスト・教科書 Textbooks			
講義指定図書 Reading List			
時数旧に凶首 Meading List			
参照ホームページ Websites			
	as part of the Hol	kaido Summer Institute., For more inform	nation (invited lecturers, course
details, etc.),	please	visit the	website below:,
, , , , , , , , , , , , , , , , , , , ,	*	/en/courses/CourseDetail=G056	
研究室のホームページ Websit	·····		

科目名 Course Title	化学性別講美 [Ad	wanood Chomistry]	
講義題目 Subtitle	化学特別講義[Advanced Chemistry] Leading and Advanced Molecular Chemistry and Engineering IIC - 2024[Leading and		
語我起日 Subuue			ering IIC - 2024[Leading and
▲「松昌 !		ar Chemistry and Engineering IIC - 2024]	
責任教員 Instructor		OSHI Kei] (大学院理学研究院)	
担当教員 Other Instructors	ZeeHwan KIM (See	oul National University)	
科目種別 Course Type	2424		
開講年度 Year	2024	時間割番号 Course Number	094422
期間 Semester	Intensive	単位数 Number of Credits	1
授業形態 Type of Class	Lecture	対象年次 Year of Eligible Student	~
対象学科・クラス Eligible Depar			
ナンバリングコード Numbering Code CHEM_ELCOM 6401			
補足事項 Other Information			
授業実施方式 Class Method	受業実施方式 Class Method 1 対面授業科目《対面のみ》		
キーワード Key Words			
Physical Chemistry, Nano-Opt	cs, Plasmonics, Mol	ecular Spectroscopy, Photo-Catalysis, Ligh	it-Harvesting
授業の目標 Course Objectives	3		
This course aims to offer the	e students with the	principle and application of modern spe	ctroscopy, optical imaging, and
photochemistry enabled by lig	ht-field confined a	t nanometer scale. Students will learn (1)) how to confine light to a few
nanometer scales, (2) how such	a confined field int	eracts with molecules, and (3) the application	on of the interactions to physics,
chemistry, materials science,	and bio-imaging. F	or the field confinement, we will primaril	y focus on the physics of field
confinement caused by plasmo	nic nanostructures	and their validation. The application inclue	les nanoscale spectroscopy at a
single-molecule regime, nano-s	cale chemical imagin	ng, and plasmon-induced / enhanced photo-	-catalysis.
到達目標 Course Goals			
The goal of this course is to h	elp students (1) une	derstand the quantum mechanics and optic	s of nanoconfined light-molecule
interaction and (2) gain a gene	ral perspective on v	what is currently possible with the state-o	f-art spectroscopy, imaging, and
photochemistry techniques ena	bled by the nano-co	nfined electromagnetic field.	
授業計画 Course Schedule			
1. Introduction: how to focus li	ght to a nanometer :	scale and what it can do	
2. Fundamentals: Molecular qua	antum mechanics an	d elementary electrodynamics of light	
3. Fundamentals: Quantum mee	chanics of light-mole	ecule interaction	
4. Plasmonics of nanostructures	field enhancement	and field confinement	
5. Nano-scale molecular spectr	oscopy and imaging		
6. Plasmon-induced and polarit	onic chemistry		
準備学習(予習・復習)等の内容	容と分量 Homework		
To read the basic parts of Phys	ical and Quantum C	hemistry textbooks at undergraduate level	is highly recommended.
成績評価の基準と方法 Gradin			
One final written exam will be g			
他学部履修の条件 Other Facu	lty Requirements		
テキスト・教科書 Textbooks			
講義指定図書 Reading List			
参照ホームページ Websites			
This course will be provided	as part of the Hok	kaido Summer Institute., For more inform	nation (invited lecturers, course
details, etc.),	please	visit the	website below:,
https://hokkaidosummerinstitu	te.oia.hokudai.ac.jp	/en/courses/CourseDetail=G057	
研究室のホームページ Websit	es of Laboratory		
備考 Additional Information			

科目名 Course Title	化学特別講義[Advanced Chemistry]		
講義題目 Subtitle		vanced Materials Chemistry and Engine	ering IIA – 2024[Leading and
		lls Chemistry and Engineering IIA – 2024]	
責任教員 Instructor		〈OSHI Kei] (大学院理学研究院)	
担当教員 Other Instructors	Mengning DING (N	Vanjing University)	
科目種別 Course Type			
開講年度 Year	2024	時間割番号 Course Number	094423
期間 Semester	Intensive	単位数 Number of Credits	1
授業形態 Type of Class	Lecture	対象年次 Year of Eligible Student	~
対象学科・クラス Eligible Depa			
ナンバリングコード Numbering	Code	CHEM_ELCOM 6401	
補足事項 Other Information			
授業実施方式 Class Method		1 対面授業科目《対面のみ》	
キーワード Key Words			
-	mistry, Catalysis, Pl	hysical Chemistry, Sustainable Chemistry	
授業の目標 Course Objectives			
•		in electrochemistry and electrochemical pr	pocesses at the molecular level to
		se applications in clean energy and sustaina	
		ical methods, electron transfer, double layer	
		of state-of-the-art design, synthesis and a	
		as CO2 conversion, biomass upgrading, C	
for green hydrogen energy will		us controloin, stemace apprassio, -	
到達目標 Course Goals			
	o help students (1`) understand the fundamental working pri	inciples of electrochemistry and
electrochemical interfacial processes; (2) understand the examples of application of electrochemical technology in the sustainable chemistry, such as electrocatalysis, electrosynthesis and electrochemical devices; (3) understand the			
		and structure-property relationships to achi	
		nods to prepare and characterize state-of-th	
授業計画 Course Schedule			
1. Introduction to electrochemi	istrv		
2. Characterization of the electrochemical processes			
3. Advanced technology for the in-depth investigation on micro-electrokinetics and their modulation			
4. Electrocatalytic water splitting for hydrogen production			
5. Electrocatalytic conversion of CO2 (and other chemicals) to value-added products			
準備学習(予習・復習)等の内容	· · · · · -		
		book at undergraduate level is highly recom	imended.
成績評価の基準と方法 Gradin			
One final written exam will be g		r the grading.	
他学部履修の条件 Other Faci	-		
テキスト・教科書 Textbooks			
講義指定図書 Reading List			
参照ホームページ Websites			
	nort of the Hol	For more inform	
details, etc.),	-	kkaido Summer Institute., For more inform	
, , , , , , , , , , , , , , , , , , , ,	please Ito oin hokudai ao in	visit the	website below:,
- nttps:// norkaldosummerinstitu 研究室のホームページ Websit		o/en/courses/CourseDetail=G048	
研究室のホームページ websit https://mdinglab.weebly.com/	es of Laboratory		
備考 Additional Information			
漏考 Additional Information			

科目名 Course Title	化学特别講義[A	dvanced Chemistry]		
		-		
講義題目 Subtitle	Leading and Ad	Leading and Advanced Materials Chemistry and Engineering IIIA - 2024[Leading and		
	Advanced Materi	als Chemistry and Engineering IIIA – 2024]		
責任教員 Instructor	村越 敬[MURA	村越 敬 [MURAKOSHI Kei] (大学院理学研究院)		
担当教員 Other Instructors	YOSHIO Masafur	ni (NIMS), MASUDA Takuya (NIMS)		
科目種別 Course Type				
開講年度 Year	2024	時間割番号 Course Number	094424	
期間 Semester	Intensive	単位数 Number of Credits	1	
授業形態 Type of Class	Lecture	対象年次 Year of Eligible Student	~	
対象学科・クラス Eligible Depa	rtment/Class			
ナンバリングコード Numbering	ナンバリングコード Numbering Code CHEM_ELCOM 6401			
補足事項 Other Information				
授業実施方式 Class Method		1 対面授業科目《対面のみ》		
キーロード Kay Warda				

Supramolecular Chemistry, Printed Electronics, Electrochemistry, Advanced Characterization Techniques

授業の目標 Course Objectives

In this course, students will delve into molecular assembly chemistry, exploring the design and device applications of ion and electron functional organic and polymer materials, including printed electronics. They will also delve into advanced interface analysis techniques. The lectures will primarily focus on the intricate process of creating functional organic materials through nanostructure formation via molecular self-assembly. This encompasses various applications such as soft actuators and separation membranes utilizing liquid crystals, block copolymers, and covalent organic frameworks. Moreover, the course will cover advanced characterization techniques, including X-ray photoelectron spectroscopy, X-ray absorption/fluorescence spectroscopy, vibrational spectroscopy, electron microscopy, scanning probe microscopy, etc, specifically targeting cathode, anode, and electrolyte materials used in lithium-ion and fuel cells. Throughout the course, students will explore how structural design and orientational control in organic materials can enhance their electrical and mechanical properties in functional devices. Additionally, they will gain insights into the changes occurring in the surface chemistry of electrodes and electrolyte interfaces during capacitor and battery charging.

到達目標 Course Goals

The goal of this course are as follows: Understand the intermolecular interactions in organic assemblies and grasp the fundamental working principles of organic ionic and electronic devices. Gain insight into materials design, engineering, processing, and the relationships between structure and properties to achieve optimal material function. Develop problem-solving skills and explore solutions based on acquired knowledge. By pursuing these objectives, students will develop the skills necessary to make global contributions in their field.

授業計画 Course Schedule

1. Supramolecular materials chemistry

2. Soft Actuators and Sensors

3. Functional Nanostructured Membranes

4. Printed Electronics

5. Lithium-ion batteries and Fuel Cells

6. Advanced Characterization Techniques

準備学習 (予習・復習)等の内容と分量 Homework

To read the basic parts of Organic and Physical Chemistry textbooks at undergraduate level is highly recommended.

成績評価の基準と方法 Grading System

Two reports will be given to students for the grading.

他学部履修の条件 Other Faculty Requirements

テキスト・教科書 Textbooks

講義指定図書 Reading List

参照ホームページ Websites

This course will be provided as part of the Hokkaido Summer Institute., For more information (invited lecturers, coursedetails,etc.),pleasevisitthewebsitebelow:,https://hokkaidosummerinstitute.oia.hokudai.ac.jp/en/courses/CourseDetail=G050

研究室のホームページ Websites of Laboratory

科目名 Course Title	化学特別講義	化学特別講義[Advanced Chemistry]		
講義題目 Subtitle	Leading and A	Leading and Advanced Biological and Polymer Chemistry and Engineering I – 2024[Leading		
	and Advanced	and Advanced Biological and Polymer Chemistry and Engineering I – 2024]		
責任教員 Instructor	坂口 和靖[S	AKAGUCHI Kazuyasu] (大学院理学研究院)	
担当教員 Other Instructors	Pascale Legau	lt (University of Montreal),		
	KAMADA Rui	KAMADA Rui (理学研究院), NAKAGAWA Natsumi (理学研究院)		
科目種別 Course Type				
開講年度 Year	2024	時間割番号 Course Number	094425	
期間 Semester	Intensive	単位数 Number of Credits	1	
授業形態 Type of Class	Lecture	対象年次 Year of Eligible Student	~	
対象学科・クラス Eligible Dep	artment/Class			
ナンバリングコード Numbering	g Code	CHEM_ELCOM 6401		
補足事項 Other Information				
授業実施方式 Class Method 1 対面授業科目《対面のみ》				
キーワード Key Words				
RNA structure and function	microRNAs mi	croRNA biogenesis microRNA regulation	let-7 Parkinson's Disease alpha-	

RNA structure and function, microRNAs, microRNA biogenesis, microRNA regulation, let-7, Parkinson's Disease, alphasynuclein, viral infections, Zika Virus

授業の目標 Course Objectives

MicroRNAs (miRNAs) constitute an important class of small non-coding RNAs that, like transcription factors, play a central role in regulating gene expression. They function by targeting complementary sequences of mRNA, generally resulting in translational inhibition. Misregulation of miRNA levels can change gene expression patterns, and these changes have been directly linked to developmental defects and several human diseases, such as cancer and neurodegenerative diseases.

The course will focus on better understanding the following topics:

- 1. The importance of miRNA in gene regulations for health and diseases
- 2. The general pathway of miRNA biogenesis and the main enzymes involved in post-transcriptional regulation
- 3. The different mechanisms for regulating miRNA levels
- 4. The discovery of novel regulators of miRNA levels

到達目標 Course Goals

Here are some of the key concepts and skills students will develop:

- 1. Appreciate the importance of miRNA levels in health and disease
- 2. Appreciate the role of RNA in regulating gene expression
- 3. Understand the structure and function of key proteins involved in miRNA maturation
- 4. Understand the role of RNA-binding proteins in regulating miRNA levels
- 5. Apply simple web-based tools for miRNA research
- 6. Become familiar with experimental techniques used in miRNA biology
- 7. Critical evaluation of miRNA research litterature

授業計画 Course Schedule

July 29th (Mon) 10:30 ~ 12:00 Lecture 13:00 ~ 14:30 Computer exercise using web-based tools (students will need to bring their own computer)

July 30th (Tue) 10:30 $\stackrel{\sim}{}$ 12:00 Lecture 13:00 $\stackrel{\sim}{}$ 14:30 Discussion

July 31st (Wed) 10:30 ~ 12:00 Lecture 13:00 ~ 14:30 Discussion

August 1st (Thu) 10:30 ~ 12:00 Lecture 13:00 ~ 14:30 Seminar

準備学習 (予習・復習)等の内容と分量 Homework

To be provided at the first class

成績評価の基準と方法 Grading System

Assignment on specified topics regarding "microRNA function" and "microRNA maturation" (60%); Active student participation in class (40%)

他学部履修の条件 Other Faculty	/ Requirements
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テキスト・教科書 Textbooks

None

講義指定図書 Reading List

To be provided at the first class

参照ホームページ Websites

This course will be provided as part of the Hokkaido Summer Institute., For more information (invited lecturers, coursedetails,etc.),pleasevisitthewebsitebelow:,https://hokkaidosummerinstitute.oia.hokudai.ac.jp/en/courses/CourseDetail=G042

研究室のホームページ Websites of Laboratory

http://airen.bcm.umontreal.ca

https://www.chem.sci.hokudai.ac.jp/~biochem/

備考 Additional Information

Other Instructor: Prof. Pascale Legault (University of Montreal)

科目名 Course Title	広田化受陸別講講	轰[Advanced-Applied Chemistry]		
講義題目 Subtitle	有機プロセス工学特別 Lecture 2024[Chemical Process Engineering 2024]			
責任教員 Instructor	猪熊 泰英 [INOKUMA Yasuhide] (大学院工学研究院)			
担当教員 Other Instructors	久木 一朗 (大阪大学)			
科目種別 Course Type				
開講年度 Year	2024	時間割番号 Course Number	094431	
期間 Semester	Intensive	単位数 Number of Credits	1	
授業形態 Type of Class	Lecture	ッロッ Year of Eligible Student	\sim	
対象学科・クラス Eligible Department/Class				
ナンバリングコード Numbering		CHEM_ELCOM 6410		
補足事項 Other Information				
授業実施方式 Class Method 1 対面授業科目《対面のみ》				
キーワード Key Words				
Supramolecules, Organic Cryst	als Organic Synthe	osis Chirality		
授業の目標 Course Objectives		Sis, Chinanty		
		mong organic molecules play an important	role in producing giant functional	
		ntal intermolecular interactions such, as hy		
		nic molecules. This will lead to an under		
		created by reversible intermolecular inte		
		e functions of cavities surrounded by su		
		of molecular chirality, from basic principles		
sectional manner.				
到達目標 Course Goals				
After successful completion of	this course, you wil	l be able to		
1. explain how and why giant m	olecular assemblies	are formed from small molecules.		
2. acquire the basic knowledge	necessary to unde	rstand the latest research papers and gain	insight into the molecular design	
of supramolecules-based functional organic materials.				
授業計画 Course Schedule				
1. Basic concept of Supramolecules				
2. Intermolecular interactions between small molecules				
3. Formation of Giant Molecular Assemblies				
4. Synthetic chemistry of supramolecules 5^{2}				
$5^{\sim}7$. Chemical phenomena in void space of supramolecules				
8. Public lecture				
準備学習(予習・復習)等の内容				
_		knowledge of organic chemistry taken in t	-	
		h papers on supramolecules and organic of		
		d approximately 2 hours each for preparati	on and multiple assignments using	
the textbook and the assigned	· · ·			
成績評価の基準と方法 Gradin		uning the lecture (20%) and the perpert case	$r_{\rm mm}$ and (80%)	
他学部履修の条件 Other Face		uring the lecture (20%) and the report assig		
テキスト・教科書 Textbooks				
	合子構造と反応。右		1.•玉尾皓亚•本自垢壮一•些临正	
大学院 Lecture 有機化学 I 分子構造と反応・有機金属化学 第 2 版/野依良治・中筋一弘・玉尾皓平・奈良坂紘一・柴崎正 勝・絵本啓介(編)・東京化学同人				
勝・鈴木啓介(編):東京化学同人 大学院 Lecture 有機化学 II 有機合成化学・生物有機化学 第2版/野依良治・中筋一弘・玉尾皓平・奈良坂紘一・柴崎正勝・				
入字阮 Lecture 有機化字 II 有機合成化字 生物有機化字 弟 2 版/ 野侬 長冶 中肋一弦 · 玉尾皓平 · 奈良坂絋一 · 朵崎止勝 · 橋本俊一 · 鈴木啓介 · 山本陽介 · 村田道雄(編):東京化学同人				
请義指定図書 Reading List				
参照ホームページ Websites				
研究室のホームページ Websit	es of Laboratory			
http://www.chem.es.osaka-u.a	-			
備考 Additional Information				

10 A		жг., , , , ,	
科目名 Course Title		复[Advanced-Applied Chemistry]	
講義題目 Subtitle		轰 2024[Materials Chemistry 2024]	
責任教員 Instructor		[ADA Toshihiro] (大学院工学研究院)	
担当教員 Other Instructors	浅川 鋼児 (キオ)	ワシア(株))	
科目種別 Course Type			1
開講年度 Year	2024	時間割番号 Course Number	094432
期間 Semester	Intensive	単位数 Number of Credits	1
授業形態 Type of Class	Lecture	対象年次 Year of Eligible Student	~
対象学科・クラス Eligible Depa			
ナンバリングコード Numbering Code		CHEM_ELCOM 6411	
補足事項 Other Information			
授業実施方式 Class Method		1 対面授業科目《対面のみ》	
キーワード Key Words			
semiconductor processes and n	naterials, integrated	circuits, memory, nano-fabrication. lith	ography, resists, electronics
授業の目標 Course Objective			
This course is intended to pro	vide basics of fabric	cation process and materials of modern	large-scale digital integrated circuits
(CPUs, memories etc.). It is a	lso aims at teaching	basics of related electronics for student	ts who do not major in electronics.
到達目標 Course Goals			
Through a series of lectures,	students understar	nd processes of semiconductor fabricat	ion including physics and chemistry
behind the process.			
授業計画 Course Schedule			
Detailed schedule will be inform	ned one month befor	re the start of this course. The content	s will be as follows:
- Introduction to modern elect	ronics		
- History of semiconductor min	niatualization (scalin	g)	
- Memory devices			
- Overview of semiconductor p	rocessing		
- Lithography: Optics and resi	sts for nanofabricati	on	
- Thin film growth: Chemical v	apor deposition (CV	D) and atomic layer deposition (ALD)	
- Etching: Plasma and reactive	-ion etching (RIE)		
- Chemical mechanical polishir			
準備学習 (予習・復習)等の内			
		ance. Students are expected to study b	efore and after each lecture.
成績評価の基準と方法 Gradir			
		after the lecture and the degree of parti	icipation in the lecture.
他学部履修の条件 Other Fac			
	granted based on c	riteria about the economic security of th	ne company the lecturer belongs to.
テキスト•教科書 Textbooks			
講義指定図書 Reading List			
参照ホームページ Websites			
研究室のホームページ Websit	es of Laboratory		
備考 Additional Information			

科目名 Course Title	広用化学特別語	毒義[Advanced-Applied Chemistry]		
講義題目 Subtitle		生物機能高分子特別 Lecture 2024[Advanced Applied Biochemistry 2024]		
責任教員 Instructor		MATSUMOTO Kenichiro] (大学院工学研		
			71.19元/	
担当教員 Other Instructors	山田 美和(岩	手大字)		
科目種別 Course Type				
開講年度 Year	2024	時間割番号 Course Number	094433	
期間 Semester	Intensive	単位数 Number of Credits	1	
授業形態 Type of Class	Lecture	対象年次 Year of Eligible Student	~	
対象学科・クラス Eligible Depa	artment/Class			
ナンバリングコード Numbering	g Code	CHEM_ELCOM 6410		
補足事項 Other Information				
授業実施方式 Class Method		1 対面授業科目《対面のみ》		
キーワード Key Words				

environmentally friendly polymers, biodegradable plastics, bioplastics, microorganisms, enzymes, metabolic engineering, protein engineering, biomass utilization, chemical recycling

授業の目標 Course Objectives

The aim of this course is to develop human resources who can respond to the global environmental problems we are facing today and to develop technologies that can contribute to solving these problems.

In this lecture, students will learn about the synthesis and degradation of bioplastics, as well as the current status of degradation of persistent plastics.

The objective of this course is to understand the synthesis and degradation of bioplastics, as well as the current status of degradation of persistent plastics, from both practical and basic research perspectives.

到達目標 Course Goals

In this class, students will mainly learn about the synthesis and degradation mechanisms of bioplastics. In addition, students will learn about enzymatic degradation technologies for persistent plastics.

授業計画 Course Schedule

Synthesis and degradation of plastics using microorganisms and enzymes

Lecture 1: "Environmental Problems and Bioplastics

Students will learn about the global environmental problems we are facing today, the general concept, types, and synthesis methods of bioplastics, and why bioplastics are expected to contribute to solving environmental problems.

Lecture 2: "Bioplastics Biosynthesized by Microorganisms

Students will learn about bioplastics biosynthesized by microorganisms, their properties, the current status of industrialization, biosynthesis pathways in microbial cells, and the mechanism of polymerization by enzymes.

Lecture 3: "Frontiers of Research on Bioplastic Synthesis by Microorganisms Students will learn about the latest research on bioplastic synthesis by microorganisms using microbial screening, metabolic engineering, and protein engineering. In addition, potential future applications will be discussed.

Lecture 4: "Degradation Mechanisms of Bioplastics

Students will understand the degradation mechanisms of bioplastics that exhibit biodegradability and the factors that influence degradation.

Lecture 5: Degradation of Biodegradable Plastics in the Environment and the Plastisphere Students learn about the plastisphere and discuss the impact of biodegradation of biodegradable plastics on the environment.

Lecture 6: Enzymatic Degradation of Biodegradable Plastics by Microorganisms and Recycling Technologies Learn about the latest research and applications of enzymatic degradation and recycling technologies for persistent plastics.

準備学習 (予習・復習)等の内容と分量 Homework

Read carefully the materials distributed in advance, if any. Students are expected to spend approximately 2 hours preparing for and reviewing the material. During the class, students will be asked for their opinions on their own ideas and will be given opportunities to discuss them with each other.

After the class, students are required to submit a written report on the issues they have set.

成績評価の基準と方法 Grading System

Attendance of 70% or more of the class sessions is a prerequisite for grading. Grades will be based on (1) attitude toward study (40%) and (2) reports (60%).

The report will be evaluated based on the level of understanding of the lecture content, the depth of discussion, and the logical development of the description. Excellent (100 to 90 points), Superior (89 to 80 points), Good (79 to 70 points), Acceptable (69 to 60 points), Impossible (less

than 60 points), Superior (59 to 50 points), Good (79 to 70 points), Acceptable (59 to 60 points), Impossible (less

他学部履修の条件 Other Faculty Requirements

テキスト・教科書 Textbooks

講義指定図書 Reading List

参照ホームページ Websites

研究室のホームページ Websites of Laboratory

http://appl-micro.agr.iwate-u.ac.jp/index.html

https://biosynchem.eng.hokudai.ac.jp/

科目名 Course Title	古田ル学時期	要美「Advanged-Applied Chemistry]		
講義題目 Subtitle	応用化学特別講義[Advanced-Applied Chemistry]			
青任教員 Instructor	異分野ラボビジット 高橋 正行 [TAKAHASHI Masayuki] (大学院理学研究院)			
担当教員 Other Instructors	同個 正门 [I AKAI IASI II Masayuki] (人于他连于明元的)			
科目種別 Course Type				
開講年度 Year	2024	時間割番号 Course Number	094434	
期間 Semester	Intensive	単位数 Number of Credits	1	
授業形態 Type of Class	Lecture	対象年次 Year of Eligible Student	~	
対象学科・クラス Eligible Depa				
ナンバリングコード Numbering	Code	CHEM_ELCOM 6412		
補足事項 Other Information				
授業実施方式 Class Method		1 対面授業科目《対面のみ》		
キーワード Key Words				
Interdisciplinary research, Cro	ss-disciplinary ex	change, comprehensive perspective		
授業の目標 Course Objective				
The goal is to lean about the	e expert knowled	ges and skills in the different fields prov	ided by a host laboratory. For this	
		he host laboratory for a period of about 2		
到達目標 Course Goals				
	rch with research	ners with different backgrounds by cultiva	ating a wide range of communication	
skills through discussions.				
•To be able to correlate the	student's own re	search and those in different fields by ac	quiring comprehensive perspectives,	
which is necessary to promote				
授業計画 Course Schedule				
	o graduate stude	nts of "Ambitious program for smart m	aterials science" and those joining	
MANABIYA program of WPI ICReDD.				
•Staying a host laboratory will be for a period of two weeks to two months between April to next March.				
•Students are requested to leave their own laboratory and stay in the host laboratory to engage the research project provided				
by the host laboratory and to acquire specialized knowledge and skills in different fields.				
準備学習 (予習・復習)等の内	容と分量 Homewo	ork		
		activities of each laboratory thoroughly	and select a laboratory that matches	
the research field you wish to s			-	
成績評価の基準と方法 Gradir				
		f the submitted report and the discussion v	with the teacher of this lecture about	
the training content.		-		
他学部履修の条件 Other Fac	ulty Requirement	S		
テキスト・教科書 Textbooks				
講義指定図書 Reading List				
参照ホームページ Websites				
	ts/. https://www	icredd.hokudai.ac.jp/ja/manabiya		
研究室のホームページ Websit				
備考 Additional Information				

Follow the instructions of the host laboratory.

科目名 Course Title	応用化学特別	溝義[Advanced-Applied Chemistry]		
講義題目 Subtitle		dvanced Molecular Chemistry and Engineer	ring I - 2024[Leading and Advanced	
	-	nistry and Engineering I – 2024]		
責任教員 Instructor		伊藤 肇 [ITOH Hajime] (大学院工学研究院)		
担当教員 Other Instructors		Andrei K. YUDIN (University of Toronto), KUBOTA Koji (工学研究院)		
科目種別 Course Type				
開講年度 Year	2024	時間割番号 Course Number	094441	
期間 Semester	Intensive	単位数 Number of Credits	1	
授業形態 Type of Class	Lecture	对象年次 Year of Eligible Student	~	
対象学科・クラス Eligible Dep			<u> </u>	
ナンバリングコード Numberin		CHEM ELCOM 6411		
補足事項 Other Information	g oout			
授業実施方式 Class Method		2 対面授業科目《一部遠隔》		
を来交遣方式 Class Method キーワード Key Words		2 八面汉来竹口《 叩逐隔//		
•	nthasia maahanaa	homical synthesis		
organic chemistry, organic sy 授業の目標 Course Objectiv		nemical synthesis		
		s important for the effective use of reso	ources and for supporting people's	
		leading researchers from abroad and Hol		
-		been developed remarkably recently and		
	-	r new synthetic reagents, peptide conform		
synthesis.	Courses will cove	i new synthetic reagents, peptide collionii	actori, and methanothemical organic	
到達目標 Course Goals				
	course, vou will be	able to know concepts and recent progres	s in new synthetic reagents peptide	
conformation, and mechanoch			s in new synthetic reagents, peptide	
授業計画 Course Schedule	lennear of game syn			
Course Schedule (the order o	f the following lect	ures is subject to change)		
1. Mechanochemical organic				
2. Mechanochemical organic				
3. Structure and conformation		S		
4. New synthetic reagents wit				
5. Research proposal I	*			
6. Research proposal II				
準備学習(予習・復習)等の内	羽容と分量 Homew	ork		
Students will make proposal p	presentations and r	eports.		
成績評価の基準と方法 Grad	ing System			
Grades are judged based on c	lass attitude, pres	entations, and reports during the course.		
他学部履修の条件 Other Fa				
テキスト・教科書 Textbooks				
講義指定図書 Reading List				
将我们に凶言 Neading List				
参照ホームページ Websites				
	l as part of the H	Hokkaido Summer Institute., For more in	formation (invited lecturers, course	
details, etc.),	please		website below:	
	tute.oia.hokudai.ac	c.jp/en/courses/CourseDetail=G054		
nttps://hokkaidosummerinsti				
		1		
研究室のホームページ Webs	ites of Laboratory	,		
	ites of Laboratory dai.ac.jp/en.html			

科目名 Course Title	応用化学特別	講義[Advanced-Applied Chemistry]		
講義題目 Subtitle	Leading and Advanced Molecular Chemistry and Engineering IIA - 2024[Leading and			
	Advanced Mole	olecular Chemistry and Engineering IIA – 2024]		
責任教員 Instructor	清水 研一[SH	水 研一 [SHIMIZU Kenichi] (触媒科学研究所)		
担当教員 Other Instructors		NG (Peking University), Congyang WANG (Chinese Academy of Scien		
	SON Tsui (触妙	某科学研究所)		
科目種別 Course Type				
開講年度 Year	2024	時間割番号 Course Number	094442	
期間 Semester	Intensive	単位数 Number of Credits	1	
授業形態 Type of Class	Lecture	対象年次 Year of Eligible Student	~	
対象学科・クラス Eligible Depa				
ナンバリングコード Numbering	Code	CHEM_ELCOM 6411		
補足事項 Other Information				
授業実施方式 Class Method		2 対面授業科目《一部遠隔》		
キーワード Key Words				
catalysis, chemical industry, bi	omass, organome	tallics, organic synthesis		
授業の目標 Course Objective	s			
		global warming, energy crisis, and food pro-		
		mission of finding ways to solve these		
	-	lytic chemistry, which supports the mod		
		us on catalytic technologies for solving		
		evelopment of next generation chemical in		
		arch on innovative catalysis based on new		
		e course "Fundamentals and Application oblems and pioneer next-generation techn		
到達目標 Course Goals	solve modern pro	Solems and ploneer next-generation techn	ologies.	
到 建日標 Oourse Goals By the end of this course you v	vill be able to up	dorstand		
1. to understand principles abo				
2. to learn CO2 conversion us				
		or utilization of natural carbon resources		
4. to learn how to develop nan	-			
•		nsformation of P4 or N2 into fine chemicals		
6. to learn applications of trans				
授業計画 Course Schedule		<u> </u>		
1.Principals for design and cha	racterization of h	eterogeneous catalysts		
2.Application of heterogeneous				
3.Frontiers of catalyst research	h for utilization o	f natural carbon resources		
4.Development of novel cataly	tic functions of na	anomaterials		
5.Activation and Transformation	on of White Phos	phorus to Organophosphorus Compounds		
6.Nitrogen Fixation: Fundamer	ntals and Catalysi	s		
7. The History, Current Status	and Future of Ra	re-Earth Organometallic Chemistry		
8. Titanium in Organic Synthes				
準備学習(予習・復習)等の内				
Students will be asked to write		end of each lecture.		
成績評価の基準と方法 Gradir				
	i active attendanc	ce records and reports at the end of each l	ecture.	
Grades will be judged based or テキスト・教科書 Textbooks				
テキスト・教科書 Textbooks 講義指定図書 Reading List 参照ホームページ Websites	as part of the l	Hokkaido Summer Institute., For more in	nformation (invited lecturers, cours	
テキスト・教科書 Textbooks 講義指定図書 Reading List 参照ホームページ Websites This course will be provided details, etc.), https://hokkaidosummerinstitu	please ite.oia.hokudai.ad	visit the c.jp/en/courses/CourseDetail=G055	nformation (invited lecturers, cours website below:	
テキスト・教科書 Textbooks 講義指定図書 Reading List 参照ホームページ Websites This course will be provided details, etc.),	please ite.oia.hokudai.ad	visit the c.jp/en/courses/CourseDetail=G055		

科目名 Course Title	応用化学特別講論	応用化学特別講義[Advanced-Applied Chemistry]				
講義題目 Subtitle	Leading and Adva	Leading and Advanced Materials Chemistry and Engineering I - 2024[Leading and Advanced				
	Materials Chemist	try and Engineering I – 2024]				
責任教員 Instructor	三浦 章[MIURA	Akira](大学院工学研究院)				
担当教員 Other Instructors	Laurent Cario (Cl	NRS), Shunsuke SASAKI (CNRS)				
科目種別 Course Type						
開講年度 Year	2024	時間割番号 Course Number	094443			
期間 Semester	Intensive	単位数 Number of Credits	1			
授業形態 Type of Class	Lecture	対象年次 Year of Eligible Student	~			
対象学科・クラス Eligible Depa	rtment/Class					
ナンバリングコード Numbering Code		CHEM_ELCOM 6411				
補足事項 Other Information						
授業実施方式 Class Method 1 対面授業科目《対面のみ》						
キーワード Key Words						

Materials Chemistry, Interdisciplinary collaborations, Solid-state chemistry

授業の目標 Course Objectives

Materials chemistry serves as the intersection where various branches of chemistry come together to explore new materials and enhance their functionalities. For success in such interdisciplinary fields, this course will provide an idea about how to formulate research questions, adapt their expertise to other fields, and collaborate effectively with researchers from different backgrounds. The case studies will be presented to introduce various approaches in materials design and highlight common thoughts that bridge different sub-disciplines. The course also includes hands-on sessions where students participate in roleplaying exercises. These exercises simulate the process of launching a new research project in a completely different field, allowing students to apply their expertise in novel contexts.

到達目標 Course Goals

By the course's end, students will cultivate adaptable mindsets, helping them to envision research projects across diverse disciplines throughout their academic and professional careers.

授業計画 Course Schedule

1. Guidance of lectures

2. Synthetic chemistry of discrete molecules and extended inorganic solids: Common concepts, differences and transversal approaches

3. A case study of transversal approach in solid-state chemistry

4. Ascending technology readiness levels (TRL) from fundamental solid-state chemistry to microelectronic devices: a case study in Mottronics applications

5. Collective brainstorming for new transversal projects

6. A short briefing about writing scientific proposals in Japanese, French and/or European styles, followed by elaboration of each student's transversal project through role playing

準備学習 (予習・復習)等の内容と分量 Homework

A short questionnaire/report will be assigned at every lecture to help students to construct their transversal ideas.

成績評価の基準と方法 Grading System

提出されたレポートにより判断する

他学部履修の条件 Other Faculty Requirements

テキスト・教科書 Textbooks

講義指定図書 Reading List

参照ホームページ Websites

This course will be provided as part of the Hokkaido Summer Institute., For more information (invited lecturers, coursedetails,etc.),pleasevisitthewebsitebelow:,https://hokkaidosummerinstitute.oia.hokudai.ac.jp/en/courses/CourseDetail=G047

研究室のホームページ Websites of Laboratory

科目名 Course Title	応用化学特別講	義[Advanced-Applied Chemistry]			
講義題目 Subtitle	Leading and Advanced Materials Chemistry and Engineering IIB – 2024[Leading and				
	-	als Chemistry and Engineering IIB – 2024			
責任教員 Instructor	忠永 清治 [TADANAGA Kiyoharu] (大学院工学研究院)				
担当教員 Other Instructors	Masashi KOTOBUKI (Ming Chi University of Technology), FUJII Yuta (工学研究院)				
科目種別 Course Type					
開講年度 Year	2024	時間割番号 Course Number	094444		
期間 Semester	Intensive	単位数 Number of Credits	1		
授業形態 Type of Class	Lecture	対象年次 Year of Eligible Student	~		
対象学科・クラス Eligible Depa	rtment/Class				
ナンバリングコード Numbering	Code	CHEM_ELCOM 6411			
補足事項 Other Information					
授業実施方式 Class Method		1 対面授業科目《対面のみ》			
キーワード Key Words					
Electrochemical devices; Elect	rolyte; Electrode; N	Nano-structure; Batteries			
授業の目標 Course Objective	S				
Recently, safe, low-cost, high-	-energy density, an	d long-lasting electrochemical devices fo	or energy conversion and storage are		
highly required for mobile de	evices, electric ve	hicles, and storage for renewable ener	rgy to build a sustainable society.		
•		orphological control of these materials ar	-		
		cal devices and materials science inv	_		
	-	n electrochemical energy conversion and	•		
=		emical devices will be described. The p			
	effect of nano-stru	actures in electrodes for batteries, and	the development of all-solid-state		
batteries are also described. 到達目標 Course Goals					
By the end of this course you v	will be able to				
-		nergy conversion and storage systems			
		s used in electrochemical energy convers	ion and energy storage devices		
•		on the properties of electrochemical dev			
•		emical devices in future energy storage s			
授業計画 Course Schedule			-		
As a HSI course, Professor Ma	sashi Kotobuki (Ba	ttery Research Center of Green Energy,	Ming Chi University of Technology)		
will give most of the lectures.					
The following topics will be cov	vered during this co	ourse.			
1. Fundamental concepts about	t electrochemical e	nergy conversion and storage			
2. Materials used in electroche	mical devices				
3. Introduction of inorganic ma	terials science for e	electrochemical devices			
		for lithium and sodium ion batteries			
5. Fundamentals of solid electr					
6. All-solid-state lithium/sodiu					
		rochemical devices and future energy sto	prage system		
8. Students presentation on to 準備学習(予習・復習)等の内部					
		▲ from WEB page and read designated cha	anter in advance		
		ical devices during this course and make			
stadonto prodita redu bonie pap		active and the course and make	p. 55011001011.		
成績評価の基準と方法 Gradin	g Svstem				
Grade will be determined by ho		vement in this course through			
1. a report on nanostructured materials in electrochemical devices (weightage 80%), and					
2. a presentation on one's research or some topics in electrochemical devices (weightage 20%).					
他学部履修の条件 Other Faculty Requirements					
テキスト・教科書 Textbooks					
No textbook required. Handou	ts will be distribute	d.			
講義指定図書 Reading List					
	Storage Materials	and Devices", Li Lu edited, Materials	Research Forum LLC, ISBN 978-		
1945291265 (2017).					

"Ceramic Electrolytes for All-Solid-State Li Batteries", M. Kotobuki, S. Song, C. Chen, and Li Lu, World Scientific Pub Co Inc ISBN: 978-9813233881(2018).

参照ホームページ Websites

This course will be provided as part of the Hokkaido Summer Institute., For more information (invited lecturers, coursedetails,etc.),pleasevisitthewebsitebelow:,

https://hokkaidosummerinstitute.oia.hokudai.ac.jp/en/courses/CourseDetail=G049 研究室のホームページ Websites of Laboratory

https://brcge.mcut.edu.tw/?Lang=en

https://www.eng.hokudai.ac.jp/labo/inorgsyn/

科目名 Course Title	応用化学特別講義[Advanced-Applied Chemistry]					
講義題目 Subtitle	Leading and Advanced Materials Chemistry and Engineering IIID - 2024[Leading and					
	Advanced Materials Chemistry and Engineering IIID - 2024] 島田 敏宏 [SHIMADA Toshihiro] (大学院工学研究院)					
責任教員 Instructor	島田 敏宏 [SHIM	IADA Toshihiro」(大字院上字研究院)				
担当教員 Other Instructors						
科目種別 Course Type						
開講年度 Year	2024	時間割番号 Course Number	094445			
期間 Semester	Intensive	単位数 Number of Credits	1			
授業形態 Type of Class	Lecture	対象年次 Year of Eligible Student	~			
対象学科・クラス Eligible Depar						
ナンバリングコード Numbering 補足事項 Other Information	Gode	CHEM_ELCOM 6411				
		(美宿校業利日//美宿のな))				
授業実施方式 Class Method		4 遠隔授業科目《遠隔のみ》				
キーワード Key Words						
materials informatics, python						
授業の目標 Course Objectives						
•		rcises. In the lecture, basic knowledge o				
		we start from basic python programming	; and instruct now to use various			
libraries including tensorflow, s	cikit learn, stail, Gr	y etc. and databases.				
到達目標 Course Goals						
	data science and m	achine learning, especially about terminolo	077			
2. Learning how to use libraries			5.7.			
3. Practical usage of packages						
授業計画 Course Schedule						
1. Neural networks						
2. Rdkit library for chemicals						
3. Machine learning for molecul	les					
4. Sckit learn – library for mach	hine learning					
5. Reinforced learning toward p	protein-folding analy	vsis				
6. Genetic algorithm						
7. Bayesian concept						
8. Interpretation of machine lea						
準備学習 (予習・復習)等の内容						
		eyboard and internet connection				
Homework: After each day, ho		ned.				
成績評価の基準と方法 Gradin						
After each day, homework will	be assigned. The a	nswer and final report will be used for grad	ling.			
	.u D!					
他学部履修の条件 Other Face	uity Requirements					
テキスト・教科書 Textbooks						
None						
講義指定図書 Reading List						
Any textbooks or websites on p	ovthon language					
参照ホームページ Websites						
This course will be provided as part of the Hokkaido Summer Institute., For more information (invited lecturers, course						
details, etc.), please visit the website below:,						
https://hokkaidosummerinstitute.oia.hokudai.ac.jp/en/courses/CourseDetail=G053						
研究室のホームページ Websites of Laboratory						
https://www.eng.hokudai.ac.jp	/labo/kotai/en/inde	ex.html				
備考 Additional Information	, .					
Required Equipment for a class						
		f installation will be given to registered	students prior to the course.The			
participants may be contacted in advance for preparation of python language.						

科目名 Course Title	応用化学特別講	応用化学特別講義[Advanced-Applied Chemistry]				
講義題目 Subtitle	Leading and Adv	Leading and Advanced Biological and Polymer Chemistry and Engineering II - 2024[Leading				
	and Advanced Bio	ological and Polymer Chemistry and Engin	eering II - 2024]			
責任教員 Instructor	磯野 拓也[ISO]	NO Takuya] (大学院工学研究院)				
担当教員 Other Instructors	Hsin-Lung CHEN	Hsin-Lung CHEN (National Tsing Hua University),				
	SATOH Toshifumi (工学研究院), LI FENG (工学研究院)					
科目種別 Course Type						
開講年度 Year	2024	時間割番号 Course Number	094446			
期間 Semester	Intensive	単位数 Number of Credits	1			
授業形態 Type of Class	Lecture	対象年次 Year of Eligible Student	~			
対象学科・クラス Eligible Depa	rtment/Class					
ナンバリングコード Numbering	Code	CHEM_ELCOM 6411				
補足事項 Other Information						
授業実施方式 Class Method 1 対面授業科目《対面のみ》						
キーワード Key Words						
Polymer, Structure, Phase tran	sition, Properties,	Small angle scattering				

授業の目標 Course Objectives

The connectivity and collective behavior of monomers give rise to intriguing properties that set polymers apart from small molecules. Polymer physics is a specialized field within polymer science that concentrates on the study of the structure, dynamics, and physical properties of polymers. The structures observed in polymers within the experimental time scale are typically metastable and exhibit distinctive features across a wide spectrum of length scales. Therefore, comprehending the fundamental thermodynamic and kinetic principles governing the structure formation is crucial for controlling and designing the hierarchical structures and properties of polymers aiming for practical applications as well as developing novel functional materials.

This course is designed to impart the fundamental concepts of polymer physics to students. We will commence with an exploration of single-chain behavior and gradually delve into the topics including polymer solution thermodynamics, glass transition, self-assembly behavior, viscoelasticity, and dynamics. Additionally, we will briefly touch upon the application of small-angle scattering techniques in the analysis of polymer nanostructures. The goal of this course is to provide students from diverse backgrounds with a foundational understanding of polymer physics that can serve as a stepping stone to grasp the intricacies of the processing-structure-property relationship and the mechanisms dictating the morphological formation of polymers.

到達目標 Course Goals

This course aims to assist students with little or no prior background in polymer science in developing a fundamental understanding of polymer physics. It will cover the essential principles that can be exploited to elucidate the structureproperty relationships of polymers. We will also briefly discuss the recent developments in pertinent topics to ignite students' curiosity and motivate them to participate in the research within or related to the domain of polymer physics.

授業計画 Course Schedule

- 1. Brief review of thermodynamics and basic concepts of polymers
- 2. Conformational statistics of single polymer chain
- 3. Thermodynamics of polymer solution and blend
- 4. Glass transition of polymer
- 5. Self-assembly of crystalline polymer and block copolymer
- 6. Viscoelasticity and dynamics of polymers
- 7. Application of small angle scattering in polymer science
- 8. Seminar: Physics of the self-assembly of block copolymers

準備学習 (予習・復習)等の内容と分量 Homework

Final report on the subjects relating to the structure and physical properties of polymers involving the application of the concepts learned from the lectures.

成績評価の基準と方法 Grading System

Your grade will be determined by how well you demonstrate your achievement of the course goals through

- 1. Participation to the discussion (10%)
- 2. Final report (90%)

他学部履修の条件 Other Faculty Requirements

テキスト・教科書 Textbooks

Lecture notes in PDF files will be provided.

講義指定図書 Reading List

Polymer Physics/M	ichael Rubinste	ein, Ralph H. Colby	:Oxford Univ	Pr, 2003		
Introduction to Phys	ical Polymer So	cience∕Leslie H. S	perling:Wiley-	Interscience, 2005	i	
Polymer Physics/U	.W. Gedde:Spi	ringer, 1995				
参照ホームページ V	lebsites					
This course will be	provided as p	oart of the Hokkaid	lo Summer In	stitute., For more	information (invited	lecturers, course
details,	etc.),	please	visit	the	website	below:,
https://hokkaidosun	nmerinstitute.o	ia.hokudai.ac.jp/en/	/courses/Cour	rseDetail=G043		
研究室のホームペー	-ジ Websites o	f Laboratory				
https://sites.google.	.com/gapp.nth	1.edu.tw/polymer-p	hysics-laborat	ory/home		
http://poly-ac.eng.h	nokudai.ac.jp/i	ndex_e.html				
備考 Additional Info	rmation					
Other Instructor: Hsin-Lung Chen (National Tsing-Hua University)						
The class will be hele	d on campus ar	nd/or in real-time w	eb system.			
We will announce the	e details via EL	MS. Please carefull	y see ELMS.			

科目名 Course Title	业学安莱 中学 [Iw	dustrial Dresting in Chamical Dressages	1				
	化学産業実学[Industrial Practice in Chemical Processes]						
講義題目 Subtitle							
責任教員 Instructor	長谷川 淳也 [HASEGAWA Junya] (触媒科学研究所)						
担当教員 Other Instructors							
科目種別 Course Type							
開講年度 Year	2024	時間割番号 Course Number	094451				
期間 Semester	Intensive	単位数 Number of Credits	1				
授業形態 Type of Class	Lecture	対象年次 Year of Eligible Student	~				
対象学科 クラス Eligible Depa							
ナンバリングコード Numbering	Code	CHEM_ELCOM 5200					
補足事項 Other Information		1 払デ「「「「「「」」」 「「」」 「」 「」 「」 「」 「」 「」 「」 「」 「					
授業実施方式 Class Method		1 対面授業科目《対面のみ》					
キーワード Key Words							
		and Development, Chemical Technology	, Industry–academia Collaboration				
授業の目標 Course Objective							
		nd are working at the forefront in indus					
		cate how the things you learn at univer	sities will help you in the future and				
what is requested by companie		whing in industry and consider your fits	we and the way in which you palate to				
-	asp the image of we	orking in industry and consider your futu	re and the way in which you relate to				
the society. 到達目標 Course Goals							
	ly the real necessi	ty of chemical technology for the socie	ty how researches should behave in				
		ing safety, environmental protection,					
information, and to cultivate a		ing salety, environmental protection,					
授業計画 Course Schedule							
	rs as well as manag	gers working at the forefront at a compa	ny and an national research institute.				
		re. For the schedule, see "Additional In					
The concrete plan of lectures i		,					
-	,						
1. Forefront of research and de	1. Forefront of research and development of companies						
		; its background as well its social signific	ance				
2. Outlook and Task of chemic							
Explanation on future outlook	and agenda-setting	•research strategy by point of global vie	w including concrete examples.				
3. Chemist image pursued in th	ne society						
Explanation on the necessary	capability for chemi	cal researchers who are to be involved	in research in future and items which				
need to be studied during scho	ol days based on th	neir experiences of the lecturers					
4. Explanation on the weight of responsibility and its efforts of the chemical technology towards environmental protection.							
Consideration on the chemical	technology which o	contribute to establishing sustainable so	ciety including food issues and energy				
problem.							
準備学習 (予習・復習)等の内		k					
Review the lesson contents by							
成績評価の基準と方法 Gradir							
As a general rule, the percentage of your attendance rate should be more than 75%.							
An absent report should be submitted in advance.							
The test conducted in the each lecture is evaluated.							
他学部履修の条件 Other Faculty Requirements							
テキスト・教科書 Textbooks							
プイスト・叙科書 Textbooks 教科書はとくに指定せず、Lecture 時はパワーポイントを使用する。							
取得音なことに指定せず、Lecture 時はパック かパンドを使用する。 Textbooks are not used. Slides prepared with PowerPoint are used.							
Textbooks are not used. Slides prepared with PowerPoint are used. 講義指定図書 Reading List							
参照ホームページ Websites							
研究室のホームページ Websites of Laboratory							

備考 Additional Information This lecture will be offered as an intensive lecture in October 7-9. The schedule and place of the lecture will be noticed later.

科目名 Course Title	マイクロ・ナノ化	之学[Micro-Nanochemistry]			
講義題目 Subtitle					
責任教員 Instructor	村誠 敬「MUF	AKOSHI Kei] (大学院理学研究院)			
担当教員 Other Instructors	UENO Kosei (理学研究院), TOKESHI Manabu (工学研究院), TANI Hirofumi (工学研究院),				
		NAKASAKA Yuta (工学研究院)			
科目種別 Course Type					
開講年度 Year	2024	時間割番号 Course Number	094452		
期間 Semester	Fall	単位数 Number of Credits	1		
授業形態 Type of Class	Lecture	対象年次 Year of Eligible Student	~		
対象学科・クラス Eligible Depa	rtment/Class				
ナンバリングコード Numbering	Code	CHEM_ELCOM 5222			
補足事項 Other Information					
授業実施方式 Class Method		1 対面授業科目《対面のみ》			
キーワード Key Words					
Microchemistry, Nanochemistr	y, Microchip, Bio	ochip, Microreactor, Single Atom/Molecule	e Manipulation		
授業の目標 Course Objective					
		micrometer – nanometer dimensions inclu	iding microfabrication technologies in		
chemistry, microchips/biochips	s, and microreact	ors.			
到達目標 Course Goals					
The students will be able to lea					
-Fundamental aspects in micro		-			
-Chemical applications of micr					
-Single molecular and atom ma	nipulation techni	ques			
授業計画 Course Schedule					
K. Ueno (2 lectures)					
 Micro/nanofabrication techn 	iques / Micro/na	nostructures / Light-field enhancement /	Radiation force		
K. Murakoshi (2 lectures)					
-Single atom / Molecule manip	oulation / Nanoch	emistry			
M. Tokeshi (2 lectures)					
-Historical background of micr	ro-nanochemistry	/ State of the art technologies and recent	t topics in Microchips/Biochips		
H. Tani (1 lecture)					
-Biochip					
Y. Nakasaka (1 lecture)					
-Microreactors					
準備学習 (予習・復習)等の内	容と分量 Homew	ork			
Basic analytical and physical c	hemistry in under	graduate level			
成績評価の基準と方法 Gradir	ng System				
Learning attitude and report					
他学部履修の条件 Other Fac	ulty Requirement	ts			
テキスト・教科書 Textbooks					
なし。適宜, 資料を配布する					
講義指定図書 Reading List					
参照ホームページ Websites					
研究室のホームページ Websit	tes of Laboratory	,			
備考 Additional Information					

科目名 Course Title	生命分子化学	学特論[Modern Trends in Biomolecular Chem	nistry]	
講義題目 Subtitle				
責任教員 Instructor		SAKAGUCHI Kazuyasu](大学院理学研究院		
担当教員 Other Instructors	MATSUMOTO Kenichiro (工学研究院), UCHIDA Takeshi (理学研究院),			
	TAJIMA Kenji (工学研究院), OGASAWARA Yasushi (工学研究院)			
科目種別 Course Type				
開講年度 Year	2024	時間割番号 Course Number	094453	
期間 Semester	Summer	単位数 Number of Credits	1	
授業形態 Type of Class	Lecture	対象年次 Year of Eligible Student	~	
対象学科・クラス Eligible Depa	rtment/Class			
ナンバリングコード Numbering	Code	CHEM_ELCOM 5230		
補足事項 Other Information				
授業実施方式 Class Method		1 対面授業科目《対面のみ》		
キーワード Key Words				
-	tructure mole	cular mechanism, biosynthetic mechanism,	animal cells secondary metabolites	
biopolymers, bioremediation	nucluic, mole	and meenumon, prosynthetic meenulish,	ammu cene, secondary metabolites,	
授業の目標 Course Objective	2			
		ineering subjects on of bio-molecules will be	studied focusing on the fields of life	
science, information, medicine,			, statied locusing on the helds of life	
到達目標 Course Goals		-11 .		
	lorstand doop!	y the topics of genetic information, protei	n structure animal coll cultivation	
environment.	iymers, and c	elean environments in the fields of life s	science, information, medicine, and	
·····································				
	CCE :11 :	lectures on the following topics, from basic		
	tional regulatic ibrational Spec using a bacteriu cyme conversio Dogma from the	on in proteins troscopy um and its application n and fermentation production of compounds e Biosynthesis of Natural Products		
成績評価の基準と方法 Gradin You will be evaluated by active	report on the g System participation i egular classes	subject which instructor give every time. ncluding quiz (30%), and assignment on each is the minimum condition to evaluate.	n topic (70%).	
テキスト・教科書 Textbooks 適宜資料を配布する。 講義指定図書 Reading List				
参照ホームページ Websites				
研究室のホームページ Websit	es of Laborato	ry		
備考 Additional Information				

科目名 Course Title	総合化学特調	総合化学特論 I (Modern Trends in Physical and Material Chemistry)[Modern Trends in				
	Physical and N	Physical and Material Chemistry]				
講義題目 Subtitle						
責任教員 Instructor	島田 敏宏[S	HIMADA Toshihiro] (大学院工学研究院)				
担当教員 Other Instructors	HARADA Jun	(理学研究院),KOBAYASHI Atsushi (理学	^全 研究院), FUSHIMI Koji (工学研究			
	院), MASUBUCHI Yuji (工学研究院), KITAGAWA Yuichi (工学研究院), TOYAO Takashi (触媒科学研究所), WAIZUMI Hiroki (工学研究院), KITANO Sho (工学研究院)					
科目種別 Course Type						
開講年度 Year	2024	時間割番号 Course Number	094454			
期間 Semester	Intensive	単位数 Number of Credits	1			
授業形態 Type of Class	Lecture	対象年次 Year of Eligible Student	~			
対象学科・クラス Eligible Depa	artment/Class					
ナンバリングコード Numbering Code		CHEM_ELCOM 5241				
補足事項 Other Information						
授業実施方式 Class Method		1 対面授業科目《対面のみ》				

molecular materials, ferroelectrics, metal complexes, corrosion, electrochemistry, inorganic materials, ceramics, optofunctional materials, heterogeneous catalysts, 2D semiconductors, chemical sensors, nanomaterials

授業の目標 Course Objectives

This course is intended to provide cutting-edge research topics on physical and materials chemistry. The topics include molecular ferroelectrics, metal complexes with various functions, observation of detailed surface processes in electrochemistry, inorganic materials, opto-functional materials, heterogeneous catalysts, 2D semiconductors as sensors, chemo-functional nano-materials.

到達目標 Course Goals

Through a series of lectures, students understand various fields of chemistry and are expected to expand their horizons.

授業計画 Course Schedule

Detailed schedule will be informed one month before the start of this course.

List of lecture titles in this course:

•Molecular ferroelectrics

- •Coordination chemistry for solar-fuel production
- Detailed analysis of electrode reactions on practical material surfaces using modern electrochemical methods

•New functional ceramics and inorganic materials – structure and properties

- •Photofunctional lanthanide complexes designed through quantum chemistry
- •Heterogeneous catalysis research using machine learning
- ·Chemical sensors with atomically thin two-dimensional semiconductors

•Chemical and catalytic functions of composite materials studied with in situ electrochemical spectroscopy

準備学習 (予習・復習)等の内容と分量 Homework

Students will be required to submit reports after the lectures.

成績評価の基準と方法 Grading System

Students are required to attend at least 70% of the lectures. Evaluation as pass/fail will be based on the submitted reports.

他学部履修の条件 Other Faculty Requirements

テキスト・教科書 Textbooks

講義指定図書 Reading List

参照ホームページ Websites

 This course will be provided as part of the Hokkaido Summer Institute., For more information (invited lecturers, course details, etc.), please visit the website below:, https://hokkaidosummerinstitute.oia.hokudai.ac.jp/en/courses/CourseDetail=G058

研究室のホームページ Websites of Laboratory

科目名 Course Title	総合化学特論	総合化学特論 II (Modern Trends in Organic Chemistry and Biological Chemistry)[Modern				
	Trends in Org	Trends in Organic Chemistry and Biological Chemistry]				
講義題目 Subtitle						
責任教員 Instructor	鈴木 孝紀[S	UZUKI Takanori] (大学院理学研究院)				
担当教員 Other Instructors	TANINO Keiji	TANINO Keiji (理学研究院), MITA Tsuyoshi (ICReDD), KAMADA Rui (理学研究院),				
	SHIMIZU Yoh	SHIMIZU Yohei (理学研究院), ISHIYAMA Tatsuo (工学研究院),				
	YAMAMOTO	YAMAMOTO Takuya (工学研究院)				
科目種別 Course Type						
開講年度 Year	2024	時間割番号 Course Number	094455			
期間 Semester	Intensive	単位数 Number of Credits	1			
授業形態 Type of Class	Lecture	対象年次 Year of Eligible Student	~			
対象学科・クラス Eligible Dep	artment/Class					
ナンバリングコード Numbering Code		CHEM_ELCOM 5251				
補足事項 Other Information						
授業実施方式 Class Method 1 対面授業科目《対面のみ》						
キーワード Key Words						

Physical Organic Chemistry, Organic Synthesis, Organic Reaction, Organic Transformations, Biological Chemistry, Polymer Chemistry.

授業の目標 Course Objectives

The progress in the fields of organic chemistry and biochemistry is remarkable. In this course, you will learn the basic concepts necessary for understanding research in the fields of advanced organic chemistry and biochemistry, give an overview of the latest trends, and then learn about cutting-edge research results. You will discuss various topics in organic chemistry and biochemistry research. The goal is to be able to write reports that include suggestions for your own ideas on cutting-edge organic and biochemical research.

到達目標 Course Goals

- 1. You can explain the basic concepts needed to understand advanced organic chemistry and biochemical research.
- 2. You can explain an overview of cutting-edge organic chemistry and biochemical topics.
- 3. You can discuss among students with different backgrounds.
- 4. You can make research proposals that incorporate your own ideas.

授業計画 Course Schedule

- 1. Guidance (Suzuki)
- 2. Advanced organic synthetic chemistry (Tanino): Carbocycles, Ring Strain, Ene-diyne
- 3. Advanced computational reaction chemistry (Mita): Radial reaction, Carbon dioxide, Computational chemistry
- 4. Advanced bioorganic chemistry: Sugar chemistry, Glycosyltransferase
- 5. Advanced life chemistry (Kamada): Biochemistry, Innate immunity
- 6. Advanced organic reaction chemistry (Shimizu): Catalysis, Chemoselectivity
- 7. Advanced organic transformation chemistry (Ishiyama): Transition metal-catalyst, borylation, diboron
- 8. Advanced polymer chemistry (Yamamoto): Polycyclic polymer, Supramolecular chemistry, Self-organization

準備学習 (予習・復習)等の内容と分量 Homework

In this course, you will be given an assignment each time. You choose two assignments and submit an answer (report) by the specified date.

成績評価の基準と方法 Grading System

You will be evaluated by learning attitude (20%) and two submitted reports (80% in total). You will be given assignment by instructors each time, among which you choose two assignments to submit. Attendance of 70% or more classes is the minimum condition to evaluate a student.

テキスト・教科書 Textbooks

講義指定図書 Reading List

参照ホームページ Websites

This course will be provided as part of the Hokkaido Summer Institute., For more information (invited lecturers, coursedetails,etc.),pleasevisitthewebsitebelow:,https://hokkaidosummerinstitute.oia.hokudai.ac.jp/en/courses/CourseDetail=G045

研究室のホームページ Websites of Laboratory

최다 수 구	tt 74 al and 1				
科目名 Course Title	▲ 虚物理化	学特論[Introductory Physical Chemistry]			
講義題目 Subtitle					
責任教員 Instructor	丸田 悟朗 [MARUTA Goro] (大学院理学研究院)				
担当教員 Other Instructors	ISHIMORI Koichiro (理学研究院), MURAKOSHI Kei (理学研究院)				
科目種別 Course Type					
開講年度 Year	2024	時間割番号 Course Number	094456		
期間 Semester	Spring	単位数 Number of Credits	1		
授業形態 Type of Class	Lecture	対象年次 Year of Eligible Student	~		
対象学科・クラス Eligible Depa					
ナンバリングコード Numbering	Code	CHEM_ELCOM 5002			
補足事項 Other Information					
授業実施方式 Class Method		1 対面授業科目《対面のみ》			
キーワード Key Words					
Molecular orbital theory, Spect	roscopy, Surfa	ce, Equilibrium and Kinetics			
授業の目標 Course Objective					
The purpose of this course is	to understan	d the fundamental concepts of molecular o	orbital theory, spectroscopy, surface,		
equilibrium as well as kinetics i	n physical che	mistry.			
到達目標 Course Goals					
Goals are to develop skills to	solve problems	in physical chemistry and acquire the capa	city how the knowledge is applied for		
chemical application.					
授業計画 Course Schedule					
1. Processes on solid surfaces	(Atkins' Phys	ical Chemistry 10th edition, Chapter 22)			
Structure of solid surfaces, the	extent of adso	orption, heterogeneous catalysis, processes a	at electrode		
2. Rotational and vibrational sp	oectra (Atkins'	Physical Chemistry 10th edition, Chapter 1	2)		
General features of spectrosco	ру				
3. Electronic transitions and m	agnetic resona	nce (Atkins' Physical Chemistry 10th editio	n, Chapter 13, 14)		
		the fates of electronically excited states, the	e effect of magnetic fields on electrons		
and nuclei, nuclear magnetic re					
		Chemistry 10th edition, Chapter 10)			
	-	chemical bonding, Hückel approximation			
準備学習(予習・復習)等の内	容と分量 Home	ework			
To be announced.					
成績評価の基準と方法 Gradir		<i>.</i>			
The attitude at the lecture (30					
他学部履修の条件 Other Fac	ulty Requireme	ents			
テキスト・教科書 Textbooks					
講義指定図書 Reading List					
	n∕P. W. Atki	ns, Julio De Paula:Oxford University Press,	2014		
参照ホームページ Websites					
研究室のホームページ Websit	es of Laborato	bry			
備考 Additional Information					

科目名 Course Title	無機化学特論[I	Frontiers of Inorganic Chemistry]	
溝義題目 Subtitle		~~ · · · ·	
責任教員 Instructor	小林 厚志「KC	DBAYASHI Atsushi] (大学院理学研究院)	
但当教員 Other Instructors			
科目種別 Course Type			
開講年度 Year	2024	時間割番号 Course Number	094457
期間 Semester	Spring	単位数 Number of Credits	1
受業形態 Type of Class	Lecture	対象年次 Year of Eligible Student	~
対象学科・クラス Eligible Depar	tment/Class		
ナンバリングコード Numbering (Code	CHEM_ELCOM 5012	
哺足事項 Other Information			
受業実施方式 Class Method			
キーワード Key Words			
-	tate chemistry, n	naterial chemistry, nano materials, nano s	science, photocatalysts, bioinorgani
chemistry	5,		
受業の目標 Course Objectives	;		
		d the properties, structures, and functiona	lities of the coordination compound
which play important roles in th	ne various fields :	such as materials, bioinorganic chemistry,	and nano science, To get the lates
nformation of cutting-edge rese	earch concerning	inorganic and coordination chemistry.	
到達目標 Course Goals			
The goal of this course is total	understanding of	the importance of coordination compounds	s from the viewpoints of coordination
		elop the ability to predict structures,	
-		tudents learn the sense of study in the	fields of inorganic and coordination
chemistry (typical concepts are	listed below).		
l) Ligand-field theory			
2) Marcus Theory	,		
3) Nano-science of coordinatior	-		
4) Importance of metal complex	es in applied che	mistry and biochemistry	
授業計画 Course Schedule (1) Basics and application of liga	and field the own		
(2) Ligand exchange and electro		tal complexes	
(3) Photo-induced electron tran		-	
(4) Important effect of impuritie			
(5) Interesting properties of nam			
(6) Group discussion about rece			
準備学習(予習・復習)等の内容			
(1) You must answer to mini-ex			
		blished research paper by the final class o	f this course. Your submitted repor
will be used in the group discus			
成績評価の基準と方法 Grading	g System		
You will be evaluated by mini-	-exam in each c	lass (40%), and report and presentation ((60%). More than 70% attendance is
mimimum condition to evaluate	a student.		
他学部履修の条件 Other Facu	Ity Requirements	S	
テキスト・教科書 Textbooks			
講義指定図書 Reading List			
Shriver & Atkins' Inorganic Che	emistry∕Peter A	tkins:Oxford University Press, 2010	
参照ホームページ Websites			
	es of Laboratory		
参照ホームページ Websites	es of Laboratory		
参照ホームページ Websites 研究室のホームページ Website 備考 Additional Information		ng format that combines on-demand video	

科目名 Course Title	有機化学特論[Special Lecture on Organic Chemistry]				
講義題目 Subtitle					
責任教員 Instructor	谷野 圭持[TAN	INO Keiji] (大学院理学研究院)			
担当教員 Other Instructors	ITOH Hajime (工				
科目種別 Course Type					
開講年度 Year	2024	時間割番号 Course Number	094458		
期間 Semester	Summer	単位数 Number of Credits	1		
授業形態 Type of Class	Lecture	対象年次 Year of Eligible Student	~		
対象学科・クラス Eligible Depa					
ナンバリングコード Numbering	Code	CHEM_ELCOM 5262			
補足事項 Other Information					
授業実施方式 Class Method		1 対面授業科目《対面のみ》			
キーワード Key Words					
basic organic chemistry, physic	al organic chemistr	y, organometallic chemistry, synthetic orgar	nic chemistry, polymer chemistry		
授業の目標 Course Objective	5				
		latest trends and progresses in organic che			
	who have not stud	ied the specialized course of organic chemist	try.		
到達目標 Course Goals					
		able to understand the recent trends and fu	ture problems in physical organic		
- · · ·	histry, synthetic org	anic chemistry, and polymer chemistry.			
授業計画 Course Schedule					
Lecture 1. Electroorganic synt					
Lecture 2. Introduction to asyr					
		ganic compounds: the boration approach			
Lecture 4. How to understand		and future in controlling the alignment of mo	alogulos		
Lecture 6. Lessons from enzym			Jiecules		
		ducts: Comparison between chemical synthe	sis and enzymatic synthesis		
準備学習(予習・復習)等の内容					
			on and review for each topic are		
given by the lecturer.	Students are expected to comprehend the lecture for preparing reports. Details for preparation and review for each topic are given by the lecturer.				
成績評価の基準と方法 Gradin	g System				
It is required to attend at least	t 70% of the lecture	s. Evaluation as pass/fail will be based on t	he level of attendance (20%) and		
submitted reports (twice, 40% each).					
他学部履修の条件 Other Faculty Requirements					
テキスト・教科書 Textbooks					
Textbooks are not assigned.					
講義指定図書 Reading List					
参照ホームページ Websites					
研究室のホームページ Websites of Laboratory					
WI元王V/小一ムハーン Websiles of Laboratory					
備考 Additional Information					

科目名 Course Title	主 淋 牛 畑 ル 学	持論[Introduction to Basic Biological Chemi	istmi
構設的 Course Title 講義題目 Subtitle	· 巫啶生物化子	Tramelineroduction to basic biological Chem	1211 À Ì
請報題日 Subutle 責任教員 Instructor	本十	OTEGI Fumio] (遺伝子病制御研究所)	
担当教員 Other Instructors		inori (遺伝子病制御研究所), ABE Kazuhiro	。(理学研究院)
科目種別 Course Type		IIION(夏伍丁州前御如九別),ADE Kazunin	5(埕子坝九院)
開講年度 Year	2024	時間割番号 Course Number	004450
開講平度 Year 期間 Semester		时间剖金亏 Course Number 単位数 Number of Credits	094459
	Intensive	单位致 Number of Credits 对象年次 Year of Eligible Student	\sim
授業形態 Type of Class 対象学科・クラス Eligible Depa	Lecture	为家中次 fear of Eligible Student	
対象子科・クラス Eligible Depa ナンバリングコード Numbering		CHEM EL COM 5091	
オレート Numbering 補足事項 Other Information	Code	CHEM_ELCOM 5021	
		1 ムデ技業的ロルムデッス、	
授業実施方式 Class Method		1 対面授業科目《対面のみ》	
キーワード Key Words			11 1
		oncogene, immunity, infectious disease, ce	ellular asymmetry
授業の目標 Course Objective		1 1 1 · . · . · · ·	
	-	molecular mechanisms that underlie basic	
growth, cell differentiation, in including cancer and	ninunity and cell	ular asymmetry. How disorder of the reg	ulatory mechanism causes diseases
-	used In addition	, various tochnologies for imaging dynami	a malagulan babayigun in liying galla
will be also discussed.	ussed. In addition	n, various technologies for imaging dynamic	c molecular behaviour in hving cens
到達目標 Course Goals			
	tand the basis r	egulatory mechanisms of gene expression,	call growth and immuno system and
developing mechanisms for the			cen growth and minune system and
授業計画 Course Schedule	Telateu uiseases.		
Day 1, 2: Fumio Motegi			
Interior design of cellular asym	motry		
Day 3, 4: Akinori Takaoka	line er y		
Molecular signalings in host de	fense system		
準備学習(予習・復習)等の内		ork	
Review the contents of each le			
成績評価の基準と方法 Gradir			
Report of the task (100%)			
他学部履修の条件 Other Fac	ulty Requirement	S	
テキスト・教科書 Textbooks			
講義指定図書 Reading List			
参照ホームページ Websites			
https://hokkaidosummerinstitu	ıte.oia.hokudai.ad	c.jp/en/courses/CourseDetail=G044,	https://www.motegilab.com,
https://www.igm.hokudai.ac.jp			
	tes of Laboratory	,	
https://www.igm.nokduar.ac.jp 研究室のホームページ Websit https://www.motegilab.com	tes of Laboratory		
研究室のホームページ Websit	-		

科目名 Course Title	分子物理化学特論[Molecular Physical Chemistry]			
講義題目 Subtitle	为于物理化学特确[Molecular Physical Chemistry]			
責任教員 Instructor	佐藤 信一郎 [SATOH Shinichiro] (大学院工学研究院)			
担当教員 Other Instructors				
科目種別 Course Type				
開講年度 Year	2024	時間割番号 Course Number	094460	
期間 Semester	Spring	単位数 Number of Credits	1	
授業形態 Type of Class	Lecture	対象年次 Year of Eligible Student	~	
対象学科・クラス Eligible Depa				
ナンバリングコード Numbering	Code	CHEM_ELCOM 5100		
補足事項 Other Information				
授業実施方式 Class Method		1 対面授業科目《対面のみ》		
キーワード Key Words	`			
Quantum Mechanics, Perturbat	ion Theory, Stark E	ffect, Zeeman Effect, Photoabsorption and	Emission	
授業の目標 Course Objectives	\$			
Quantum theory is essential to	understand molecul	ar physical chemistry.		
The lecture is intended for	graduate students	who have a general background in elem	nentary quantum dynamics, and	
concentrates on the perturbation	tion theory to give	e students a deep and essential understa	nd on the interactions between	
molecular system and external	fields such as electr	ic, magnetic, and photon fields.		
到達目標 Course Goals				
By the end of the semester you				
		echanics to solve simple model problems.		
		echanical nature of matter to gain insight in	to the structure and dynamics of	
atoms, molecules, and nanomat	erials.			
授業計画 Course Schedule				
	theory: first-orde	er perturbation theory including degene	rate system and second-order	
perturbation theory				
2. Stark effects of hydrogen atom: the first-order interactions for 2s, 2px, 2py, 2pz degenerate states and the second-order				
	Slarizability of hydr	ogen atoms will be discussed on the basis of	of the second-order perturbation	
theory.	n theory			
3. Time-dependent perturbatio	-	discussed on the basis of time-dependent n	orturbation theory	
4. Fliotoabsolption and emissic	n processes win be	discussed on the basis of time-dependent p	er turbation theory.	
準備学習 (予習・復習)等の内容	····································			
		n the textbook beforehand: page ranges will	l he announced at least in a week	
ahead.	Televant contents I	in the textbook beforenand, page ranges will	i be announced at least in a week	
anead. 成績評価の基準と方法 Grading System				
		ified to take the final exam. Evaluations wi	ill be made based on (1) learning	
The attendance rate must be over 70% to be qualified to take the final exam. Evaluations will be made based on (1) learning attitude (20%), (2) reports (80%).				
他学部履修の条件 Other Faculty Requirements				
テキスト・教科書 Textbooks				
現代量子化学の基礎/中島威 藤村勇一:共立出版, 1999				
講義指定図書 Reading List				
-				
参照ホームページ Websites				
研究室のホームページ Websites of Laboratory				
https://cma.eng.hokudai.ac.jp/index_english.html				
備考 Additional Information				
Attend "Quantum Chemistry" or an equivalent lecture ("Quantum Mechanics") in undergraduate school.				

Attend "Quantum Chemistry" or an equivalent lecture ("Quantum Mechanics") in undergraduate school.

科目名 Course Title	物所推出你们	±=☆[C++	-41-]		
講義題目 Subtitle	物質構造解析学特論[Structural Analysis of Inorganic Materials]				
員任教員 Instructor 担当教員 Other Instructors	責任教員 Instructor 三浦 章 [MIURA Akira] (大学院工学研究院)				
科目種別 Course Type					
	0004		004461		
開講年度 Year 期間 Semester	2024	時間割番号 Course Number	094461		
授業形態 Type of Class	Spring Lecture	単位数 Number of Credits 対象年次 Year of Eligible Student			
対象学科・クラス Eligible Depa					
ナンバリングコード Numbering		CHEM_ELCOM 5110			
補足事項 Other Information	0000				
授業実施方式 Class Method		1 対面授業科目《対面のみ》			
キーワード Key Words					
	etron microscone	neutron diffraction, X-ray absorption	on spectroscopy solid-state NMR		
Computational chemistry	etron meroscope,	neutron unitaction, X ray absorption	shi spectroscopy, sond state runt,		
授業の目標 Course Objectives					
•		understand the relation between cry	stal structure and electron density		
		lied for the analysis of inorganic mate			
		of light elements. The principle of X -			
-	-	ssed. Structural analysis of inorganic m			
introduced.					
到達目標 Course Goals					
Understanding the principles	of average structur	re analysis using diffraction and of a	variety of local structure analysis.		
Understanding why we should u	ise both average and	d local structure analysis.			
授業計画 Course Schedule					
		tering, absorption of x-ray etc.			
	-	r diffractometer, qualitative and qua	ntitative analyses, lattice parameter		
determination, crystallite size a					
3. Neutron diffraction: Differen					
4. X-ray scattering and X-ray					
structure analysis.	ismission, analytica	l and scanning electron microscopies	s for microstructure and electronic		
6. Solid State NMR					
7. Computational chemistry: D	FT and data science				
8. Examination	i i and data science	·			
準備学習(予習・復習)等の内容	容と分量 Homework				
		l analysis methods for the materials und	der investigation by each student.		
成績評価の基準と方法 Gradin	g System				
(1) report(40%) and (2) End of	term examination (6	60%).			
他学部履修の条件 Other Faculty Requirements					
テキスト・教科書 Textbooks					
教科書は用いず、プリントを配布する。					
講義指定図書 Reading List					
これならわかる X 線結晶解析 これならわかる X 線結晶解析 / 安岡則武:化学同人, 2000					
セラミックスのキャラクタリゼーション技術:日本セラミックス協会					
参照ホームページ Websites					
研究室のホームページ Websit	-				
	https://www.eng.hokudai.ac.jp/labo/strchem/				
備考 Additional Information					
	Basic knowledge about physical chemistry, inorganic chemistry, solid state chemistry and inorganic materials chemistry are				
required.					

科目名 Course Title	生物資源化学特論[Bioresources Chemistry]		
講義題目 Subtitle			
責任教員 Instructor	田島 健次[TAJI	MA Kenji] (大学院工学研究院)	
担当教員 Other Instructors			
科目種別 Course Type			
開講年度 Year	2024	時間割番号 Course Number	094462
期間 Semester	Spring	単位数 Number of Credits	1
授業形態 Type of Class	Lecture	対象年次 Year of Eligible Student	~
対象学科・クラス Eligible Depa	rtment/Class		
ナンバリングコード Numbering Code		CHEM_ELCOM 5132	
補足事項 Other Information			
授業実施方式 Class Method		1 対面授業科目《対面のみ》	

Natural polymer materials, Eco-friendly material, Polyhydroxyalkanoates, Nano-fibers, Bacterial cellulose, Collagen

授業の目標 Course Objectives

Biomacromolecules are the basic units of living organisms, and can be divided into proteins, nucleic acids, and polysaccharides. Of these, the polymers that exist in large quantities are called natural polymers, and have been used by mankind since ancient times. In this course, students will understand the structure and physical properties of these natural polymers (in other words, biological resource polymers), and then acquire knowledge about their advanced utilization and functionalization.

到達目標 Course Goals

Understand the synthesis mechanism, structure, and physical properties of bioresource polymers such as proteins, polysaccharides, lignin, and biopolyesters, which are abundant in nature, and be able to read and understand the latest papers on their applications and explain their material applications.

授業計画 Course Schedule

- 1. Guidance and introduction
- 2. Natural polymers as materials
- 3. Cellulose
- 4. Polyhydroxyalkanoate
- 5. Nano-fiber(collagen)
- 6. Nano-fiber(plant cellulose)
- 7. Nano-fiber(bacterial cellulose)
- 8. Creation of a report

準備学習 (予習・復習)等の内容と分量 Homework

Students will be given assignments that correspond to the content of the lecture and will be required to write reports. Students will deepen their understanding by reading the latest academic papers and writing reports on them.

成績評価の基準と方法 Grading System

Evaluation will be based on the submission of a report at the end of the lecture. Grades will be based on whether the student has a basic knowledge of the molecular structure and functionality of biomacromolecular materials and their applications, and whether the report is written in a convincing and logical manner. To pass, students must earn at least 60 points out of 100 points. [Syu]: > ca.90 points, [Yu]: > ca.80 points, [Ryo]: > ca.70 points, [Ka]: > ca.60 points

他学部履修の条件 Other Faculty Requirements

テキスト・教科書 Textbooks

適宜資料を配布する。参考書を適宜示すが,教科書は用いない。

講義指定図書 Reading List

参照ホームページ Websites

研究室のホームページ Websites of Laboratory

備考 Additional Information

It is desirable that the students have had previous courses in polymer chemistry and biochemistry. The maximum number of students is approximately 30.

科目名 Course Title	化学反応創成学入門[Introduction to Chemical Reaction Design and Discovery]			
講義題目 Subtitle				
責任教員 Instructor	陳 旻究 [JIN]	Mingoo] (創成研究機構化学反応創成研究:	拠点)	
担当教員 Other Instructors	Min Gao (ICReDD), HUANG Chung-Yang (ICReDD), SIDOROV Pavel (ICReDD),			
	AKAMA Tomoko (ICReDD), LIST Benjamin (ICReDD)			
科目種別 Course Type				
開講年度 Year	2024	時間割番号 Course Number	094463	
期間 Semester	Summer	単位数 Number of Credits	1	
授業形態 Type of Class	Lecture	対象年次 Year of Eligible Student \sim		
対象学科・クラス Eligible Depa	artment/Class			
ナンバリングコード Numbering Code		CHEM_ELCOM 5271		
補足事項 Other Information				
授業実施方式 Class Method		1 対面授業科目《対面のみ》		

Design of Chemical reaction and molecular assembly with functions, Chemoinformatics, Computational Chemistry

授業の目標 Course Objectives

This course introduces a brand-new research way for investigating molecular chemistry. Especially, the design of new chemical reactions and molecular assembly systems with photo-physical functions will be gently introduced, and the methodology for these research subjects will be described. Also, the basics of computational chemistry and chemoinformatics to solve chemical problems will be introduced. Totally four sessions will be delivered to introduce these contents.

1. Applications of Transition Metal Catalysis:

In these lectures, fundamentals of transition metal catalysis will be introduced to provide an overview on important chemical reactions that utilize metal catalysts. Representative examples of their applications in industry and recent research will then be described.

2. Fundamental Idea of Designing Molecular Crystals and Related Functions:

The lecture introduces basic ideas of designing molecular assembly in a solid state in terms of molecular crystals. Also how the molecular crystals can be related to photo-functional properties.

3. Introduction to Chemoinformatics:

The class introduces the field of chemoinformatics - or, simply put, the application of informatics methods to solve chemical problems. As the amount of information on chemical compounds and reactions grows, there is a need for rationalization of that information. Chemoinformatics provides useful tools for chemical search, rational design of compounds with desired properties, synthesis prediction, etc.

4. Introductory Computational Catalysis:

The lectures related to introductory computational catalysis are aimed to understand the basics of computational chemistry, and how to analyze the computational result and energy profile.

到達目標 Course Goals

The main goal of this course is "Knowing the molecular chemistry research fields with experimental and computational methodologies".

Especially, students will know "the fundamentals of transition metal catalysis and their application and recent research", "the basic ideas to design molecular crystals and photo-functions", "What the cheminformatics is and how to use it" and "fundamental knowledge to use computational chemistry on catalysis".

授業計画 Course Schedule

The entire course contains four sessions as below;

1. Applications of Transition Metal Catalysis

- Fundamentals of Organometallic Chemistry
- Examples of Transition Metal Catalysis
- 2. Fundamental Idea of Designing Molecular Crystals and Related Functions:
- Introduction to Molecular Crystal Engineering
- ${\boldsymbol{\cdot}}$ Introduction to Photo–functions with Molecular Crystals

3. Introduction to Chemoinformatics:

- Introduction to Chemoinformatics
- Machine Learning in Chemistry

4. Introductory Computational Catalysis:

Introduction to Computational Catalysis I

• Introduction to Computational Catalysis II

準備学習 (予習・復習)等の内容と分量 Homework

Basic knowledge of chemistry in the undergraduate level might be required.

成績評価の基準と方法 Grading System

We will give a take-home exam with several open-answer questions for each session, that students have to submit before some deadline.

他学部履修の条件 Other Faculty Requirements

テキスト・教科書 Textbooks

講義指定図書 Reading List

参照ホームページ Websites

研究室のホームページ Websites of Laboratory

https://www.icredd.hokudai.ac.jp/all-members/the-huang-lab

https://www.icredd.hokudai.ac.jp/the-jin-group

https://www.icredd.hokudai.ac.jp/the-sidorov-group

https://www.icredd.hokudai.ac.jp/the-gao-group

원모성이 포네	→₩/1,241→1 左/			
科目名 Course Title	月機化字と計算1	有機化学と計算化学の融合論[Strategy for Integrating Organic Chemistry with Computational		
	Chemistry]	Chemistry]		
講義題目 Subtitle				
責任教員 Instructor	美多 剛[MITA]	美多 剛 [MITA Tsuyoshi] (創成研究機構化学反応創成研究拠点)		
担当教員 Other Instructors				
科目種別 Course Type				
開講年度 Year	2024	時間割番号 Course Number	094464	
期間 Semester	Spring/Summer	単位数 Number of Credits	2	
授業形態 Type of Class	Lecture	Lecture 対象年次 Year of Eligible Student ~		
対象学科・クラス Eligible Department/Class				
ナンバリングコード Numbering Code		CHEM_ELCOM 5282		
補足事項 Other Information				
授業実施方式 Class Method 1 対面授業科目《対面のみ》				

Synthetic organic chemistry, Quantum chemistry calculations, Pericyclic reactions, Radical reactions, Transition metal catalyzed reactions, DFT calculations, Automated reaction path search methods, Theoretical chemistry

授業の目標 Course Objectives

The primary objective of this course is to foster a computational understanding of organic reactions, emphasizing the comprehension of reaction mechanisms, including transition state structures, through quantum chemical calculations. Pericyclic reactions, pivotal for their direct contribution to reactions via transition state energy, will be elucidated, distinguishing thermally allowed and forbidden reactions. Through the study of pericyclic reactions, students will qualitatively discern between aromatic and antiaromatic transition states, and they will conduct calculations for each transition state using Gaussian 16. To accomplish this, a solid grasp of the Woodward-Hoffmann rules will be established initially. In the latter part of the course, alongside pericyclic reactions, students will delve into recently emphasized radical reactions and transition metal-catalyzed reactions through quantum chemical calculations, empowering them to analyze and predict organic reactions effectively.

到達目標 Course Goals

First, students will attain a comprehensive understanding of pericyclic reactions, which are considered the third reaction mechanism following ion reactions and radical reactions. They will thoroughly grasp the Woodward-Hoffmann rules, governed by orbital symmetry conservation (without using Frontier Orbital Theory), and employ the concepts of aromatic and antiaromatic transition states, as per Dewar-Zimmerman's interpretation, to comprehend the characteristics and mechanisms of pericyclic reactions. Furthermore, students will acquire the ability to independently compute these transition state structures. Subsequently, the course will introduce mechanistic analyses of radical propagation steps and catalytic cycles of transition metal-catalyzed reactions (including oxidative addition, transmetalation, insertion of unsaturated bonds, β -hydride elimination, and reductive elimination), laying the groundwork for predicting organic reactions using quantum chemical calculations.

授業計画 Course Schedule

- 1. Basics of quantum chemical calculations, methods for obtaining transition state structures.
- 2. Complete understanding of Woodward-Hoffmann rules. Difference between aromatic and antiaromatic transition states.
- 3. Understanding of cycloaddition reactions.
- 4. Understanding of electrocyclic reactions.
- 5. Understanding of sigmatropic rearrangement, keto-enol tautomerism.
- 6. Understanding pericyclic reactions that violate Woodward-Hoffmann rules, and determining the activation barrier heights
- for both allowed and forbidden pathways.
- 7. Deep understanding through exercise problems.
- 8. Introduction of automated reaction path search methods.
- 9. Basics of radical reactions_1.
- 10. Basics of radical reactions_2.
- 11. Examples of mechanistic analysis of radical reactions using quantum chemical calculations.
- 12. Basics of transition metal-catalyzed reactions.
- 13. Analysis of the catalytic cycle for transition metal-catalyzed reactions using quantum chemical calculations.

準備学習 (予習・復習)等の内容と分量 Homework

No preparation required. Only review.

Bring a laptop computer with virus protection to the designated classes.

成績評価の基準と方法 Grading System

Exercise problem (not examination) (20%) • Reports (80%)

他学部履修の条件 Other Faculty Requirements

テキスト・教科書 Textbooks

資料を用意する。

Materials will be provided.

講義指定図書 Reading List

有機化学のための量子化学計算入門 Gaussian の基本と有効利用のヒント/西長 亨・本田 康 共著:裳華房, 2022 ペリ環状反応 第三の有機反応機構, I./フレミング著, 鈴木 啓介・千田 憲孝 訳:化学同人, 2002 Pericyclic reactions (second edition)/Ian Fleming: Oxford University Press, 2015

https://pubs.acs.org/doi/10.1021/jacs.2c09830

参照ホームページ Websites

https://www.icredd.hokudai.ac.jp/ja, https://www.icredd.hokudai.ac.jp/

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https://www.icredd.hokudai.ac.jp/mita-tsuyoshi

備考 Additional Information

A laptop with virus protection is required.

Gaussian 16 is used under a campus license. Software installation will be done during class hours.